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Abstract Efficient large-scale and feasible industrial synthesis of the 1-oxacephem core structure from 6-aminopenicillanic acid (6-APA) has been reported for s decades. Via the industrial synthesis route, a byproduct (compound 9) containing a butenolide unit was purified and characterized by NMR and HRMS [].Rear (This article belongs to the collection Molecules from Catalytic Processes (/journal/molbank/special_issues/molecules_catalytic.))	everal d more.
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Abstract (3-Ammonio-2,2-dimethylpropyl)carbamate dihydrate was synthesised. The title compound was characterised by single crystal X-ray diffraction and IF spectroscopy. It has been demonstrated that a mixture of dilute acetic acid and 2,2-dimethyl-1,3-diaminopropane is able to capture CO <sub>2</sub> spontaneously from the atmosphere. An intramolecular hydrogen bond [] Read more. (This article belongs to the Section Structure Determination (/journal/molbank/sections/structure_determination_molbank))	≀-/Raman- ;
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1-[5-(4-Tolyl)-1,3,4-oxadiazol-2-yl]methanamine (/1422-8599/2018/3/M1014) by C Ganesh Shimoga (https://sciprofiles.com/profile/author/M2ZEa1RIVFRDZzcvQkRSTmxkZnpkRjBCM2pjQ2dUOHQ4K1IDRVZQc2hQRT0=), C Eun-Jae Shin (https://sciprofiles.com/profile/451838) and C Sang-Youn Kim (https://sciprofiles.com/profile/289515) Molbank 2018, 2018(3), M1014; https://doi.org/10.3390/M1014 (https://doi.org/10.3390/M1014) - 24 Aug 2018 Viewed by 706 Abstract 1/5-(4-Tolyl)-1 3 4-oxadiazol-2-yllmethanamine (3) has been successfully synthesized by reacting p-tolyic hydrazide (1) and dycine (2) via the polyof	oosphoric
acid condensation route. The course of the reaction was found to be high yielding (87%) and the title compound [] Read more. (This article belongs to the Special Issue Heteroatom Rich Organic Heterocycles (/journal/molbank/special issues/Heteroatom Heterocycles ))	losphone
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▶ Show Figures         (molbank/molbank-2018-M1014/article_deploy/molbank-2018-M1014-ag.jpeg)         ○pen Access Short Note         £[3-(4-Bromophenyl)-1-(2.5-dimethoxyphenyl)-3-oxopropyl]-1.3-dimethylpyrimidine-2.4.6(1 <i>H</i> .3 <i>H</i> .5 <i>H</i> )-tri-one (/1422-8599/2018/3/M1013)         by @ Hery.Suwito (https://sciprofiles.com/profile/3283) .         @ Ria Hesty Purnama Sari (https://sciprofiles.com/profile/author/RillMmtwcFo0WFV2bUp1akdGSmpJY3M2ODBpZmlJZXZVVndoVGtJUDFBZz0=) .         @ Kautsar UI Haq.(https://sciprofiles.com/profile/author/NnE3V1d1enBkanp0Z0F0ekxFcXgwei9MdkN3NH/VWmRKMFJhUk5SR283SDA4aUFPYV1S1I         and         @ Alfinda Novi Kristanti (https://sciprofiles.com/profile/325866)         Molbank 2018, 2018(3), M1013; https://doi.org/10.3390/M1013 (https://doi.org/10.3390/M1013) - 23 Aug 2018         Viewed by 710         Abstract       The title compound was prepared by a two-step reaction. The first step was the formation of a chalcone derivative using Claisen-Schmidt condensation was fullesen soft by the Michael addition of the formed chalcone with 1,3-dimethylbarbituric acid. The structure of the prepared compound was [] Read more.         (This article belongs to the collection Molecules from Catalytic Processes (/journal/molbank/special_issues/molecules_catalytic.))         Open Access Short Note       Image: Age Advector Adve	013/pdf) MrVzdVZHN on, which 012/pdf)
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Momoyo Kawamoto (https://sciprofiles.com/profile/author/dDJJeHpoVFdJNTZldGR4TmtNSWFXREplcjRIRHpnNnV1YT	c1dTBpZHJBZz0=) and
( <u>Yoo Tanabe (https://sciprofiles.com/profile/122999)</u>	
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$a_{\rm h}$ alogue, was performed through a clean S <sub>N</sub> 2 displacement reaction using available AcSK with tris[2-(2-methoxyethoxy)]ethylami	ine (TDA-1), starting from commercially
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Abstract Thia-Michael addition of 2-[(4-hydroxy-3,5-dimethoxyphenyl)methylidene]hydrazine-1-carbothioamide (1) with maleic and	hydride results in the formation of the title
compound 2-{[(4-hydroxy-3,5-dimethoxyphenyl)methylidene]hydrazinylidene}-4-oxo-1,3-thiazolidin-5-yl acetic acid 2. The precurso	or <b>1</b> is synthesized by the reaction of 4-
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Abstract N-[1-Hydrazinyl-3-(1H-indol-3-yl)-1-oxopropan-2-yl]-4-methylbenzenesulfonamide (1) on cyclization with carbon disulfide	e in ethanolic potassium hydroxide affords
<i>N</i> -[2-(1 <i>H</i> -indol-3-yl)-1-(5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)ethyl]-4-methylbenzenesulfonamide (2) in 84% yield. The structur	e of compound 2 was supported by mass
