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Molbank is a communication journal of synthetic chemistry and natural product chemistry. It publishes "short notes" of experimental data records for previously unpublished single molecules (one compound per paper) as well as "communications" of preliminary but significant results that can involve more than a single compound. For "short notes", any scattered unassembled experimental data for individual compounds which is conventionally not publishable is particularly welcome. Articles that focus primarily on new structure determinations are acceptable also for previously known compounds.

Molbank has been launched to preserve and exploit molecular diversity of both chemical information and chemical substances.

The scope of *Molbank* is reflected by its three journal sections:

Organic Synthesis

Natural Products

Structure Determination

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Basel, September 2019

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Synthesis and Crystal Structure of A Pyrithione Derivative: Bis{2-[(1-oxidopyridin-2-yl)sulfanyl]-4,5-dihydro-1H-imidazol-3-ium} tetrachlorocuprate(2-) (/1422-8599/2019/2/M1067)

by (9 Łukasz Balewski (https://sciprofiles.com/profile/728720), (9 Franciszek Sączewski (https://sciprofiles.com/profile/52003) and

Maria Gdaniec (https://sciprofiles.com/profile/87951)

Molbank 2019, 2019(2), M1067; https://doi.org/10.3390/M1067 (https://doi.org/10.3390/M1067) - 25 Jun 2019

Abstract The pyrithione derivative, bis{2-[(1-oxidopyridin-2-yl)sulfanyl]-4,5-dihydro-1*H*-imidazol-3-ium} tetrachlorocuprate(2-) (1a) has been obtained by the reaction of one equivalent of 2-[(4,5-dihydro-1*H*-imidazol-2-yl)thio]pyridine 1-oxide hydrochloride with one and a half equivalents of copper (II) chloride dihydrate in methanol in a very good yield. [...] Read more.

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🕐 Mustafa Camas (https://sciprofiles.com/profile/736064) , 🔍 Aboagye Kwarteng Dofuor (https://sciprofiles.com/profile/769694) ,

University of the second se

Molbank 2019, 2019(2), M1066; https://doi.org/10.3390/M1066 (https://doi.org/10.3390/M1066) - 16 Jun 2019

Abstract The Mycobacterium sp. BRS2A-AR2 is an endophyte of the mangrove plant Rhizophora racemosa G. Mey., which grows along the banks of the River Butre, in the Western Region of Ghana. Chemical profiling using ¹H-NMR and HRESI-LC-MS of fermentation extracts produced by the [...] Read more. (This article belongs to the Section Natural Products (/journal/molbank/sections/natural products molbank))

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N-Propargylation of Indolo-Triterpenoids and Their Application in Mannich Reaction (/1422-8599/2019/2/M1065)

by CElmira F. Khusnutdinova (https://sciprofiles.com/profile/684295), CAnastasiya V. Petrova (https://sciprofiles.com/profile/792528),

C Gulnaz M. Bashirova (https://sciprofiles.com/profile/author/WTR4Zk0yMXpySVQ1K0o0VjRQMCt2UDdsNCt1TVA2MlpqS2JUWnhhdzdoRT0=) and

Oxana B. Kazakova (https://sciprofiles.com/profile/2829)

Molbank 2019, 2019(2), M1065; https://doi.org/10.3390/M1065 (https://doi.org/10.3390/M1065) - 13 Jun 2019

Abstract

The introduction of the alkynyl moiety to the triterpenic core through a linkage to the indole nitrogen is described. The reaction of *N*-propargylindoles with *N*-methylpiperazine using Mannich reaction led to propargylaminoalkynyl-triterpenoids, whose structures were established by NMR spectroscopy. <u>Full article (/1422-8599/2019/2/M1065)</u>

(This article belongs to the Section Natural Products (/journal/molbank/sections/natural_products_molbank))

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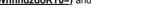
5,5'-Thiobis(3-methoxy-4H-1,2,6-thiadiazin-4-one) (/1422-8599/2019/2/M1064)

by (Andreas S. Kalogirou (https://sciprofiles.com/profile/395192) and (Panayiotis A. Koutentis (https://sciprofiles.com/profile/193)

Melank 2019, 2019 M 1064: https://doi.org/10.3390/M1064 (https://doi.org/10.3390/M1064) - 09 Jun 2019

