

Tetrahedron Letters

COUNTRY	SUBJECT AREA AND CATEGORY	PUBLISHER	H-INDEX
United Kingdom Universities and research institutions in United Kingdom	Biochemistry, Genetics and Molecular Biology Biochemistry Chemistry Organic Chemistry Pharmacology, Toxicology and Pharmaceutics Drug Discovery	Elsevier Ltd.	165
PUBLICATION TYPE	ISSN	COVERAGE	INFORMATION
Journals	00404039, 18733581	1959-2020	Homepage
			How to publish in this journal
			timothy.donohoe@chem.ox.ac.uk

SCOPE

Tetrahedron Letters provides rapid dissemination of short accounts of advances of outstanding significance and timeliness in the broad field of organic chemistry and its related disciplines, such as organic materials and bio-organic chemistry. Communications in Tetrahedron Letters are expected to represent brief summaries of preliminary work or initial results at the cutting edge of the field. Rapid publication of such research enables authors to transmit their new contributions quickly to a large, international audience. Tetrahedron Letters also publishes 'Digests', commissioned short reviews, highlights or perspectives, focusing on recent advancements in a field.

 $\ensuremath{\bigcirc}$ Join the conversation about this journal





TETRAHEDRON LETTERS

EXECUTIVE BOARD OF EDITORS FOR TETRAHEDRON PUBLICATIONS

Chairman: **Professor S. F. Martin** Editor Emeritus: **Professor H. H. Wasserman**

Peter R. Bernstein, PhaRmaB LLC, 14 Forest View Rd., Rose Valley, PA, USA

Professor D. L. Boger, The Scripps Research Institute, La Jolla, CA, USA **Professor S. Chandrasekaran,** Department of Organic Chemistry, Indian Institute of Science, Bangalore, India

Professor M. Christmann, Dortmund University of Technology, Dortmund, Germany

Professor S. G. Davies, University of Oxford, Oxford, UK

Professor L. Ghosez, Institut Européen de Chimie et de Biologie (IECB), France

Professor L. Guo-Qiang, Shanghai Institute of Organic Chemistry, Chinese Academy of Sciences, Shanghai, China

Professor Y. Hashimoto, University of Tokyo, Tokyo, Japan

Professor T. Hayashi, Kyoto University, Kyoto, Japan

Dr. J. Hu, Chinese Academy of Sciences (CAS), Shanghai, China **Professor K. D. Janda,** The Scripps Research Institute, La Jolla, CA, USA

Professor M. Katimura, Nagoya University, Nagoya, Japan Professor K. Maruoka, Department of Chemistry, Graduate School of

Science, Kyoto University, Kyoto, Japan

Professor S. Neidle, UCL School of Pharmacy, London, UK

Professor M. Shibasaki, The University of Tokyo, Japan

Professor R. J. K. Taylor, University of York, UK

Professor E. J. Thomas, School of Chemistry, The University of Manchester, Manchester, UK (Associate Editor, Professor J. A. Joule)

Professor K. Tomioka, Doshisha Women's College of Liberal Arts, Kyotanabe, Japan

Professor H. Waldmann, Max-Planck-Institut für Molekulare Physiologie, Dortmund, Germany

Professor H. H. Wasserman, Yale University, New Haven, CT, USA

Professor R. M. Williams, Colorado State University, Fort Collins, CO, USA

Professor C.-H. Wong, The Scripps Research Institute, La Jolla, CA, USA

Professor J. Wood, Department of Chemistry and Biochemistry, Baylor University, TX, USA

Professor S. Z. Zard, Laboratoire de Synthèse Organique, Ecole Polytechnic, Palaiseau Cedex, France (Associate Editor, Dr. B. Sire)

Editors of the Tetrahedron Letters Digest Papers: Professor Dr. M. Christmann, Freie Universität Berlin, Berlin, Germany Dr. J. Hu, Chinese Academy of Sciences (CAS), Shanghai, China Professor M. Kitamura, Nagoya University, Nagoya, Japan Professor B. M. Stoltz, California Institute of Technology, Pasadena, CA, USA

Editors of the Tetrahedron Organic Chemistry Series: Professor J.-E. Bäckvall, University of Stockholm, Sweden Professor Sir J. E. Baldwin, FRS, Dyson Perrins Laboratory, Oxford, UK Professor R. M. Williams, Colorado State University, Fort Collins, CO, USA