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A new havehold derivative, namely 5,7-dirightoxy-3,6-dimethoxy-3,4-methylenedioxyllavone (1), was isolated from the leaves of <i>I</i> structure of <b>1</b> was elucidated based on their UV IR HRESIMS and 1D and 2D NMR spectral data <b>Full article (/1422-8599/2018</b> )	(Blume) T.G. Harriey. The
(This article belongs to the Section <u>Natural Products (/journal/molbank/sections/natural_products_molbank</u> ))	<u>, , , , , , , , , , , , , , , , , , , </u>
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Molbank 2018, 2018(3), M1006; https://doi.org/10.3390/M1006 (https://doi.org/10.3390/M1006) - 20 Jul 2018	<u>55</u> )
<u>Cited by 4 (/1422-8599/2018/3/M1006#citedby)</u>   Viewed by 736	
Abstract Polyphenols are well-known health promoting agents, but they have some limitations due to their spontaneous oxidation	. This evidence has limited their use as
drugs in recent years. In this field, several chemical modifications have been proposed to overcome these restrictions; among the	se, esterification [] Read more.
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<u>Abstract</u> The 1-(4-hexylbenzoyl)-3-methylthiourea compound has been successfully synthesized by reacting 4-hexylbenzoyl chl method using a triethylamine catalyst. The 1-(4-hexylbenzoyl)-3-methylthiourea compound was identified by UV-visible, FT-IR, <sup>1</sup> . From the activity test on four cancer [] Read more.	loride and 1-methylthiourea via the reflux <sup>3</sup> C/ <sup>1</sup> H-NMR and Mass spectrophotometry.
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5-Hydroxy-3-(4-hydroxyphenyl)-8,8-dimethyl-6-(3-methylbut-2-enyl)pyrano[2,3-h]chromen-4-one (/1422-8599/2018/3/M10	<u>104)</u>
by ( Giovanni Ribaudo (https://sciprofiles.com/profile/828603), ( Alberto Ongaro (https://sciprofiles.com/profile/author/T3NWQTIVYjZHYnBkU2QwK29rQklkd2NiY1R0MU5YelpIdHVIVk and ( Giuseppe Zagotto (https://sciprofiles.com/profile/189634)	kU3VWpodjM5ZG5yOGIQM3Q2RWFhZUR
Moldank 2018, 2018(3), M1004; <u>https://doi.org/10.3390/M1004 (https://doi.org/10.3390/M1004)</u> - 09 Jul 2018 Viewed by 682	
Abstract. Natural and semi-synthetic compounds are being studied as novel phosphodiesterase 5 (PDE5) inhibitors for the treated hypertension, and lower urinary symptoms. <i>Maclura pomifera</i> is a source of flavonoids, one of the main classes of molecules involver. (This article belongs to the Section Natural Products (/journal/molbank/sections/natural_products_molbank))	ment of erectile dysfunction, pulmonary /estigated for these purposes. [] Read
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3.3'-(Diazene-1.2-divl)bis[4-(nitroamino)-1.2.5-oxadiazole 2-oxide] (/1422-8599/2018/3/M1003)	
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Abstract The nitramino derivatives of furoxans are of specific interest as precursors for the preparation of high energy salts with communication, the 3,3'-(diazene-1,2-diyl)bis[4-(nitroamino)-1,2,5-oxadiazole 2-oxide] was prepared via nitration of available 4,4 yield of the target compound was achieved [] Read more. (This article belongs to the Special Issue Heteroatom Rich Organic Heterocycles (/journal/molbank/special_issues/Hetero	n nitrogen-rich cations. In this I'-diamino-3,3'-diazenofuroxan; the best Natom_Heterocycles_))
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2-12-Methyl-5-phenyl-1-(3.4.5-trimethovyphenyl)-1H-pyrrol-3-yl]-2-ovo-N-(pyridin-4-yl) acetamide (/1422-8599/2018/3/1002	2)
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<u>Abstract</u> We synthesized 2-[2-methyl-5-phenyl-1-(3,4,5-trimethoxyphenyl)-1 <i>H</i> -pyrrol-3-yl]-2-oxo- <i>N</i> -(pyridin-4-yl) acetamide <b>4</b> as indibulin and combretastatin scaffolds, which are known anti-mitotic agents, using a multistep reaction. We tested its cytotoxic ac namely. MCF-7, T47-D. <b>1</b> ] <b>Read more.</b>	a novel compound derived from the ctivity against three breast cancer cell lines,
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(E)-3-[3-(2-Butoxyquinolin-3-vl)acrvlovl]-2-hvdroxy-4H-chromen-4-one (/1422-8599/2018/3/1001)	
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, <b>Q</b> Jairo Quiroga (https://sciprofiles.com/profile/68297) and <b>P</b> Braulio Insuasty (https://sciprofiles.com/profile/46790) Molbank 2018, 2018(3), 1001; https://doi.org/10.3390/M1001 (https://doi.org/10.3390/M1001) 21 Jun 2018 Viewed by 834	
Abstract The coumarinyl-quinolinylchalcone hybrid ( <i>E</i> )-3-[3-(2-butoxyquinolin-3-yl)acryloyl]-2-hydroxy-4 <i>H</i> -chromen-4-one <b>3b</b> was Schmidt condensation reaction between 3-acetyl-4-hydroxy-2 <i>H</i> -chromen-2-one <b>1</b> and 2-butoxyquinoline-3-carbaldehyde <b>2</b> in me	as prepared in good yield from a Claisen- ethanol at reflux and catalyzed by KOH
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E-Mail () Website (http://people.uleth.ca/~boere/)