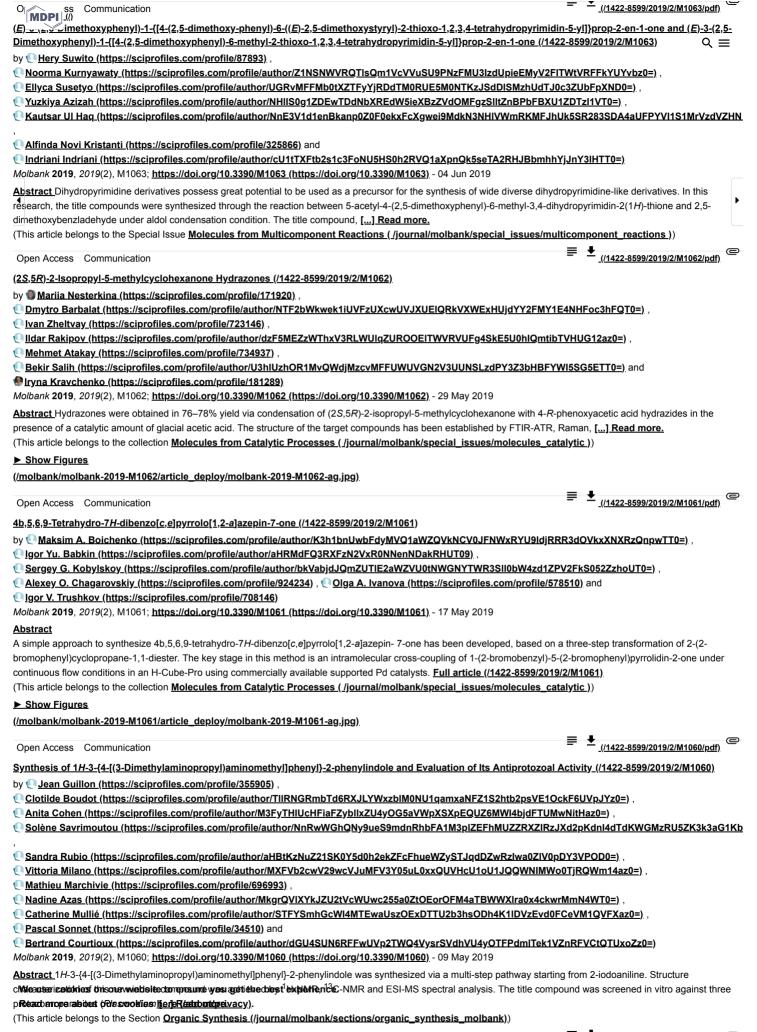
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The reaction of 3-chloro-5-methoxy-4*H*-1,2,6-thiadiazin-4-one (9) with Na₂S·9H₂O (0.5 equiv) in tetrahydrofuran (THF) at ca. 20 °C for 20 h gives 5,5'-thiobis(3-methoxy-4*H*-1,2,6-thiadiazin-4-one) (10) in a 44% yield as yellow needles. The compound was fully characterized. <u>Full article (/1422-8599/2019/2/M1064)</u> Accept (/accept_cookies) (This article belongs to the Special Issue <u>Heteroatom Rich Organic Heterocycles (/journal/molbank/special_issues/Heteroatom_Heterocycles</u>))



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5-[oro-2-oxoindolin-3-ylidene]-3-{(E)-[(4-hydroxyphenyl)imino]methyl}-2-thioxothiazolidin-4-one (/1422-8599/2019/2/M1059) MDPI (1) mounaim Safer (https://sciprofiles.com/profile/66734), bv Ukhadidja Khaldoun (https://sciprofiles.com/profile/author/NUFHdVlhOStwM1VFVnV1ZEVNVXB0QmR6Rzd4MVg1MlprRXI1NkFWbHQ4ND0=) and < ≡ U Salima Saidi-Besbes (https://sciprofiles.com/profile/author/a3hqRWQvMU9yRHBxaHdwL3hzNnlwMXQ3U05wNjVaTVNoeExESXdiN1VZYz0=) Molbank 2019, 2019(2), M1059; https://doi.org/10.3390/M1059 (https://doi.org/10.3390/M1059) - 06 May 2019 Abstract N-aminorhodanine as well as isatin are highly solicited motifs known for their wide potential for biological activity. The objective of this work was to synthesize hybrid molecules as kinase inhibitors from these two motifs. In order to study the reactivity of the [...] Read more. (This article belongs to the Special Issue Molecules from Multicomponent Reactions (/journal/molbank/special issues/multicomponent reactions)) O Open Access Short Note (E)-N'-(4-Fluorobenzylidene)-5-methyl-2-(pyridin-3-yl)thiazole-4-carbohydrazide (/1422-8599/2019/2/M1058) by (Vinuta Kamat (https://sciprofiles.com/profile/686347), Construction of the second Suresh P. Nayak (https://sciprofiles.com/profile/author/TIRLcXBRVU9LcUx1RzJ5MmpQeXZ0Ry9wMHBLR1ZxQnhDdkZHamtPTS93Yz0=) Molbank 2019, 2019(2), M1058; https://doi.org/10.3390/M1058 (https://doi.org/10.3390/M1058) - 02 May 2019 Abstract 5-methyl-2-(pyridin-3-yl)-1,3-thiazole-4-carbohydrazide (1) on treatment with 4-fluorobenzaldehyde in presence of catalytic amount of acetic acid, accessed the target compound (2) with the yield of 79%. The target compound was confirmed by ¹H-NMR, ¹³C-NMR, FT-IR and LCMS. In vitro [...] Read more. 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The title Ni^{II} complex, Ni(L)(LH₂) (1), where LH₂ is S-2-methybenzyl-β-N-(2-hydroxy-3-methoxybenzylmethylene) dithiocarbazate, was isolated from the reaction of Ni(acetate)₂·4H₂O and two molar equivalents of LH₂. The complex was characterized [...] Read more. (This article belongs to the Section Structure Determination (/journal/molbank/sections/structure_determination_molbank)) Show Figures (/molbank/molbank-2019-M1057/article_deploy/molbank-2019-M1057-ag.png) C Open Access Short Note 8,18-Dithia-1,4,11,14-tetraazapentacyclo[11.7.0.0^{3,11}.0^{5,9}.0^{15,19}]icosa-3,5(9),6,13,15(19),16-hexaene-10,20-dione (/1422-8599/2019/2/M1056) by (Vladimir A. Ogurtsov (https://sciprofiles.com/profile/685251) and (Oleg A. Rakitin (https://sciprofiles.com/profile/125410) Molbank 2019, 2019(2), M1056; https://doi.org/10.3390/M1056 (https://doi.org/10.3390/M1056) - 13 Apr 2019 Abstract 4H-33²-Thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. In this communication, 2-(chloromethyl)-4H-33²-thieno[3,2-d]pyrimidin-4-one derivatives are of interest as biologically active compounds. d]pyrimidin-4-one (1) was investigated in the reaction with ammonia, potassium phthalimide, and other basic agents. The dimerization [...] Read more. 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Editor-in-Chief

Department of Chemistry and Biochemistry, University of Lethbridge, 4401 University Drive W., Lethbridge, Alberta T1K3M4, Canada

