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Organic Synthesis Natural Products Structure Determination

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▶ Journal Menu

- Molbank Home (/journal/molbank)
- Aims & Scope (/journal/molbank/about)
- Editorial Board (/journal/molbank/editors)
- Instructions for Authors (/journal/molbank/instructions)
- Special Issues (/journal/molbank/special_issues)
- Sections & Collections (/journal/molbank/sections)
- Article Processing Charge (/journal/molbank/apc)
- Indexing & Archiving (/journal/molbank/indexing)
- Most Cited & Viewed (/journal/molbank/most_cited)
- Journal Statistics (/journal/molbank/stats)
- Journal History (/journal/molbank/history)
- Editorial Office (/journal/molbank/editorial_office)

Journal Browser

▶ Journal Browser

volume issue Go

> Forthcoming issue (/1422-8599/2019/4)

Current issue (/1422-8599/2019/3)

Vol. 2019 (/1422-8599/2019) Vol. 2011 (/1422-8599/2011) Vol. 2018 (/1422-8599/2018) Vol. 2010 (/1422-8599/2010) Vol. 2017 (/1422-8599/2017) Vol. 2009 (/1422-8599/2009) We use cookies on our website to ensure you get the best experience. Vol. 2008 (/1422-8599/2008) Read more about our cookies here (/about/privacy). Vol. 2007 (/1422-8599/2007)

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Vol. 2012 (/1422-8599/2012) Vol. 2004 (/1422-8599/2004) Back to TopTop

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Vol. 1999 (/1422-8599/1999) Vol. 1998 (/1422-8599/1998) Vol. 1997 (/1422-8599/1997)

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Table of Contents



<u>Molbank, Volume 2017 (/1422-8599/2017)</u>, Issue 2 (June 2017)

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10a,11,11-Trimethyl-10a,11-dihydro-8H-benzo[e]imidazo[1,2-a]indol-9(10H)-one (/1422-8599/2017/2/M944)

by

- Elena Ščerbetkaitė (https://sciprofiles.com/profile/author/RTc4UGJPa3JJQ3N3QWtkK3BBRzNDZldNY2l2VVVJZIAzQVZ2WGl1MFRXND0=)
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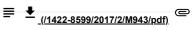
Molbank 2017, 2017(2), M944; https://doi.org/10.3390/M944 (https://doi.org/10.3390/M944) - 22 Jun 2017

Cited by 1 (/1422-8599/2017/2/M944#citedby) | Viewed by 1022

<u>Abstract</u> The alkylation of 1,1,2-trimethyl-1*H*-benzo[e]indole with 2-chloroacetamide, followed by work-up of the reaction mixture with a base and the subsequent treatment of a crude product with acetic acid gives 10a,11,11-trimethyl-10a,11-dihydro-8*H*-benzo[e]imidazo[1,2-a]indol-9(10*H*)-one. The structure [...] Read more.

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 $\underline{N-(4-Nitrophenyl)-2-\{2-[3-(4-chlorophenyl)-5-[4-(propan-2-yl)phenyl]-4,5-dihydro-1\textit{H-}pyrazol-1-yl]-4-oxo-4,5-dihydro-1,3-thiazol-5-yl\}acetamide} \\ (/1422-8599/2017/2/M943)$

by

- 🔍 Vinutha V. Salian (https://sciprofiles.com/profile/author/d2NPMkpWVWNDaVdSRXBKTkxDUDQ4UmxlTDBDcHkxOWZPR0pMMzBJTy9Ccz0=) ,
- Badiadka Narayana (https://sciprofiles.com/profile/15012) and Balladka K. Sarojini (https://sciprofiles.com/profile/361925)

Molbank 2017, 2017(2), M943; https://doi.org/10.3390/M943 (https://doi.org/10.3390/M943) - 10 Jun 2017