Editor-in-Chief

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**Interests:** organoelement chemistry; sulfur-nitrogen compounds; heterocycles; amidines; heteroamidines; freeradicals; electrochemistry; supramolecular chemistry; structural methods; crystallography; electron paramagnetic resonance spectroscopy

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Former Editor-in-Chief

Department of Pharmaceutical Chemistry, University of Vienna, Althanstrasse 14, 1090 Vienna, Austria Tel. +436642204677

Interests: medicinal heterocyclic chemistry; pyridazines; carbazoles; nitrogen hetarenes

\* former Editor-in-Chief from 2007 up to 30 September 2019; The Editorial Board of Molecules also governs Molbank. Former Editor-in-Chief: up to 2001 Dr. Luc Patiny; in 2002 Prof. Dr. Bruce A. Hathaway; in 2003 Reto Mueller (organicchemistry.org)

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Dr. R. Alan Aitken

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School of Chemistry, University of St Andrews North Haugh, St Andrews Fife, KY16 9ST, UK Tel. (01334) 463865; Fax: (01334) 463865

**Interests:** synthesis; synthetic use of flash vacuum pyrolysis; heterocyclic chemistry; reactive intermediates; organophosphorus; organosulfur; heavier main group chemistry



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Interests: free radical organic and polymer chemistry; heterocyclic and medicinal chemistry

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(/journal/molecules/special\_issues/synthetic\_hetero)



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Department of Chemistry, Durham University, University Science Laboratories, South Road, Durham DH1 3LE, UK **Interests:** organic synthesis and methodology; automation; chemical reactors; flow chemistry; enabling technologies; immobilised reagents and scavengers; heterocyclic chemistry