Interests: organoelement chemistry; sulfur-nitrogen compounds; heterocycles; amidines; heteroamidines; free-radicals; electrochemistry; supramolecular chemistry; structural methods; crystallography; electron paramagnetic resonance spectroscopy

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 Interests: medicinal heterocyclic chemistry; pyridazines; carbazoles; nitrogen hetarenes

* former Editor-in-Chief from 2007 up to 30 September 2019; The Editorial Board of Molecules also governs Molbank. Former Editor-in-Chief: up to 2001 Dr. Luc Patiny; in 2002 Prof. Dr. Bruce A. Hathaway; in 2003 Reto Mueller (organicchemistry.org)

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Dr. R. Alan Aitken

E-Mail () Website (http://ch-www.st-and.ac.uk/staff/raa/group)

School of Chemistry, University of St Andrews North Haugh, St Andrews Fife, KY16 9ST, UK Tel. (01334) 463865; Fax: (01334) 463865

Interests: synthesis; synthetic use of flash vacuum pyrolysis; heterocyclic chemistry; reactive intermediates; organophosphorus; organosulfur; heavier main group chemistry



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Department of Pharmacy, School of Life Sciences, Pharmacy & Chemistry, Kingston University, Penrhyn Road, Kingston upon Thames, KT1 2EE, UK

Interests: free radical organic and polymer chemistry; heterocyclic and medicinal chemistry

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<u>E-Mail ()</u> <u>Website (https://community.dur.ac.uk/i.r.baxendale/index.php)</u>

Department of Chemistry, Durham University, University Science Laboratories, South Road, Durham DH1 3LE, UK **Interests:** organic synthesis and methodology; automation; chemical reactors; flow chemistry; enabling technologies; immobilised reagents and scavengers; heterocyclic chemistry

Dr. Jose Berna

E-Mail () Website (http://webs.um.es/ppberna/miwiki/doku.php)

Department of Organic Chemistry, Faculty of Chemistry, University of Murcia, 30100 Murcia, Spain **Interests:** self-assembly; mechanically interlocked molecules (rotaxanes and catenanes); hydrogen bond; template synthesis; molecular recognition; supramolecular chemistry

Prof. Dr. Fang-Rong Chang

E-Mail () Website (http://nphs.kmu.edu.tw/index.php/en-GB/faculty)

Graduate Institute of Natural Products, College of Pharmacy, Kaohsiung Medical University, No. 100, Shih-Chuan 1st Road, Kaohsiung, 80708, Taiwan

Interests: natural products chemistry; medicinal chemistry; transgenic plant (arabidopsis) reportor assay; epigenetic modulation for hicrobial secondary metabolites, functional food, etimopharmacology

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E-Mail () Website (https://chem.uiowa.edu/people/gregory-k-friestad)

De of Chemistry, University of Iowa, Iowa City, IA 52242 USA Tel 35-1364; Fax: +1 319 335 1270

Interests: asymmetric synthesis methodology; free radical reactions; organometallic reagents; natural product $Q \equiv$ synthesis; asymmetric catalysis



Prof. Dr. Bartolo Gabriele

E-Mail () Website1

(http://www.unical.it/portale/strutture/dipartimenti_240/ctc/didattica/homedid/docenti/ordinari/gabriele/) Website2 (https://www2.scopus.com/authid/detail.uri?authorld=7003277013)

Laboratory of Industrial and Synthetic Organic Chemistry (LISOC), Department of Chemistry and Chemical Technologies, University of Calabria, Via Pietro Bucci, 12/C, 87036 – Arcavacata di Rende (Cosenza), Italy Tel. +390984492815; Fax: +390984492044

Interests: new syntheses of high value added molecules through catalytic assembly of simple units; innovative syntheses of heterocyclic molecules of pharmaceutical, agrochemical, or applicative interest; carbonylation chemistry; use of non-conventional solvents in organic synthesis; synthesis and semi-synthesis of bioactive compounds of pharmaceutical or agrochemical interest; synthesis of new materials for advanced applications; extraction, characterization, and evaluation of the biological activity of bioactive principles from natural matrices **Special Issues and Collections in MDPI journals:**

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Prof. Dr. Panayiotis A. Koutentis

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Department of Chemistry, University of Cyprus, P. O. Box 20537, 1678 Nicosia, Cyprus Tel. 0035722892783

Interests: heterocyclic chemistry; sulfur-nitrogen heterocycles; synthetic methods; azaacenes; zwitterionic acenes; stable organic radicals; biologically active heterocycles; isothiazoles; 1,2,3-dithiazoles; 1,2,6-thiadiazines;1,2,4-benzotriazines

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E-Mail () Website (https://www.tu-braunschweig.de/pharmchem/forschung/kunick)

Ins ledizinische und Pharmazeutische Chemie, Technische Universität Braunschweig, Beethovenstraße 55, 38 mppi (/) nSchweig, Germany

Interests: medicinal chemistry; pharmaceutical chemistry; synthesis of heterocycles; protein kinase inhibitors $Q \equiv$



Dr. Raffaella Mancuso

<u>E-Mail ()</u>

Laboratory of Industrial and Synthetic Organic Chemistry (LISOC), Department of Chemistry and Chemical Technologies, University of Calabria, Via Pietro Bucci 12/C, 87036 Arcavacata di Rende (CS), Italy **Interests:** innovative syntheses of high value molecules through catalytic process; new syntheses of heterocyclic compounds of pharmaceutical interest; carbonylation catalyzed chemistry; application of unconventional solvents in advanced organic synthesis; synthesis of novel materials for advanced applications

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Prof. Dr. Ge Meng <u>E-Mail () Website (http://gr.xjtu.edu.cn/web/mengge)</u>