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Abstract

In the present investigation, the synthesis and spectroscopic characterization of *N*-(4-nitrophenyl)-2-{2-[3-(4-chlorophenyl)-5-[4-(propan-2-yl)phenyl]-4,5-dihydro-1*H*-pyrazol-1-yl]-4-oxo-4,5-dihydro-1,3-thiazol-5-yl}acetamide (**2**) is performed. The title compound (**2**) is synthesized by the reaction of 3-(4-chlorophenyl)-5-[4-(propan-2-yl)phenyl]-4,5-dihydro-1*H*-pyrazole-1-carbothioamide (**1**) with *N*-(4-nitrophenyl)maleimide. The cyclization of title compound is evidenced by FT-IR, NMR, and LCMS data. Full article (/1422-8599/2017/2/M943)

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by Sobhi M. Gomha (https://sciprofiles.com/profile/13594), Zeinab A. Muhammad (https://sciprofiles.com/profile/221530) and

Mastoura M. Edrees (https://sciprofiles.com/profile/28904)

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197 Ab 197 Ab 197 198 198
(This article belongs to the Section Organic Synthesis (/journal/molbank/sections/organic_synthesis_molbank))
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7-Bromo-1-methyl-2-phenyl-1 <i>H</i> -indole-3-carbonitrile (/1422-8599/2017/2/M941)
by Rosanna Meine (https://sciprofiles.com/profile/253006), Max Blech (https://sciprofiles.com/profile/author/OVp1eFNxTWt5cVIQcTZ0dmxWenJlbXVqcXNOc1U0SUJGZ2FQdmJkMEF0az0=), Jens Lindhof (https://sciprofiles.com/profile/author/bHBnVEF2RktvbU1hcGp5QWlsekpSNUkvVFZZQnBveGh6aHdqREs1MXVDYz0=), and Conrad Kunick (https://sciprofiles.com/profile/author/N3NQUzVSVzB4K2RkL08vOWNzMVZpMjRQbEJOOHozWXlaVHpjSEVQUk5UZz0 Molbank 2017, 2017(2), M941; https://doi.org/10.3390/M941 (https://doi.org/10.3390/M941) - 07 Apr 2017 Viewed by 1012
<u>Abstract</u>
The title compound was prepared by electrophilic aromatic substitution of 7-bromo-1-methyl-2-phenyl-1 <i>H</i> -indole with NCTS (<i>N</i> -cyano- <i>N</i> -phenyl- <i>p</i> -toluenesulfonamide). The structural identity of the title compound was proven by elemental analysis and spectroscopic methods (IR, NMR, APCI-MS). Purity was assessed by two independent HPLC methods. <u>Full article (/1422-8599/2017/2/M941)</u> (This article belongs to the Section <u>Organic Synthesis (/journal/molbank/sections/organic_synthesis_molbank)</u>)
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Laurent Guyard (https://sciprofiles.com/profile/author/RXQ1ZU9hNmRGZ1E1RnJsL01SSmUyMIFIT3V2bWNScVRkZStGTzV2RFFQTT0=) Molbank 2017, 2017(2), M940; https://doi.org/10.3390/M940 (https://doi.org/10.3390/M940) - 07 Apr 2017 Viewed by 922 Abstract A new member of the 2-acetylpyridine family has been prepared and characterized. Its synthesis is a two-step process starting from a pyridylalcohol in which the ketone moiety is protected as a cyclic acetal. Alkylation of the alcohol followed by hydrolysis of the acetal [] Read more. (This article belongs to the Section Organic Synthesis (/journal/molbank/sections/organic_synthesis_molbank)) ► Show Figures
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4-Methoxy-3-(3-methylbut-2-en-1-yl)-7-[(3-methylbut-2-en-1-yl)oxy]quinolin-2(1H)-one from Melicope Moluccana T.G. Hartley (/1422-8599/2017/2/M939)
by Mulyadi Tanjung (https://sciprofiles.com/profile/190848), Ratih Dewi Saputri (https://sciprofiles.com/profile/257183), Ryan Ayub Wahjoedi (https://sciprofiles.com/profile/author/QU5kc0xWWUJ2LzE0M2N3eUtVdVptd29LSUw4dXMzbGo4M25LVmdBVGRQTT0=) and
☑ Tjitjik Srie Tjahjandarie (https://sciprofiles.com/profile/159161) Molbank 2017, 2017(2), M939; https://doi.org/10.3390/M939 (https://doi.org/10.3390/M939) - 07 Apr 2017 Cited by 4 (/1422-8599/2017/2/M939#citedby) Viewed by 922
Abstract 4-Methoxy-3-(3-methylbut-2-en-1-yl)-7-[(3-methylbut-2-en-1-yl)oxy]quinolin-2(1 <i>H</i>)-one (1) was isolated from the leaves of <i>Melicope moluccana</i> T.G. Hartley. The chemical structure of 1 was elucidated using mainly UV, IR, HRESIMS, 1D and 2D-NMR spectroscopy. Full article (/1422-8599/2017/2/M939). (This article belongs to the Section Natural Products (/journal/molbank/sections/natural products molbank))
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4-({4-[(2E)-3-(2,5-Dimethoxyphenyl)prop-2-enoyl]phenyl}amino)-4-oxobutanoic Acid (/1422-8599/2017/2/M938)
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and Ni Nyoman Tri Puspaningsih (https://sciprofiles.com/profile/873433)
Molbank 2017, 2017(2), M938; https://doi.org/10.3390/M938 (https://doi.org/10.3390/M938) - 05 Apr 2017 Cited by 3 (/1422-8599/2017/2/M938#citedby) Viewed by 956
Abstract A dimethoxy amide chalcone has been synthesized in a two-step reaction. First, an amine chalcone was synthesized by the reaction of 4'- we use cookies on our website to ensure you get the best experience. aminoacetophenone and 2.5-dimethoxybenzaldehyde using 40% NaOH solution as a catalyst in ethanol, and then followed by amidation through the reaction of the [] Read more.
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by \bigcirc Nicolas Bélanger-Desmarais (https://sciprofiles.com/profile/253563) and \bigcirc Christian Reber (https://sciprofiles.com/profile/168634) \bigcirc \bigcirc Molbank 2017, 2017(2), M937; https://doi.org/10.3390/M937 (https://doi.org/10.3390/M937) - 30 Mar 2017