#### Dr. Jose Berna

# E-Mail () Website (http://webs.um.es/ppberna/miwiki/doku.php)

Department of Organic Chemistry, Faculty of Chemistry, University of Murcia, 30100 Murcia, Spain Interests: self-assembly; mechanically interlocked molecules (rotaxanes and catenanes); hydrogen bond; template synthesis; molecular recognition; supramolecular chemistry

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# Prof. Dr. Bartolo Gabriele <u>E-Mail ()</u> <u>Website1</u>

# (http://www.unical.it/portale/strutture/dipartimenti\_240/ctc/didattica/homedid/docenti/ordinari/gabriele/) Website2 (https://www2.scopus.com/authid/detail.uri?authorld=7003277013)

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**Interests:** new syntheses of high value added molecules through catalytic assembly of simple units; innovative syntheses of heterocyclic molecules of pharmaceutical, agrochemical, or applicative interest; carbonylation chemistry; use of non-conventional solvents in organic synthesis; synthesis and semi-synthesis of bioactive compounds of pharmaceutical or agrochemical interest; synthesis of new materials for advanced applications; extraction, characterization, and evaluation of the biological activity of bioactive principles from natural matrices

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# Prof. Dr. Panayiotis A. Koutentis

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Department of Chemistry, University of Cyprus, P. O. Box 20537, 1678 Nicosia, Cyprus Tel. 0035722892783

**Interests:** heterocyclic chemistry; sulfur-nitrogen heterocycles; synthetic methods; azaacenes; zwitterionic acenes; stable organic radicals; biologically active heterocycles; isothiazoles; 1,2,3-dithiazoles; 1,2,6-thiadiazines;1,2,4-benzotriazines

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Institut für Medizinische und Pharmazeutische Chemie, Technische Universität Braunschweig, Beethovenstraße 55, 38106 Braunschweig, Germany

Accept (/accept\_cookies) Interests: medicinal chemistry; pharmaceutical chemistry; synthesis of heterocycles; protein kinase inhibitors



# Dr. Raffaella Mancuso

# <u>E-Mail ()</u>

Laboratory of Industrial and Synthetic Organic Chemistry (LISOC), Department of Chemistry and Chemical Technologies, University of Calabria, Via Pietro Bucci 12/C, 87036 Arcavacata di Rende (CS), Italy **Interests:** innovative syntheses of high value molecules through catalytic process; new syntheses of heterocyclic compounds of pharmaceutical interest; carbonylation catalyzed chemistry; application of unconventional solvents in advanced organic synthesis; synthesis of novel materials for advanced applications

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#### Prof. Dr. Ge Meng

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Chemistry Department, Fudan University, Shanghai 200433, China

Fax: +86 029 82655424

**Interests:** medicinal heterocyclic chemistry; indoles; indazoles; pyridines; thiazoles; thiadiazoles; pyrroles; computer aided drug design; molecular probe



# Prof. Dr. Hideto Miyabe

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School of Pharmacy, Hyogo University of Health Sciences, 1-3-6 Minatojima, Chuo-ku, Kobe 650-8530, Japan **Interests:** organic synthesis and methodology; radical reactions; organocatalysis; asymmetric catalysis; photochemistry; aryne chemistry

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Organic Pharmaceutical Chemistry, Department of Medicinal Chemistry, BMC, Uppsala University, Box 574, SE-751 23 Uppsala, Sweden

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Interests: heterocyclic chemistry; multicomponent reactions; catalysis; medicinal chemistry; synthetic methodology Special Issues and Collections in MDPI journals:

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De of Chemistry, Pavillon Alexandre-Vachon, 1045, Avenue de la Médecine, Université Laval, Québec, G1V

Tel. +1 418 656 5034; Fax: +1 418 656 7916

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#### **Dr. Luc Patiny**

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Director of Chemical Information, Institute of Chemical Sciences and Engineering (ISIC), Ecole Polytechnique Fédérale de Lausanne (EPFL), CH-1015 Lausanne, Switzerland

**Interests:** chemoinformatics; chemical information; infrastructure for chemical research; online processing and data mining

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<u>E-Mail ()</u> Website

#### (https://www.docenti.unina.it/#!/professor/56494e43454e5a4f5049434349414c4c4950434356434e3538453237423

Department of Chemical Sciences, Università degli Studi di Napoli Federico II, Via Cintia 21, 80126 Napoli, Italy Interests: organic and medicinal chemistry; organic synthesis; catalytic oxidative processes; marine natural products; nucleosides chemistry