Chemistry Department, Fudan University, Shanghai 200433, China

Fax: +86 029 82655424

Interests: medicinal heterocyclic chemistry; indoles; indazoles; pyridines; thiazoles; thiadiazoles; pyrroles; computer aided drug design; molecular probe



Prof. Dr. Hideto Miyabe

E-Mail () Website (http://www2.huhs.ac.jp/~h070012h/)

School of Pharmacy, Hyogo University of Health Sciences, 1-3-6 Minatojima, Chuo-ku, Kobe 650-8530, Japan **Interests:** organic synthesis and methodology; radical reactions; organocatalysis; asymmetric catalysis; photochemistry; aryne chemistry

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Prof. Dr. Luke R. Odell

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Interests: synthetic methodology; asymmetric catalysis; Lewis acids

Dr. Luc Patiny

E-Mail () Website (http://cheminformatics.epfl.ch)

Director of Chemical Information, Institute of Chemical Sciences and Engineering (ISIC), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland

Interests: chemoinformatics; chemical information; infrastructure for chemical research; online processing and data minina

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Prof. Dr. Vincenzo Piccialli

E-Mail () Website

(https://www.docenti.unina.it/#!/professor/56494e43454e5a4f5049434349414c4c4950434356434e35384532374230 Department of Chemical Sciences, Università degli Studi di Napoli Federico II, Via Cintia 21, 80126 Napoli, Italy Interests: organic and medicinal chemistry; organic synthesis; catalytic oxidative processes; marine natural products;

nucleosides chemistry

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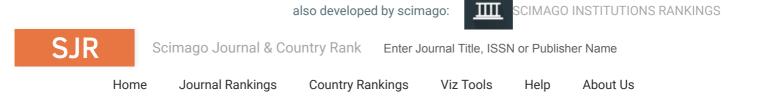
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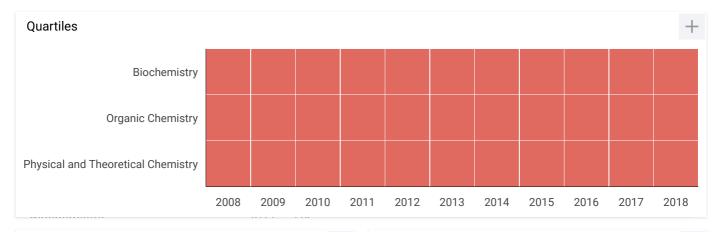
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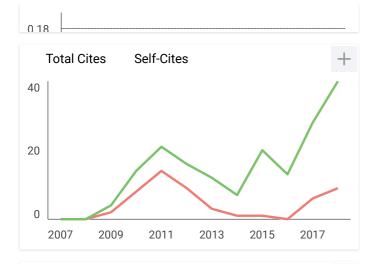


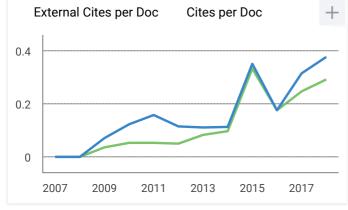
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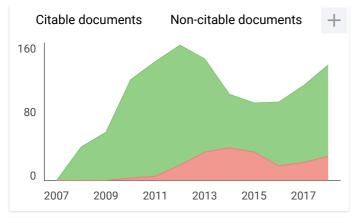
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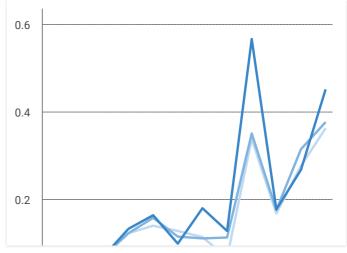
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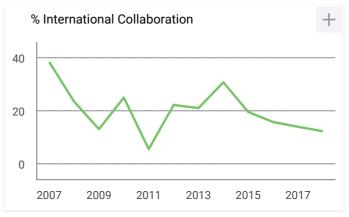


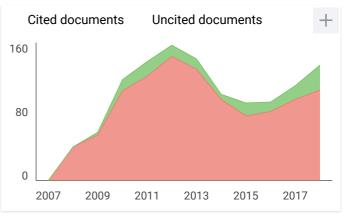












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Communication

(*E*)-3-(2,5-Dimethoxyphenyl)-1-{[4-(2,5-dimethoxyphenyl)-6-((*E*)-2,5-dimethoxystyryl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl]}prop-2-en-1-one and (*E*)-3-(2,5-Dimethoxyphenyl)-1-{[4-(2,5-dimethoxyphenyl)-6-methyl-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl]}prop-2-en-1-one

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Abstract: Dihydropyrimidine derivatives possess great potential to be used as a precursor for the synthesis of wide diverse dihydropyrimidine-like derivatives. In this research, the title compounds were synthesized through the reaction between 5-acetyl-4-(2,5-dimethoxyphenyl)-6-methyl-3,4-dihydropyrimidin-2(1*H*)-thione and 2,5-dimethoxybenzladehyde under aldol condensation condition. The title compound, (*E*)-3-(2,5-dimethoxyphenyl)-1-{[(4-(2,5-dimethoxyphenyl)-6-((*E*)-2,5-dimethoxystyryl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl)]}prop-2-en-1-one (yield 15%), was obtained as major product, whereas (*E*)-3-(2,5-dimethoxyphenyl)-1-{[(4-(2,5-dimethoxyphenyl)-6-((*E*)-2,5-dimethoxyptenyl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl)]}prop-2-en-1-one (yield 15%), as side product through vinylogous aldol condensation.