BOARD OF CONSULTING EDITORS

	P. Knochel, Ludwigs-Maximilians-University,	K. C. Nicolaou, Rice University, Houston, TX, USA
n	Munich, Germany	R. Noyori, Nagoya University, Japan
nberra,	P. Kocienski, University of Leeds, UK	L. E. Overman, University of California, Irvine,
	M. Krische, University of Texas, USA	CA, USA
	E. Lee, Seoul National University, Seoul, Korea	A. Padwa, Emory University, USA
	S. V. Ley, University of Cambridge, UK	I. Paterson, University of Cambridge, UK
	XY. Lu, Shanghai Institute of Organic Chemistry,	G. Pattenden, University of Nottingham, UK
	China	S. Schreiber, Harvard University, USA
	D. Ma, State Key Laboratory of Bioorganic & Natural	T. Shioiri, Mejio University, Japan
, USA	Products Chemistry, Shanghai Institute of Organic	N. Simpkins, Birmingham, UK
	Chemistry, China	E. Sorensen, Princeton University, Princeton, NJ,
	D. MacMillan, Princeton University, Princeton, NJ,	USA
rk, NY,	USA	B. M. Stoltz, California Institute of Technology, USA
	I. Marko, University of Louvain, Belgium	K. Tatsuta, Waseda, Japan
	G. Mehta, Indian Institute of Science, Bangalore, India	J. D. Wuest, University of Montreal, Canada
ool, UK	S. J. Miller, Yale, USA	Z. Xi, College of Chemistry, Peking University (PKU),
ı	N. Miyata, Nagoya City University, Japan	China
	J. Moore, University of Illinois at Urbana-Champaign,	H. Yamamoto, University of Chicago, IL, USA
ISA	IL, USA	
	K. Narasaka, Nanyang Technical University,	
	Singapore	

- A. Alexakis, University of Geneva, Switzerland
- J.-E. Bäckvall, University of Stockholm, Sweden M. Banwell, Australian National University, Canberr
- Australia
- A. G. M. Barrett, Imperial College, London, UK
- J. Bode, ETH Zurich, Zurich, Switzerland
- C. Bolm, RWTH Aachen, Germany
- S. Buchwald, MIT, Cambridge, MA, USA
- E. M. Carreira, ETH, Zürich, Switzerland
- E. J. Corey, Harvard University, Cambridge, MA, U
- I. Cossy, ESPCI, Paris, France
- D. P. Curran, University of Pittsburgh, PA, USA

S. J. Danishefsky, Columbia University, New York, NY

- USA
- S. Denmark, University of Illinois, USA
- P. A. Evans, The University of Liverpool, Liverpool, UK
- J.-M. Fang, National Taiwan University, Taiwan
- **G. C. Fu,** MIT, Cambridge, MA, USA
- Y. Kishi, Harvard University, Cambridge, MA, USA
- Y. Kita, Osaka University, Japan
- D. W. Knight, Cardiff, UK

PUBLISHED WEEKLY

Orders, claims, and journal inquiries: please contact the Elsevier Customer Service Department nearest you: **St. Louis:** Elsevier Customer Service Department, 3251 Riverport Lane, Maryland Heights, MO 63043, USA; phone: (877) 8397126 [toll free within the USA]; (+1) (314) 4478878 [outside the USA]; fax: (+1) (314) 4478077; e-mail: JournalCustomerService-usa@elsevier.com; **Oxford**: Elsevier Customer Service Department, The Boulevard, Langford Lane, Kidlington, Oxford OX5 1GB, UK; phone: (+44) (1865) 843434; fax: (+44) (1865) 843970; e-mail: JournalScustomerServiceEMEA@elsevier.com; **Tokyo**: Elsevier Customer Service Department, 4F Higashi-Azabu, 1-Chome Bldg., 1-9-15 Higashi-Azabu, Minato-ku, Tokyo 106-0044, Japan; phone: (+81) (3) 5561 5037; fax: (+81) (3) 5561 5047; e-mail: JournalScustomerServiceJapan@elsevier.com; **Singapore**: Elsevier Customer Service Department, 3 Killiney Road, #08-01 Winsland House I, Singapore 239519; phone: (+65) 63490222; fax: (+65) 67331510; e-mail: JournalScustomerServiceAPAC@elsevier.com



Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Tetrahedron Letters Vol. 55, Issue 7, 2014







COMMUNICATIONS

An expeditious approach to access 2-arylimidazo[1,2-a]pyridin-3-ol from 2-amino pyridine through a novel Petasis pp 1281-1284 based cascade reaction