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# Prof. Dr. Oleg A. Rakitin

<u>E-Mail ()</u> <u>Website (http://zioc.ru/institute/laboratories/laboratory-of-polysulphur-nitrogen-heterocycles-n31?</u> lang=en)

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2. Nanotechnology Education and Research Center, South Ural State University, 454080 Chelyabinsk, Russia Tel. +7-499-1355327

**Interests:** heterocyclic chemistry; sulfur-nitrogen heterocycles; selenium heterocycles; synthetic methods; biologically active compounds; organic sensitizers for DSSCs and OLEDs; sulfur monochloride

#### Special Issues and Collections in MDPI journals:

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#### Dr. Bernd Schneider

E-Mail () Website (http://www.ice.mpg.de)

Max Planck Institute for Chemical Ecology, Hans-Knöll-Str. 8, Beutenberg Campus, 07745 Jena, Germany We use cookies on our website to ensure you get the best experience. Interests: natural products chemistry; chemical ecology; plant natural products; NMR of small molecules Read more about our cookies <u>here (/about/privacy)</u>.

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Interests: supramolecular chemistry; gels; molecular materials and the molecular solid state; pharmaceutical solid

#### Prof. Dr. Ping-Jyun Sung

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2. Graduate Institute of Marine Biology, National Dong Hwa Univesity, Pingtung 944, Taiwan

3. Department of Marine Biotechnology and Resources, National Sun Yat-sen University, Kaohsiung 804, Taiwan

4. Chinese Medicine Research and Development Center, China Medical University Hospital, Taichung 404, Taiwan

5. Graduate Institute of Natural Products, Kaohsiung Medical University, Kaohsiung 807, Taiwan

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Interests: marine natural products; marine chemical ecology; bioactive substances from cultured marine invertebrates Special Issues and Collections in MDPI journals:

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Dr. Hidenori Tanaka

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Short Note



# 5-[3-(4-Bromophenyl)-1-(2,5-dimethoxyphenyl)-3oxopropyl]-1,3-dimethylpyrimidine-2,4,6(1*H*,3*H*,5*H*)-tri-one

#### Hery Suwito \*, Ria Hesty Purnama Sari, Kautsar Ul Haq and Alfinda Novi Kristanti

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Received: 2 August 2018; Accepted: 20 August 2018; Published: 23 August 2018



**Abstract:** The title compound was prepared by a two-step reaction. The first step was the formation of a chalcone derivative using Claisen–Schmidt condensation, which was followed by the Michael addition of the formed chalcone with 1,3-dimethylbarbituric acid. The structure of the prepared compound was established by spectral data: FTIR, HRESIMS, <sup>1</sup>H- and <sup>13</sup>C-NMR.

Keywords: chalcone derivative; 1,3-dimethylbarbituric acid; Michael addition

#### 1. Introduction

Compounds possessing a dihydropyrimidine (DHPM) core attract the interest of researchers, either due to their wide spectrum bioactivities or from a synthesis point of view. The core of dihydropyrimidine can be constructed from a C–C–C and N–C–N scaffold, which may be formed by a Biginelli reaction [1] or cyclocondensation between an enone and urea or its analog. In general, preparation of a dihydropyrimidine derivative through cyclocondensation can be achieved by a reaction between chalcone as the source of the C–C–C unit and urea or its analog as the source of the N–C–N unit.

The molecular structure of dihydropyrimidine from a Biginelli product has a close resemblance to Hantsch 1,4-dihydropyridine, both being aza-analogues of nifedipine, which is well known as a calcium channel modulator [2]. Furthermore, DHPM derivatives are also known to exhibit antihypertensive [3], potassium channel antagonistic [4], antifilarial [5], anti-HIV [6,7], and antitumor [8] activities.

In continuing of our research, we intended to synthesize a 5,7-diphenyl-5,8-dihydropyrido [2,3-*d*]pyrimidine-2,4(1*H*,3*H*)-dione (**3**) derivative through a Hantzsch cyclocondensation-type reaction constructed from a chalcone derivative (serving as the C–C–C unit) and 1,3-dimethylbarbituric acid (serving as the dioxopyrimidine unit) in the presence of ammonium chloride using triethylamine (TEA) acting as a Lewis base catalyst. Unfortunately, based on the spectroscopic evidence, the reaction stopped at the Michael addition process and did not proceed further to cyclocondensation. In this paper, we describe the synthesis and the characterization of the title compound.