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<u>Abstract</u> Abstract: A gold(I) complex with a triphenylphosphine and a monodentate *N*,*N*-dimethyldithiocarbamate ligand was synthesized and characterized by Raman spectroscopy and single crystal X-ray diffraction. DFT calculations (Gaussian 09, PBE1PBE/Lanl2dz) were undertaken for a single complex in the gas phase. The [...] <u>Read more.</u>

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isplaying articles 1-8 Previous Issue

Volume 2017, March (/1422-8599/2017/1)

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Volume 2017, September (/1422-8599/2017/3)

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- Aims & Scope (/journal/molbank/about)
- Editorial Board (/journal/molbank/editors)
- Instructions for Authors (/journal/molbank/instructions)
- Special Issues (/journal/molbank/special_issues)
- Sections & Collections (/journal/molbank/sections)
- Article Processing Charge (/journal/molbank/apc)
- Indexing & Archiving (/journal/molbank/indexing)
- Most Cited & Viewed (/journal/molbank/most_cited)
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- Journal History (/journal/molbank/history)
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	<u>(/1422-8599/2018)</u>
Vol. 2017	<u>(/1422-8599/2017</u>)
Vol. 2016	(/1422-8599/2016)
Vol. 2015	(/1422-8599/2015)
Vol. 2014	(/1422-8599/2014)
Vol. 2013	(/1422-8599/2013)
Vol. 2012	(/1422-8599/2012)
Vol. 2011	<u>(/1422-8599/2011)</u>
Vol. 2010	(/1422-8599/2010)
Vol. 2009	(/1422-8599/2009)
Vol. 2008	(/1422-8599/2008)

Vol. 2007 (/1422-8599/2007)
Vol. 2006 (/1422-8599/2006)
Vol. 2005 (/1422-8599/2005)
Vol. 2004 (/1422-8599/2004)
Vol. 2003 (/1422-8599/2003)
Vol. 2002 (/1422-8599/2002)
Vol. 2001 (/1422-8599/2001)
Vol. 2000 (/1422-8599/2000)
Vol. 1999 (/1422-8599/1999)
Vol. 1998 (/1422-8599/1998)
Vol. 1997 (/1422-8599/1997)







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Interests: organic synthesis, methodology and spectroscopic analysis; heterocyclic chemistry; natural products; asymmetric synthesis; nitrogenated compounds of pharmaceutical interest; multicomponent reactions