Yuanxiang Wang, Biswajit Saha, Fang Li, Brendan Frett, Hong-yu Li*



R² = MeO, Me, CF₃O, H, Cl, Br, F etc (10 examples)

Further prenylated flavonols from Platanus acerifolia's unripe buds Mourad Kaouadji*

H₂(H₂C ĊΗ 5 R = H3 R = H4 $R = OCH_3$ $6 R = OCH_3$

The toluene extract of defatted Platanus acerifolia's fresh unripe buds afforded 8 metabolites. In all, prenylated flavonols 3-6 are reported for the first time in the plant kingdom.

pp 1285-1288



anti-Selective enolboration-aldolization of propanoic acid

P. Veeraraghavan Ramachandran*, Prem B. Chanda, Barnabas Otoo



Preparation of 2,2-difluoro-1-Jong Hee Jeon, Ju Hee Kim, Yeo

Microwave assisted synthesis of 2-aminooxazolo [4,5-b] pyridine derivatives via intramolecular C–O bond formation in pp 1296–1298 aqueous medium

 $\begin{array}{c} R \xrightarrow{\quad \text{Br} \\ Pd(PPh_3)_4/Cul} \end{array} (X = H, F, Cl) \\ THF, reflux, 3-5 h \end{array}$

(X = H, F, Cl, Br, CH₃, OCH₃, CF

(R = aryl, alkyl, trialkylsilyl)

NO₂)

Umesh B. Kosurkar, Tulshiram L. Dadmal, K. Appalanaidu, Y. Khageswara Rao, Jagadeesh B. Nanubolu, Ravindra M. Kumbhare*



Mitsuhiro Okimoto*, Haruki Yamamori, Masayuki Hoshi, Takashi Yoshida

$$\begin{array}{c} -2e, -2H^{+} \\ \text{CO}_{2}Me \\ \text{CO}_{2$$

-trialkylsilylethenylstannanes and their cross-coupling reactions
of Jin Jeong, In Howa Jeong*
$$\sum_{F} \underbrace{OTs}_{2} SiR_{3} + \underbrace{(Bu_{3}Sn)_{2}}_{(1.0 \text{ equiv})} \underbrace{\frac{Pd_{2}(dba)_{3}/XPhos}{UiBr, THF, reflux, 8 h}}_{73-74\%} \xrightarrow{F} \underbrace{SnBu_{3}}_{3} \underbrace{\frac{X}{D}I/Pd(PPh_{3})_{4}/Cul}_{DMF, 80 \, ^{\circ}C, 10-20 h}}_{47-90\%} \xrightarrow{F}_{F}$$



pp 1289-1291





pp 1299-1302

Chemoenzymatic total synthesis of paecilocin A and 3-butyl-7-hydroxyphthalide

Ch. Sreelakshmi, A. Bhaskar Rao*, M. Laxminarsu, P. Janardhan Reddy, B. V. Subba Reddy*



Bifunctional AgOAc/ThioClickFerrophos complex-catalyzed asymmetric 1,3-dipolar cycloaddition of azomethine ylides pp 1306–1309 with aryl- and alkylidene malonates

Sayo Watanabe, Atsuo Tada, Yuichiro Tokoro, Shin-ichi Fukuzawa*



Base-catalyzed stereoselective intermolecular addition of imidazoles onto alkynes: an easy access to imidazolyl pp 1310–1315 enamines

Monika Patel, Rakesh K. Saunthawal, Akhilesh K. Verma*



An efficient transition-metal free approach for the regio- and stereoselective addition of imidazoles **1a–f** onto alkynes **2a–l** to provide the *Z*- and *E* isomers of imidazolyl enamines **3a–q** and **4a–d** using catalytic amount of KOH is described. Stereoselectivity of the addition products (*Z* and *E* isomer) was found to be dependent upon time. Competitive experiments show that imidazole is less reactive than pyrrole and more reactive than aniline toward hydroamination.