#### 2. Results

The title compound was synthesized in a two-step reaction. The synthesis was started by chalcone preparation employing Claisen–Schmidt condensation according to the procedure reported by Suwito et al., (2014) [9]. The next step was Michael addition between the chalcone with 1,3-dimethylbarbituric acid using TEA as a catalyst. The reaction process is displayed in Figure 1 below. In this article, we discuss only the preparation and structural characterization of compound **2** because compound **1** is already known. The title compound **2** was obtained as a white solid (171 mg; 34%).



Figure 1. The reaction process for the synthesis of the title compound.

5-[3-(4-Bromophenyl)-1-(2,5-dimethoxyphenyl)-3-oxopropyl]-1,3-dimethyl-pyrimidine-2,4,6-trione: White solid (171.7 mg, 34%); Rf 0.512 (CHCl<sub>3</sub>:ethyl acetate:*n*-hexane = 2:1:1); HRESIMS  $[M - H]^-$  calculated for C<sub>23</sub>H<sub>22</sub>N<sub>2</sub>O<sub>6</sub>Br 501.0661, found 501.0774; IR (DRS, KBr, cm<sup>-1</sup>): 3080 (C–H aromatic), 2958 (m, CH aliphatic), 1745 (str, C=O ketone), 1618 (str, C=O amide), 1581.63 (str, C=C aromatic), 1203 (str, C–O–C ether), and 717 (C–Br); <sup>1</sup>H-NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ<sub>H</sub> (ppm) 7.92 (d, *J* = 8.5 Hz, 2H), 7.75 (d, *J* = 8.4 Hz, 2H), 6.85 (d, *J* = 8.9 Hz, 1H), 6.77 (dd, *J* = 8.9 Hz, *J* = 2.9 Hz, 1H), 6.61 (d, *J* = 2.9 Hz, 1H), 4.43 (ddd, *J* = 8.3 Hz, *J* = 6.2 Hz, *J* = 4.4 Hz, 1H), 3.90 (dd, *J* = 18.3 Hz, *J* = 8.3 Hz, 1H), 3.79 (d, *J* = 4.4 Hz, 1H), 3.65 (s, 3H), 3.64 (s, 3H), 3.49 (dd, *J* = 18.3 Hz, *J* = 6.2 Hz, 1H), 2.93 (s, 3H), 2.92 (s, 3H); <sup>13</sup>C-NMR (101 MHz, DMSO-*d*<sub>6</sub>) δ<sub>C</sub> (ppm)197.8, 168.7, 168.3, 153.3, 151.9, 151.3, 136.1, 132.3, 130.5, 128.2, 127.9, 115.2, 113.3, 112.1, 56.3, 55.9, 53.2, 40.5, 37.1, 28.3, 28.2.

Two peaks of the molecular negative ion at m/z = 501.0774 and 503.0763 of the HRESIMS spectrum indicate that the prepared compound contains a bromine atom with the molecular formula  $C_{23}H_{22}N_2O_6Br$  and possesses 13 degrees of unsaturation (Supplementary Materials, Figure S1). The analysis of IR spectra showed the existence of C–H aromatic, C–H aliphatic, C=O ketone, C=O amide, C=C aromatic,  $C_{alkyl}$ –O– $C_{aryl}$  ether, and C–Br bonds which were indicated consecutively by peaks at  $v_{max}$  (cm<sup>-1</sup>) 3080, 2958, 1745, 1618, 1581, 1203, and 717, respectively (Supplementary Materials, Figure S2).