Keywords: 5-acetyl-4-(2,5-dimethoxyphenyl)-6-methyl-3,4-dihydropyrimidin-2(1*H*)-thione; aldol condensation; vinylogous aldol condensation

1. Introduction

The derivatives of 3,4-dihydropyrimidin-2(1*H*)-ones (DHPMs) as products of the Biginelli reaction are known to have various pharmacological activities, such as anticancer [1], antioxidant [2], anti-inflammation, antibacterial, and antifungal [3], anti HIV [4], and antihypertensive [5]. The discovery of 4-(3-hydroxyphenyl)-3,4-dihydropyrimidine-2(1*H*)-thione, which is well known as monastrol (**1**) as moderate anticancer agent through inhibition of the microtubule-stimulated ATPase activity of Eg5 [6] inspired researchers to use this protein as a protein target to develop new anticancer agents due to its specific function during cell cycle. Replacement of ester group at the 5-position of monastrol with acyl group enhances its potency when compared with the racemic monastrol [7], whereas replacement with cyclohexanone ring (enastron (compound **2**) and dimethylenastron (compound **3**), Figure 1) significantly increased their inhibitor activity as compared to monastrol [8].

Based on the information, we focus our study on finding anticancer compounds from the 5-acyl-3,4dihydropyrimidine-2-thione family.

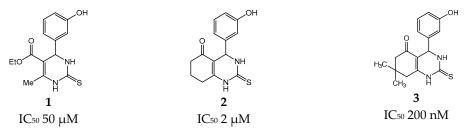


Figure 1. Replacement of ester group of monastrol (1) with acyl group increases anticancer activity.

From the organic synthesis point of view, acetyl group at five-position is a potential moiety for introducing the benzylidene group to form a chalcone analogue using aldol condensation, whereas the methyl group of six-position can be transformed into styryl moiety through vinylogous aldol condensation. Compounds with such structure are known for their antimicrobial and cytotoxicity activity, and have already been prepared by the Biginelli reaction using (thio)urea, derivatives of benzaldehyde, and curcumin as 1,3-dicarbonyl component [6]. However, this reaction route required an additional reaction step of curcumin or its derivatives to be prepared. Herein, we reported the synthesis of the 5,6-dibenzylidene DHPM-type (6) and 5-benzylidene DHPM-type (5) employing aldol condensation while using a 5-acetyl-6-methyl DHPM derivative as precursor.

2. Results and Discussion

The synthesis of the title compounds were performed in two steps, firstly the synthesis of a DHPM derivative (compound **4**) from acetylacetone, 2,5-dimethoxybenzaldehyde, and thiourea using Biginelli reaction. Compound **4** was then used as a precursor to synthesize the title compounds **5** and **6** by reaction with 2,5-dimethoxybenzaldehyde under aldol condition, as presented in Figure 2.

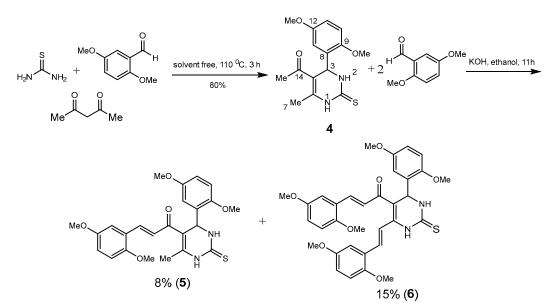


Figure 2. Synthesis reaction of the title compounds.

The synthesis of compound **4** was performed following the previous article [7] with slight modification (differed in purification steps). Compound **4** was then used as precursor to synthesize title compounds **5** and **6** by reaction with 2,5-dimethoxybenzaldehyde under the aldol condition. The reaction progress was monitored by thin layer chromatography (TLC) until completion. During TLC

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experiment, compound **5** was firstly observed, whereas compound **6** appeared later. At the end of the reaction, only these two spots were observed. This observation supported that the first step reaction is the introduction of the benzylidene moiety on the acetyl group [9]. The crude reaction product was then separated on column chromatography using silica gel as the stationary phase and chloroform: ethyl acetate (10:1) as mobile phase. After separation with column chromatography, we obtained 8% yield for compound **5** and 15% for compound **6**.

5-Acetyl-4-(2,5-dimethoxyphenyl)-6-methyl-3,4-dihydropyrimidin-2(1H)-thione (**4**): colorless needle crystal (481 mg, 80%), R_f 0.48 (*n*-hexane: ethyl acetate = 2:1); m.p 266–268 °C (EtOH-H₂O), HRMS (ESI) [M + Na]⁺ calcd. for [C₁₅H₁₈N₂O₃SNa]⁺ = 329.0936 found = 329.0952, IR (cm⁻¹): 3215 (N-H), 2981 (C-H aromatic), 2833 (aliphatic C-H), 1633 (conjugated C=O), 1573, 1496, 1469 (aromatic C=C-), 1249 (C_{aryl-O-Calkyl}), 1176 (-C=S); ¹H-NMR (400 MHz, CDCl₃) δ_H (ppm) 7.93 (s, 1H), 7.46 (s, 1H), 5.75 (d, *J* = 3.3 Hz, 1H), 6.84 (d, *J* = 8.8 Hz, 1H), 6.79 (dd, *J* = 9.0, 2.9 Hz, 1H), 6.59 (d, *J* = 3.0 Hz, 1H), 3.86 (s, 3H), 3.72 (s, 3H), 2.40 (s, 3H), 2.04 (s, 3H); ¹³C-NMR (101 MHz, CDCl₃) δ_H (ppm) 195.8, 174.7, 154.0, 150.6, 143.9, 129.0, 114.4, 113.2, 111.5, 108.0, 56.0, 55.8, 50.9, 29.5, 19.3.