Fe³⁺-exchanged clay catalyzed transamidation of amides with amines under solvent-free condition Md. Ayub Ali, S. M. A. Hakim Siddiki, Kenichi Kon, Ken-ichi Shimizu*

Aromatic, Heteroaromatic and Aliphatic Amides



pp 1316-1319

pp 1303-1305

1265



Microwave assisted synthesis of nitro phenols from the reaction of phenols with urea nitrate under acid-free conditions pp 1320–1322 Sanny Verma, Sangeeta Pandita, Suman L. Jain*





The efficient one-pot synthesis of tetraalkyl substituted furans from symmetrical acetylenes, EtAlCl₂, and carboxylic pp 1326–1328 esters catalyzed by Cp₂TiCl₂

Mariya G. Shaibakova, Leila O. Khafizova*, Nuri M. Chobanov, Rinat R. Gubaidullin, Natal'ya R. Popod'ko, Usein M. Dzhemilev

$$R = R + EtAlCl_2 + R'CO_2R'' \qquad \begin{array}{c} Cp_2TiCl_2, Mg \\ \hline THF, 60 \ ^\circ C, 6 \ h \end{array} \xrightarrow{R'} \begin{array}{c} R \\ \hline 0 \\ \hline 0 \\ \hline R \\ R, R', R'' = alkyl \end{array}$$

Facile access to a benzoazepinoquinazolinone via a free radical cyclization Khaled Q. Shawakfeh, Zakariyya N. Ishtaiwi, Naim H. Al-Said* pp 1329-1331



A new access to 2-phosphonothiophenes Dariusz Cal*



pp 1332-1335

Benzopyranone, benzophenone, and xanthone derivatives from the soil fungus *Penicillium citrinum* **PSU-RSPG95** pp 13 Kongkiat Trisuwan*, Vatcharin Rukachaisirikul, Kawitsara Borwornwiriyapan, Souwalak Phongpaichit, Jariya Sakayaroj

3

MeO

4

ΟН

pp 1336-1338



pp 1339-1341

Total synthesis of (-)-Englerin A

Jinghua Zhang, Shuyan Zheng, Wei Peng, Zhengwu Shen*

Me

1: R = OMe

2: R = OEt



Anthracene-labeled pyridinium-based symmetrical chiral chemosensor for enantioselective recognition of L-tartrate pp 1342–1346 Kumaresh Ghosh*, Tanmay Sarkar



A new anthracene-based chiral chemosensor 1 has been designed and synthesized. L-Valine has been used as the chiral source in the design. The chemosensor 1 has been established as an efficient enantioselective sensor for L-tartrate over D-tartrate. The enantiomeric fluorescence difference ratio (*ef*) has been determined to be 29.38.





A fluorescent and colorimetric chemosensor for selective detection of aluminum in aqueous solution

Kyung Beom Kim, Dong Min You, Jun Hwi Jeon, Yo Han Yeon, Jong Ha Kim, Cheal Kim*



Synthesis in ionic liquids only: access to α -oxo- γ -thio-esters via Mukaiyama coupling

Khouloud Jebri, Marie-Rose Mazières, Stéphanie Ballereau, Taïcir Ben Ayed, Jean-Christophe Plaquevent, Michel Baltas*, Frédéric Guillen*



Design and synthesis of europium luminescent bio-probes featuring sulfobetaine moieties

Virginie Placide, Delphine Pitrat, Alexei Grichine, Alain Duperray, Chantal Andraud, Olivier Maury*



Two novel tetracycles, cassibiphenols A and B from the flowers of Cassia siamea Jun Deguchi, Tadahiro Sasaki, Yusuke Hirasawa, Toshio Kaneda, Idha Kusumawati, Osamu Shirota, Hiroshi Morita*

> Me Me Ó Ó ÍНО NO[®] HO OH. Мe Мe Ńе Ńе cassibiphenol A (1) cassibiphenol B (2)

Chemical investigation of the flowers of Cassia siamea (Leguminosae), resulted in the isolation of two novel tetracycles connecting 5-(2hydroxypropyl)benzene-1,3-diol, cassibiphenols A (1) and B (2). The structures were elucidated by analysis of the 1D, 2D NMR, and HRMS spectra. Synthesis of a tetracyclic core of 1 and 2 led to determine the absolute configuration of 1 and C-12 of 2.

1268



pp 1353-1356



pp 1362-1365

Synthesis of novel 5-oxaprotoberberines as bioisosteres of protoberberines

Yifeng Jin, Daulat Bikram Khadka, Su Hui Yang, Chao Zhao, Won-Jea Cho^*



One-pot three-component reaction for the synthesis of biologically important spiro[benzo[f]quinoline-3,3'-indoline] pp 1370–1372 derivatives

CO2R2

CO_R² argon atm.