Spectroscopic data of <sup>1</sup>H-NMR (Table 1) showed an ABMX spin system which represented a –CH<sub>2</sub>–CH–CH– fragment of the scaffold of the chalcone adduct with an active methylene. For this system, two signals at  $\delta_{\rm H}$  3.90 (dd, 1H) and 3.49 ppm (dd, 1H) represented the existence of methylene diastereotopic fragment neighboring with a carbonyle group. Both signals showed a geminal coupling (*J* = 18.3 Hz). Protons appearing at  $\delta_{\rm H}$  3.90 ppm (H<sub>a</sub>) and 3.49 ppm (H<sub>b</sub>) showed consecutively trans coupling (*J* = 8.3 Hz), and cis coupling (*J* = 6.2 Hz) with a proton at  $\delta_{\rm H}$  4.43 ppm. The signal at

 $\delta_{\rm H}$  4.43 ppm (ddd) represented a benzylic proton attached to a diastereotopic methylene, while the barbiturate fragment is exhibited by the existence of a vicinal coupling (*J* = 4.4 Hz) with a signal at  $\delta_{\rm H}$  3.79 ppm. This signal represented a proton flanked by the two carbonyl groups of a barbiturate fragment. The existence of an aromatic ring was determined by five aromatic signals: two signals at  $\delta_{\rm H}$  7.92 and 7.75 ppm appeared as a doublet with *J* = 8.5 Hz and integration of both. Both signals formed an AA'XX' spin system, which lead to the conclusion of a benzene fragment possessing two substituents at *para* position. While three other signals—appearing at  $\delta_{\rm H}$  6.85 (d, *J* = 8.9 Hz, 1H),  $\delta_{\rm H}$  6.77 (dd, *J* = 8.9 Hz, *J* = 2.9 Hz, 1H), and  $\delta_{\rm H}$  6.61 (d, *J* = 2.9 Hz, 1H)—build an ABX spin system and represent a three substituted aromatic ring at position 1,2, and 4. Furthermore, the two signals at  $\delta_{\rm H}$  3.65 ppm (s, 3H) and  $\delta_{\rm H}$  3.64 (s, 3H) ppm indicated a methoxy proton attached at an aromatic ring, while the signals at  $\delta_{\rm H}$  2.93 ppm (s, 3H) and  $\delta_{\rm H}$  2.92 ppm (s, 3H) were signals of a methyl proton attached at the nitrogen atom of the barbiturate fragment (Supplementary Materials, Figure S3). The <sup>13</sup>C-NMR spectra (Table 1) showed 21 signals and represented all carbon atoms of the prepared compound (Supplementary Materials, Figure S4).

No. Atom	δ <sub>H</sub> (Mult, J Hz)	δ <sub>C</sub> (ppm)	НМВС
1	-	197.9	
2	$H_a = 3.90 (dd, 1H, J = 18.3 Hz, J = 8.3 Hz)$ $H_b = 3.49 (dd, 1H, J = 18.3 Hz, J = 6.2 Hz)$	40.5	C-1, C-3, C-4, C-1', C-1"
3	4.43 (ddd, 1H, J = 8.3 Hz, J = 6.2 Hz, J = 4.4 Hz)	37.1	C-2, C-4, C-5, C-9, C-1", C-6"
4	3.79 (d, 1H, J = 4.4 Hz)	53.2	C-2, C-3, C-5, C-9, C-1"
5	-	168.7	
6	-	-	
7	-	151.3	
8	-	-	
9	-	168.3	
10	2.93 (s, 3H)	28.21	C-9, C-7
11	2.92 (s, 3H)	28.31	C-5, C-7
1'	-	127.9	
2', 6'	7.92 (d, 2H, J = 8.5 Hz)	132.3	C-1, C-1′, C-3′
3', 5'	7.75 (d, 2H, J = 8.4 Hz)	130.5	C-1, C-1', C-2', C-4'
4'	-	136.1	
1″	-	128.3	
2″	-	153.4	
3″	6.85 (d, 1H, J = 8.9 Hz)	112.1	C-3, C-1", C-2"
$4^{\prime\prime}$	6.77 (dd, 1H, J = 8.9 Hz, J = 2.9 Hz)	113.3	C-2", C-5", C-6"
5"	-	151.9	
6''	6.61 (d, 1H, J = 2.9 Hz)	115.2	C-3, C-2", C-4", C-5"
7″	3.64 (s, 3H)	56.3	C-2″
8″	3.65 (s, 3H)	55.9	C-5″