The mass determination of compound 4 exhibited $[M + Na]^+$ ion at m/z 329.0952, which closed to the theoretical value of 329.0936 and was suitable for the molecular formula of C15H18N2O3SNa (Supplementary Material, Figure S1). The analysis of IR spectra showed the existence of N-H amide, C-H aromatic, C-H aliphatic, conjugated C=O, C=C aromatic, Calkyl-O-Caryl ether, and C=S bonds, which were consecutively indicated by bands at \tilde{v} (cm⁻¹) 3215, 2981, 2833, 1633, 1573–1469, 1249, and 1176 [10] (Figure S2). The nuclear magnetic resonance (NMR) data of compound 4 are presented in Table 1. The existence of the dihydropyrimidine ring was assigned by following signals: proton signal of chiral C-4 at δH 5.75 ppm (d, ³JHH = 3.3 Hz, 1H), proton of N-3 at δH 7.46 ppm (s, 1H), proton of N-1 at $\delta_{\rm H}$ 7.93 ppm (s, 1H), and ¹³C signal of C-5 and C-6 at $\delta_{\rm C}$ at 108.0 ppm and 143.9 ppm. The acetyl group was assigned by methyl signal at δ_{H} 2.40 ppm (s, 3H) and carbonyl signal at δ_{C} 195.8 ppm, while the methyl group at C-7 appeared as single signal at δH 2.04 ppm. Three aromatic proton signals with ABX system were assigned, as follows: one aromatic proton with *ortho* and *meta* coupling (δH 6.79 ppm (dd, J = 9.0 and 2.9 Hz, 1H)), one aromatic proton with *ortho* coupling (δ_{H} 6.84 ppm (d, J =8.8 Hz, 1H)), and one aromatic proton with *meta* coupling (δ_{H} 6.59 ppm (d, J = 3.0 Hz, 1H)). Two methoxy groups appeared at δH 3.86 ppm (s, 3H) and δH 3.72 ppm (s, 3H) (Figure S3 for ¹H-NMR and Figure S4 for ¹³C-NMR).

No. Atom	δн (Mult, J Hz)	δ c (ppm)
1	7.93 (s, 1H)	-
2	-	174.7
3	7.46 (s, 1H)	-
4	5.75 (d, J = 3.3 Hz, 1H)	50.9
5	-	108.0
6	-	143.9
7	2.04 (s, 3H)	19.3
8	-	129.0
9	-	154.0
9-OMe	3.86 (s, 3H)	56.3
10	6.84 (d, J = 9.0 Hz, 1H)	114.4
11	6.79 (dd, J = 9.0, 3.0 Hz, 1H)	113.2
12	-	150.6
12-OMe	3.72 (s, 3H)	55.8
13	6.59 (d, J = 3.0 Hz, 1H)	111,5
14	-	195.8
15	2.40 (s, 3H)	29.5

Table 1. Nuclear magnetic resonance (NMR) data of compound 4 in CDCl3.

(*E*)-3-(2,5-*Dimethoxyphenyl*)-1-[[4-(2,5-*dimethoxyphenyl*)-6-*methyl*-2-*thioxo*-1,2,3,4-*tetrahydro-pyrimidin*-5-*yl*]]*prop*-2-*en*-1-*one* (**5**). Yellow needle crystal (38 mg, 8%); m.p 154–156 °C (EtOH-H₂O); UV-Vis (EtOH) λ_{max} (nm) 298 (log ε = 3.98); 372 (log ε = 3.98) (Figure S5); HRMS (ESI): calcd. for [C₂₄H₂₆N₂NaO₅S]⁺ = 477.1460 found = 477.1457; IR (cm⁻¹): 3174 (N-H), 3005 (aromatic C-H), 2978 (aliphatic C-H), 1680 (conjugated C=O), 1645 (conjugated C=C), 1580, 1493, 1427 (aromatic C=C), 1279 (Calkyl-O-Caryl), 1173 (C=S). ¹H-NMR (400 MHz, Acetone-*d*₆) δ_{H} (ppm): 9.19 (s, 1H), 8.30 (s, 1H), 7.77 (d, *J* = 15.8 Hz, 1H), 7.18 (d, *J* = 15.7 Hz, 1H), 7.13 (d, *J* = 2.6 Hz, 1H), 6.97–6.94 (m, 2H), 6.84 (d, *J* = 3.1 Hz, 1H), 6.83–6.79 (m, 2H), 5.92 (d, *J* = 3.4 Hz, 1H), 3.84 (s, 3H), 3.81 (s, 3H), 3.77 (s, 3H), 3.69 (s, 3H), 2.46 (s, 3H). ¹³C-NMR (101 MHz, Acetone-*d*₆) δ_{C} (ppm): 189.05, 176.64, 154.80, 154.62, 153.89, 151.48, 143.63, 137.41, 132.08, 126.76, 125.26, 117.76, 115.29, 113.97, 113.66, 113.54, 112.86, 110.86, 56.44, 56.39, 56.04, 55.79, 51.28, 18.41.