=o _

R²O₂C

SbCl₃ (10 mol %) CH₃CN, 80° C,

Rajiv Karmakar, Utpal Kayal, Biswajit Bhattacharya, Gourhari Maiti*



Zheng-Wang Chen*, Dong-Nai Ye, Min Ye, Zhong-Gao Zhou, Shen-Huan Li, Liang-Xian Liu*

$$R \longrightarrow X \xrightarrow{\text{AgF (5 mol \%)}} R \xrightarrow{O} X$$

R = aryl, X = Cl, Br up to 95% yield

A AgF/TFA-promoted highly efficient synthesis of a wide range of α -haloketones from haloalkynes is described. The reactions are conducted under convenient conditions and provide products in moderate to excellent yields, with broad substrate scope, including a variety of aromatic chloroalkynes and bromoalkynes.

Development of a practical and sustainable strategy for the synthesis of ST1535 by an iron-catalyzed Kumada cross-coupling reaction

Francesca Bartoccini, Giovanni Piersanti*, Silvia Armaroli, Alberto Cerri, Walter Cabri*



pp 1376-1378

1269





A novel one-pot synthesis of oxazolidinones through direct introduction of CO₂ into allylamine derivatives Laura Soldi, Chiara Massera, Mirco Costa, Nicola Della Ca'*

pp 1379-1383

pp 1384-1386



A facile synthesis of the novel thiazolo[3,2-*a*]pyrimidine derivatives

Renata Studzińska*, Marcin Wróblewski, Aleksandra Karczmarska-Wódzka, Renata Kołodziejska



*Corresponding author

()+ Supplementary data available via ScienceDirect

Abstracted/indexed in: AGRICOLA, Beilstein, BIOSIS Previews, CAB Abstracts, Chemical Abstracts, Chemical Engineering and Biotechnology Abstracts, Current Biotechnology Abstracts, Current Contents: Life Sciences, Current Contents: Physical, Chemical and Earth Sciences, Current Contents Search, Derwent Drug File, Ei Compendex, EMBASE/Excerpta Medica, Medline, PASCAL, Research Alert, Science Citation Index, SciSearch. Also covered in the abstract and citation database Scopus[®]. Full text available on ScienceDirect[®]



Available online at www.sciencedirect.com



Tetrahedron Letters 55 (2014) 1362-1365

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

Two novel tetracycles, cassibiphenols A and B from the flowers of Cassia siamea

Jun Deguchi^a, Tadahiro Sasaki^a, Yusuke Hirasawa^a, Toshio Kaneda^a, Idha Kusumawati^b. Osamu Shirota^c. Hiroshi Morita^{a,*}

^a Faculty of Pharmaceutical Sciences, Hoshi University, Ebara 2-4-41, Shinagawa-ku, Tokyo 142-8501, Japan

^b Faculty of Pharmacy, Airlangga University, Jalan Dharmawangsa Dalam, Surabaya 60286, Indonesia

^c Faculty of Pharmaceutical Sciences at Kagawa Campus, Tokushima Bunri University, 1314-1 Shido, Sanuki City, Kagawa 769-2193, Japan

ARTICLE INFO

Article history: Received 25 November 2013 Revised 26 December 2013 Accepted 7 January 2014 Available online 13 January 2014

Keywords: Cassia siamea Cassibiphenols A and B Biphenvl Structure elucidation Partial synthesis

In our screening study on new antiplasmodial agents from plant resources, we have succeeded in the isolation of new various alkaloids,¹ sesquiterpenes,² and limonoids.³ Cassiarin A,⁴ an unprecedented tricyclic alkaloid exhibiting potent antimalarial activity against Plasmodium falciparum in vitro as well as Plasmo*dium berghei* in vivo,⁵ was isolated from the leaves of *Cassia siamea* and has attracted attention of synthetic organic chemists⁶ as well as pharmacologists.⁷ Cassia siamea Lam. (Leguminosae), has been used widely in traditional medicine, particularly for treatment of periodic fever and malaria in Indonesia.⁸ Recently we isolated cassiarins G, H, J, and K⁹, showing antiplasmodial activity from leaves of C. siamea, and achieved total synthesis of a novel tetracyclic alkaloid, cassiarin F isolated from flowers of C. siamea.¹⁰ Further isolation work on extracts from the flowers of C. siamea has led to purification and structure elucidation of two novel tetracyclic cassibiphenols A (1) and B (2). In this Letter, we would like to report the structure elucidation based on spectroscopic analyses and partial synthesis of 1 and 2. The absolute configuration at C-12 was established by the comparison of the CD spectra of tetracyclic core in 1 and 2.