Table 1. NMR data of the target compound in DMSO-d6	•
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The scaffold of the Michael adduct was assigned by the HMBC experiment which showed a correlation of the proton at C-3 with some carbon atoms; the methylene carbon ( $\delta_C$  40.5 (C-2)), methyne carbon ( $\delta_C$  53.2 (C-4)), amide carbonyl of *N*,*N'*-dimethylbarbiturate ring ( $\delta_C$  168.3 (C-9)), and ( $\delta_C$  168.7 (C-5)). The existence of a methylene group at C-2 was proved by a correlation of a proton with carbonyl ketone ( $\delta_C$  197.9 (C-1)), C-3 ( $\delta_C$  37.1), and a long-range correlation of the C-2 proton with C-4 ( $\delta_C$  53.2) and C-1" ( $\delta_C$  128.3). Additionally, the CH position of C-4 was assigned by the correlation of the C-4 proton with the amide carbonyl of the 1,3-dimethylpyrimidine ring ( $\delta_C$  168.3 (C-9)) and ( $\delta_C$  168.7 (C-5)), and the proton correlation with C-3 ( $\delta_C$  37.1) and C-2 ( $\delta_C$  40.5). The proton–carbon correlations of the HMBC experiment were suitable with the molecular structure of the prepared compound and are displayed in Figure 2 and Figure S5 (Supplementary Materials). Based on the structure elucidation, the prepared compound.



Figure 2. (a) Structure numbering, and (b) HMBC correlations of the prepared compound.

#### 3. Materials and Methods

#### 3.1. General

All reagents and solvents, purchased from E. Merck (Darmstadt, Germany) or Sigma Aldrich (St. Louis, MO, USA), were used without further purification. Reaction progress was monitored by thin-layer chromatography on silica gel GF254 aluminum sheets (0.25 mm) using various developing systems. Spots were detected under UV light ( $\lambda$  254 nm). IR spectrum was recorded in KBr powder with the diffuse reflectance method on Fourier-transform Infrared spectrometer Shimadzu IRTracer100 (Kyoto, Japan). The mass spectrum was recorded on an High-resolution mass spectrometer, Waters LCT Premier XE (Santa Clara, CA, USA). NMR spectra (<sup>1</sup>H-, <sup>13</sup>C-NMR, HMQC and HMBC) were recorded using the JEOL JNM-ECS400 (Tokyo, Japan) with CDCl<sub>3</sub> as a solvent and internal standard.

#### 3.2. Synthesis of Chalcone Derivative

The synthesis of chalcone derivative was conducted following the procedure reported by Suwito et al. [9]. A mixture of 6 mmol 4'-bromoacetophenone, 6 mmol 2,5-dimetoxybenzaldehyde, and 30 mL ethanol was placed in a three neck round bottom flask (equipped with a reflux condenser, thermometer, and dropping funnel), stirred, and cooled under 10 °C. To the reaction mixture, 6 mL NaOH 40% solution was added dropwise and the temperature was kept under 10 °C, stirred for 1 h, and then stirred at room temperature for a further 5 h. The precipitate was filtered off and recrystallized using aqueous ethanol.

#### 3.3. Synthesis of the Target Compound

A mixture of 1.2 mmol of 1,3-dimethylbarbituric acid, 1 mmol of chalcone derivative, 1 mmol of ammonium chloride, 100  $\mu$ L TEA, and 5 mL methanol was placed in a round bottom flask, and refluxed at 80 °C for 7 h. The reaction mixture was then cooled at room temperature, and 10 mL water was added. The precipitate was then filtered off, washed with a cold aqueous ethanol solution, and recrystallized using aqueous ethanol.

#### 4. Conclusions

In conclusion, we have successfully synthesized a new compound, that is 5-[3-(4-bromophenyl)-1-(2,5-dimethoxyphenyl)-3-oxopropyl]-1,3-dimethylpyrimidine-2,4,6(1*H*,3*H*,5*H*)-tri-one through Michael addition from a chalcone derivative.

**Supplementary Materials:** The following are available online: the HRESIMS, FTIR, <sup>1</sup>H-NMR, <sup>13</sup>C-NMR, HMQC, HMBC spectra are reported in the Supplementary Materials as Figures S1–S6, respectively, and the structure refinement in Table S1.

**Author Contributions:** H.S. brought the idea, managed the research, and wrote the manuscript. R.H.P.S. performed the synthesis, K.U.H. performed the structure elucidation, while A.N.K corrected the manuscript. All the authors have read the draft.

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**Conflicts of Interest:** The authors declare no conflict of interest.

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