Dr. R. Alan Aitken

E-Mail () Website (http://ch-www.st-and.ac.uk/staff/raa/group)

School of Chemistry, University of St Andrews North Haugh, St Andrews Fife, KY16 9ST, UK

Tel. (01334) 463865; Fax: (01334) 463865

 $\textbf{Interests:} \ \ \text{synthesis; synthetic use of flash vacuum pyrolysis; heterocyclic chemistry; reactive intermediates; organophosphorus; \\$

organosulfur; heavier main group chemistry



Prof. Dr. Fawaz Aldabbagh

E-Mail () Website (https://www.kingston.ac.uk/staff/profile/professor-fawaz-aldabbagh-462/)

Department of Pharmacy, School of Life Sciences, Pharmacy & Chemistry, Kingston University, Penrhyn Road, Kingston upon Thames, KT1 2EE, UK

Interests: free radical organic and polymer chemistry; heterocyclic and medicinal chemistry

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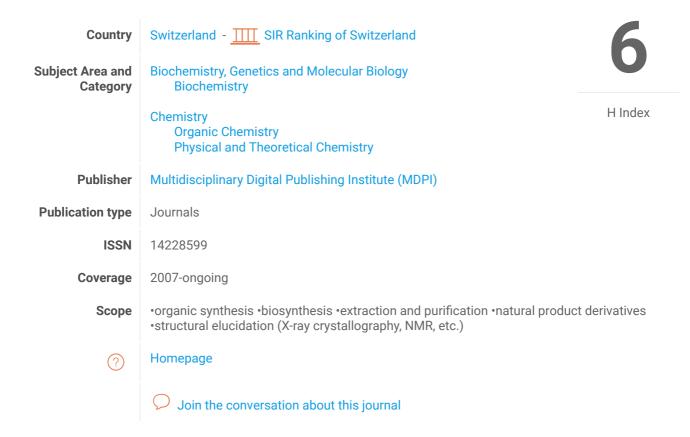
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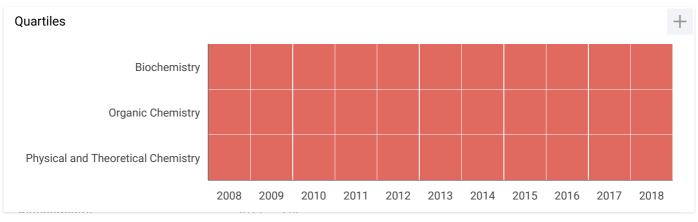
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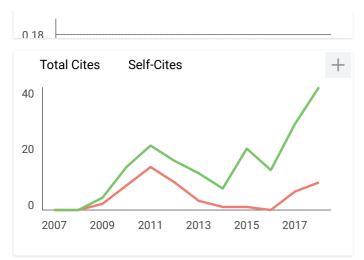
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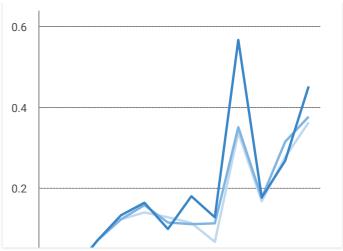
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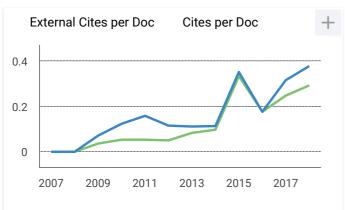