(*E*)-3-(2,5-*Dimethoxyphenyl*)-1-{[4-(2,5-*dimethoxy-phenyl*)-6-((*E*)-2,5-*dimethoxystyryl*)-2-*thioxo*-1,2,3,4*tetrahydropyrimidin*-5-*yl*]}*prop*-2-*en*-1-*one* (**6**): Yellow needle crystal (90 mg, 15%); m.p 212–214 °C (EtOH-H₂O); UV-Vis (EtOH) λ_{max} (nm) 291 (log ε = 4.44), 389 (log ε = 4.21) (Figure S12); HRMS (ESI): calcd. for [C₃₃H₃₄N₂NaO₇S]⁺ = 625.1984 found = 625.1981 ; IR (cm⁻¹): 3198 (N-H), 2945 (aliphatic C-H), 1624 (conjugated C=O), 1578, 1562, 1495 (aromatic C=C), 1276 (C_{alkyl}-O-C_{aryl}), 1225 (C=S). ¹H-NMR (400 MHz, DMSO-*d*₆) δ_{H} (ppm): 10.36 (d, *J* = 1.1 Hz, 1H), 9.41 (dd, *J* = 3.8, 1.4 Hz, 1H), 7.59 (d, *J* = 15.8 Hz, 1H), 7.43 (d, *J* = 16.4 Hz, 1H), 7.37 (d, *J* = 16.4 Hz, 1H), 7.25 (d, *J* = 15.8 Hz, 1H), 7.11 (d, *J* = 3.0 Hz, 1H), 7.10 (d, *J* = 2.7 Hz, 1H), 6.97 (d, *J* = 3.0 Hz, 1H), 5.59 (d, *J* = 4.0 Hz, 1H), 3.73 (s, 3H), 3.69 (s, 3H), 3.68 (s, 3H), 3.66 (s, 3H), 3.60 (s, 3H). ¹³C-NMR (101 MHz, DMSO-*d*₆) δ_{C} (ppm): 188.66, 175.10, 153.21, 153.10, 153.04, 152.41, 151.73, 150.33, 141.15, 135.04, 131.76, 130.79, 127.05, 124.94, 123.61, 120.65, 117.48, 116.07, 114.06, 112.93, 112.88, 112.56, 112.32, 112.31, 112.26, 112.18, 55.91, 55.90, 55.85, 55.33, 55.32, 55.29, 49.91.

Numbering of the molecular structure and selected HMBC correlation of compound **5** are presented in Figure 3, and in Figure 4 for compound **6**. Mass determination of compounds **5** and **6** showed [M + Na]⁺ ion at *m*/*z* 477.1457 for compound **5** (Figure S6) and *m*/*z* 625.1981 for compound **6** (Figure S13), respectively, which were closed to the theoretical value of 477.1460 and 625.1984. Infrared spectra of both compounds **5** and **6** showed vibration bands of N-H amide, C-H aliphatic, conjugated C=O, C=C aromatic, Calkyl-O-Caryl, and C=S bonds consecutively at wave numbers (cm⁻¹): 3174, 2978, 1680 1580, 1279, and 1173 for compound **5** (Figure S7), and 3198, 2945, 1624, 1578, 1276, and 1225 for compound **6** (Figure S14). The existence of two alkene groups with the *trans* configuration of compound **6** were assigned by four doublet proton signals at chemical shift of $\delta_{\rm H}$ 7.59 and 7.25 ppm (*J* = 15.8 Hz) and signals at $\delta_{\rm H}$ 7.43 and 7.37 ppm (*J* = 16.4 Hz) (Figure S15), whereas the existence of an alkene group in compound **5** was assigned by two doublet proton signals at $\delta_{\rm H}$ 7.77 and 7.18 ppm (*J* = 15.8 Hz) (Figure S8).

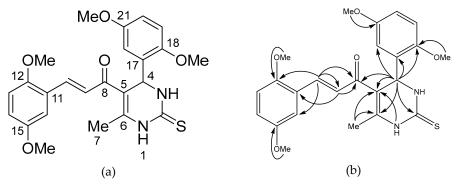


Figure 3. (a) Numbering of the structure, and (b) selected HMBC correlation for compound 5.

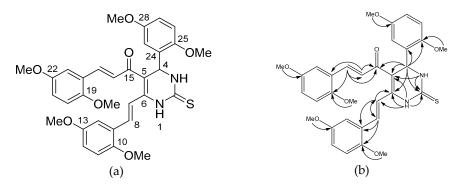


Figure 4. (a) Numbering of the structure, and (b) selected HMBC correlation for compound 6.

Two-dimensional NMR experiments (Heteronuclear Multiple Quantum Coherence (HMQC) and Heteronuclear Multiple Bond Correlation (HMBC)) were performed to prove that aldol condensation proceeded at C-8. The methyl olefinic fragment at $\delta_{\rm H}$ 2.46 ppm proceeded a long range correlation with carbon signal at C 110.86 ppm (C-5) and 143.63 ppm (C-6) (Compound 5). The HMBC correlation tables of compounds 5 and 6 are presented in Table 2, Table 3, respectively. The ¹³C-NMR spectra of compound 5 exhibited 24 signals that represented all carbon atoms of compound 5 (Figure S9), while the ¹³C-NMR spectra of compound 6 showed 33 signals reflected all the carbon atoms of compound 6 (Figure S16). The structures of compounds 5 and 6 are consistent with the molecular mass.

No. Atom	δн (ppm), mult, J (Hz)	δc (ppm)	НМВС
1	9.19 (s, 1H)		C-5
2		176.64	
3	8.30 (s, 1H)		
4	5.92 (d, J = 3.4 Hz, 1H)	51.28	C-2, C-5, C-6, C-8 C-17, C-18, C-22
5		110.86	
6		143.63	
7	2.46 (s, 3H)	18.41	C-5, C-6
8		189.05	
9	7.18 (d, J = 15.8 Hz, 1H)	126.76	C-11
10	7.77 (d, J = 15.8 Hz, 1H)	137.41	C-8, C-9, C-12, C-16
11		125.26	
12		153.89	
12-OCH ₃	3.81 (s, 3H)	56.39	C-12
13	6.83–6.79 (m, 2H)	113.54	
14	6.83–6.79 (m, 2H)	117.76	
15		154.62	
15-OCH3	3.77 (s, 3H)	56.04	C-15
16	7.13 (d, J = 2.6 Hz, 1H)	113.97	C-10, C-12, C-14
17		132.08	
18		151.48	
18-OCH ₃	3.84 (s, 3H)	56.44	C-18
19	6.97–6.94 (m, 2H)	112.86	
20	6.97–6.94 (m, 2H)	113.66	
21		154.80	
21-OCH ₃	3.69 (s, 3H)	55.79	C-21
22	6.84 (d, J = 3.1 Hz, 1H)	115.29	C-18, C-20, C-21

Table 2. NMR data of compound 5 in Acetone-d6.