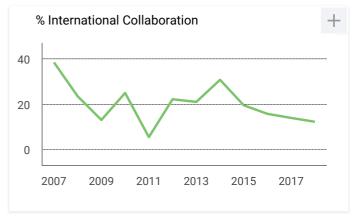


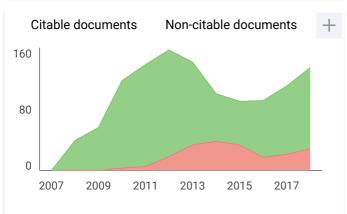


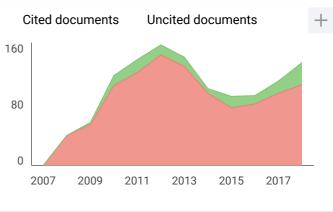


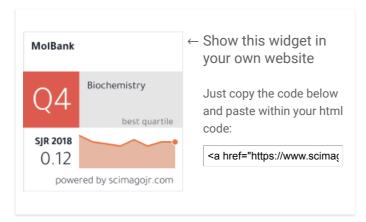












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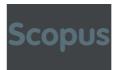
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Abstract: A dimethoxy amide chalcone has been synthesized in a two-step reaction. First, an amine chalcone was synthesized by the reaction of 4′-aminoacetophenone and 2,5-dimethoxybenzaldehyde using 40% NaOH solution as a catalyst in ethanol, and then followed by amidation through the reaction of the formed chalcone and succinic anhydride. The structure of the target compound was established by FTIR, HR-MS, ¹H- and ¹³C-NMR, and 2D-NMR spectral analysis.

Keywords: amino chalcones; amide chalcones; succinic anhydride

1. Introduction

Diaryl- α , β -unsaturated ketones are the biogenic precursor in flavonoid biosynthesis [1] and are known as chalcones. Due to their wide spectrum of pharmacological properties, such as antioxidant [2], antihepatotoxic [3], neuroprotective [4], antibacterial [5–8], inhibitor of topoisomerase I [9], antimalarial [10,11], and anticancer [12], chalcones attract many researchers to develop efficient synthetic methods and to gain various structural variations of chalcones unavailable in nature. In general, chalcones are synthesized by Claisen-Schmidt condensation.

Previously we have reported the antimicrobial activity of a series of methoxy amino chalcones [13,14]. In order to enhance their efficacy by increasing their solubility and slow release, we converted the basic amino chalcones into amide derivatives through a reaction with succinic anhydride. Herein we report a new amide methoxy chalcone prepared from a methoxy amino chalcone and succinic anhydride.

2. Results

The title compound was synthesized in a two-step reaction. The first step was the synthesis of a methoxy amino chalcone (1) employing the Claisen-Schmidt reaction, then followed by the amidation of (1) through the reaction of (1) with succinic anhydride in ethanol using pyridine as a catalyst, as shown in Scheme 1.

Molbank **2017**, 2017, M938 2 of 4

Scheme 1. Synthesis pathway of the target compound.

Firstly, the purity of the product was analyzed by determining its melting point and thin layer chromatography. The structure of the product was then characterized based on spectroscopic evidence and the results are displayed below. The product is assumed to exist in the *E* configuration, since the ¹H-NMR spectrum of the olefinic protons showed a coupling constant of 15.7 Hz indicative of the *E* configuration. The structure of the title compound and its HMBC correlations is displayed in Figure 1, whereas the chemical shifts and its HMBC correlations is tabulated in Table 1. The complete spectra are attached in supplementary materials.

(*E*)-4-((3-(2,5-Dimethoxyphenyl)acryloyl)phenyl)amino)-4-oxobutanoic acid (**2**): orange solid (266 mg; 75%), m.p. 188–190 °C, R_f = 0.61 (ethanol), HR-MS [M + H]⁺ calculated for $C_{21}H_{21}NO_6$ 384.1447, found 384.1446; IR (KBr, cm⁻¹) 3448 (br, -OH carboxylic), 3340 (str, -NH- amide), 1720 (str, C=O aliphatic carboxylic acid), 1697 (str, C=O amide), 1639 (C=O conjugated), 1593 (str, C=C conjugated), and 1261 (C_{alkyl} -O- C_{aryl}); ¹H-NMR (400 MHz, DMSO- d_6) δ (ppm) 12.15 (s, 1H), 10.32 (s, 1H), 8.10 (d, J = 8.7 Hz, 2H), 7.97 (d, J = 15.7 Hz, 1H), 7.86 (d, J = 15.7 Hz, 1H), 7.73 (d, J = 8.7 Hz, 2H), 7.51 (d, J = 2.3 Hz, 1H), 7.08–6.92 (m, 2H), 3.80 (s, 3H), 3.76 (s, 3H), 2.58 (t, J = 6.2 Hz, 2H), 2.50 (t, J = 6.0 Hz, 2H). ¹³C-NMR (101 MHz, DMSO- d_6) δ (ppm) 188.05, 174.33, 171.35, 153.77, 153.16, 144.19, 138.06, 132.64, 130.46, 124.08, 122.36, 118.73, 118.57, 113.53, 112.93, 56.66, 56.22, 31.71, 29.13.

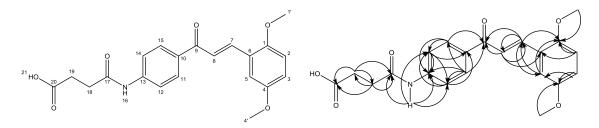


Figure 1. Structure, numbering and HMBC correlations of the title compound.

The analysis of the correlation spectrum (2D NMR; HMBC) is tabulated in the Table 1 below.

Table 1. ¹H, ¹³C chemical shifts and HMBC correlations of the title compound.

Atom	Chemical Shift (ppm)	НМВС
1 C	153.16	
1′ C	56.66	
H ₃ 2 C	3.80	1
2 C	113.53	
Н	6.99	1,6

Molbank **2017**, 2017, M938 3 of 4

Table 1. Cont.

Atom	Chemical Shift (ppm)	НМВС
3 C	118.57	
H	6.99	4,5
4 C	153.77	
4′ C	56.22	
H_3	3.76	4
5 C	112.93	
H	7.51	7, 3, 1
6 C	124.08	
7 C	138.06	
Н	7.97	9, 5, 1, 8
8 C	122.36	
H	7.86	9, 7, 6
9 C	188.05	
10 C	132.64	
11 C	130.46	
Н	8.10	9, 12, 15, 13
12 C	118.73	
Н	7.73	14, 10, 13
13 C	144.19	
14 C	118.73	
Н	7.73	12, 10, 13
15 C	130.46	
Н	8.10	9, 14, 11, 13
16 H	10.32	12, 14, 13, 17
17 C	171.35	
18 C	29.13	
H_2	2.50	19, 17, 20
19 C	31.71	
H_2	2.58	18, 17, 20
20 C	174.33	
21 H	12.15	

3. Materials and Methods

3.1. General

All reagents and solvents (E.Merck (Darmstadt, Germany) or Sigma Aldrich (St. Louis, MO, USA)) were used without further purification. Reaction progress was monitored by TLC on silica gel GF_{254} aluminum sheets (0.25 mm) using various developing system. Spots were detected under UV light (λ 254 nm). Melting point was measured by Thermo Scientific Fisher-Johns Melting Point Apparatus 220 VAC (Waltham, MA, USA) and uncorrected. FTIR spectrum was recorded in KBr pellet on FTIR spectrophotometer Shimadzu 84005 series (Kyoto, Japan). Mass spectrum was recorded on HR mass spectrometer Waters LCT Premier XE (Santa Clara, CA, USA). NMR spectrum (1 H-, 1 C-NMR, and HMBC) was recorded using JEOL 400 ECA spectrometer (Tokyo, Japan) with DMSO- d_6 as solvent and internal standard.

3.2. Preparation of the Title Compound (2)

The amino methoxy chalcone (1) was synthesized according to the protocol as described previously [2]. The title compound was synthesized as followed: 1 mmol succinic anhydride was dissolved in 5 mL DCM, then three drops of pyridine was added. The mixture was stirred at $40\,^{\circ}\text{C}$ for $10\,\text{min}$. Then 1 mmol of chalcone (1) in 2 mL DCM was added drop-wise, stirred overnight at room temperature. The precipitate was then filtered off and re-crystallized from ethanol.

4. Conclusions

We have demonstrated the synthesis of a methoxy amide chalcone derivative through the Claisen-Schmidt reaction, followed by amidation.

Molbank **2017**, 2017, M938 4 of 4

Supplementary Materials: FTIR, HRMS, ¹H-NMR, ¹³C-NMR, HMBC spectra of the synthesized compound are available online.

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Author Contributions: H.S. brought out the idea, managed the research and wrote the paper. K.U.H. and A.N.K. analyzed the spectral data. N.N.D.R. performed the synthesis, while N.N.T.P. corrected the draft. All the authors have read the draft.

Conflicts of Interest: The authors declare no conflict of interest.

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