Table 3. NMR data of compound 6 in DMSO-d6.

No. Atom	δн (ppm), mult, J (Hz)	δ c (ppm)	НМВС
1	10.36 (d, J = 1.4 Hz, 1H)		C-2, C-5, C-6, C-7
2		175.10	
3	9.41 (dd, <i>J</i> = 4.0, 1.4 Hz, 1H)		C-2, C-5
4	5.59 (d, J = 4.0 Hz, 1H)	49.91	C-2, C-5, C-6, C-15, C-24, C-25
5		112.18	
6		141.15	
7	7.37 (d, J = 16.4 Hz, 1H)	120.65	C-6, C-8, C-9
8	7.43 (d, J = 16.4 Hz, 1H)	131.76	C-6, C-7, C-10
9		124.94	
10		151.73	
10-OCH3	3.69 (s, 3H)	55.33	C-10
11	6.95–6.93 (m, 3H)	112.88	
12	6.89 (dd, J = 9.0, 3.0 Hz, 1H)	116.07	C-10
13		153.10	
13-OCH ₃	3.64 (s, 3H)	55.90	C-13
14	7.11 (d, <i>J</i> = 3.0 Hz, 1H)	112.31	C-8, C-10, C-12
15		188.66	
16	7.25 (d, J = 15.8 Hz, 1H)	127.05	C-15, C-18
17	7.59 (d, J = 15.8 Hz, 1H)	135.04	C-15, C-16, C-19
18		123.61	
19		152.41	
19-OCH3	3.68 (s, 3H)	55.85	C-19
20	6.95–6.93 (m, 3H)	112.93	
21	6.95–6.93 (m, 3H)	117.48	
22		153.21	
22-OCH3	3.60 (s, 3H)	55.91	C-22
23	7.10 (d, <i>J</i> = 2.7 Hz, 1H)	112.26	C-17, C-21, C-25
24		130.79	
25		150.33	
25-OCH3	3.73 (s, 3H)	55.29	C-25
26	6.97 (d, J = 8.8 Hz, 1H)	112.56	
27	6.85 (dd, J = 8.9, 3.1 Hz, 1H)	112.32	C-25, C-29
28		153.04	
28-OCH ₃	3.66 (s, 3H)	55.32	C-28
29	6.70 (d, J = 3.0 Hz, 1H)	114.06	C-25, C-27

3. Materials and Methods

3.1. General

The chemicals were provided from the commercial sources with pro analysis or pro synthesis grade and were used without prior purification. Thin layer chromatography was conducted on silica gel GF₂₅₄ plate (E Merck, Darmstadt, Germany) for the monitoring of the reaction progress and test of purity. The spots of TLC were identified by UV lamp (λ 254 nm). The UV spectrum was recorded on a UV-Vis spectrophotometer type UV-1800 (Shimadzu, Kyoto, Japan). The mass spectra were measured on a HRESIMS QTOF micrOTOF-Q II Bruker Daltonics (Billerica, MA, USA). The Fourier transform infrared (FTIR) spectrum was recorded on an IRTracer100 spectrophotometer (Shimadzu, Kyoto, Japan) while using the diffuse reflectance method. The nuclear magnetic resonance (NMR) spectrum (¹H-, and ¹³C-APT) was recorded on a JEOL JNM-ECS400 spectrometer (at 400 and 100 MHz) (JEOL Ltd., Tokyo, Japan), with Acetone-*d*₆, CDCl₃ and DMSO-*d*₆ as the solvent and internal standard.

In a screw-capped vial, the mixture of 2,5-dimethoxybenzaldehyde (0.331 g, 2 mmol), acetyl acetone (0.2 g, 2 mmol), and thiourea (0.19 g, 2.5 mmol) was heated without stirring at 110 °C for 3 h [7]. The reaction mixture was then allowed to cool to room temperature, dissolved in ethanol, then added with water dropwise, and the precipitate was filtered off. The solid was then recrystallized using aqueous ethanol.

3.3. Synthesis of the Title Compounds

Compound 4 (0.306 g, 1 mmol), 2,5-dimethoxybenzaldehyde (0.414 g, 2.5 mmol), KOH (0.056 g, 1 mmol), and ethanol (10 mL) were placed in a three neck round bottom flask that was equipped with a reflux condenser. The reaction mixture was refluxed at 50 °C for 11 h and then cooled to room temperature. The reaction mixture was poured to ice-water, neutralized with HCl-ethanolic solution, and the precipitate was filtered off. The crude product was then subjected to a silica gel column chromatography while using *n*-hexane:ethyl acetate (10:1) as eluent to furnish the title compounds (compounds 5 and 6).

4. Conclusions

In conclusion, we have synthesized two new compounds (*E*)-3-(2,5-Dimethoxyphenyl)-1-{[4-(2,5-dimethoxyphenyl)-6-methyl-2-thioxo-1,2,3,4-tetrahydro-pyrimidin-5-yl]}prop-2-en-1-one (compound **5**) and (*E*)-3-(2,5-Dimethoxyphenyl)-1-{[4-(2,5-dimethoxy-phenyl)-6-((*E*)-2,5-dimethoxystyryl)-2-thioxo-1,2,3,4-tetrahydropyrimidin-5-yl]}prop-2-en-1-one (compound **6**) under aldol condensation condition.

Supplementary Materials: FTIR, HRMS (ESI), ¹H-NMR and ¹³C-APT-NMR, HMBC and HMQC spectra of the prepared compounds are available on line.

Author Contributions: H.S. brought the idea, managed the research, and prepared the manuscript. N.K., E.S., and Y.A. conducted the experiment, K.U.H. brought the idea and did the structure elucidation, I.I. did structure elucidation, A.N.K. corrected the manuscript.

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