The flowers of C. siamea (1.0 kg) were extracted with MeOH, and the extract was partitioned between EtOAc and 3% aqueous tartaric acid. The aqueous layer was adjusted at pH 9 with saturated Na₂₋ CO3 aq and extracted with CHCl3. CHCl3-soluble materials were

E-mail address: moritah@hoshi.ac.jp (H. Morita).

ABSTRACT

Chemical investigation of the flowers of Cassia siamea (Leguminosae), resulted in the isolation of two novel tetracycles connecting 5-(2-hydroxypropyl)benzene-1,3-diol, cassibiphenols A (1) and B (2). The structures were elucidated by analysis of the 1D, 2D NMR, and HRMS spectra. Synthesis of a tetracyclic core of **1** and **2** led to determine the absolute configuration of **1** and C-12 of **2**.

cassibiphenol A (1)

© 2014 Elsevier Ltd. All rights reserved.

Ńe

cassibiphenol B (2)

subjected to a silica gel column, an ODS column, an LH-20 column, and ODS HPLC to give cassibiphenols A (1, 0.00002%)¹¹ and B (2, 0.00002%).¹²









^{*} Corresponding author. Tel./fax: +81 354985778.

^{0040-4039/\$ -} see front matter © 2014 Elsevier Ltd. All rights reserved. http://dx.doi.org/10.1016/j.tetlet.2014.01.023

Table 1¹H NMR data [δ_{H} (J, Hz)] and ¹³C NMR Data [δ_{C}] of cassibiphenols A (1) and B (2) inCD₂OD at 300 K

	Cassibiphenol A (1)		Cassibiphenol B (2)	
Position	$\delta_{\rm H}$	δ_{C}	$\delta_{\rm H}$	δ_{C}
2		158.1		157.7
3		113.5		113.3
4		166.8		166.8
5		137.5		137.7
6	6.41 (1H, s)	123.1	6.41 (1H, s)	123.2
7		186.0		185.9
8	6.32 (1H, s)	103.3	6.32 (1H, d, 1.4)	103.3
9		161.5		161.7
10		103.3		103.3
11a	2.82 (1H, dd, 16.3, 16.3)	34.3	2.79 (1H, dd, 16.6, 13.4)	34.5
11b	3.01 (1H, d, 16.3)		3.01 (1H, dd, 16.6, 2.9)	
12	4.41 (1H, m)	79.0	4.45 (1H, m)	79.5
13	1.16 (3H, d, 6.1)	20.2	1.16 (3H, d, 6.3)	20.2
14	7.44 (1H, s)	117.7	7.44 (1H, s)	117.7
2′		148.5		148.2
3′	7.08 (1H, s)	131.9	7.06 (1H, s)	132.4
4′		139.0		139.2
5′		138.3		139.3
6′	6.34 (1H, s)	109.4	6.34 (1H, d, 2.0)	109.1
7′		159.0		158.9
8′	6.26 (1H, s)	101.6	6.28 (1H, d, 2.0)	101.3
9′		156.7		156.5
10′		122.0		121.8
11′a	2.21 (1H, dd, 13.0, 5.4)	44.3	2.30 (1H, dd, 13.6, 6.2)	44.1
11′b	2.71 (1H, dd, 13.0, 5.4)		2.49 (1H, dd, 13.6, 6.2)	
12'	3.58 (1H, m)	68.6	3.76 (1H, m)	68.9
13′	1.01 (3H, d, 5.9)	22.9	0.95 (3H, d, 6.1)	22.9
14′	2.55 (3H, s)	22.0	2.54 (3H, s)	22.0

The gross structure of **1** was classified into two units, 3-methyl-3*H*-isochromen-6(4*H*)-one (C-4 to C-13) and a biphenyl unit (C-2, C-3, C-14, and C-2' to C-14'), which were deduced from extensive analysis of HMBC spectrum in CD₃OD (Fig. 1).

Three partial structures, a (C-11 to C-13), b (C-14, C-2', C-3', and C-14'), and c (C-11' to C-13') were deduced from analysis of the ¹H–¹H COSY spectrum. Connection between partial structure **a** and the dienone ring, which form 3-methyl-3H-isochromen-6(4H)-one, could be assigned by HMBC correlations of H-6 ($\delta_{\rm H}$ 6.41) to C-8 ($\delta_{\rm C}$ 103.3), C-10 ($\delta_{\rm C}$ 103.3), and C-11 ($\delta_{\rm C}$ 34.3) and H-8 ($\delta_{\rm H}$ 6.32) to C-7 ($\delta_{\rm C}$ 186.0), C-9 ($\delta_{\rm C}$ 161.5), and C-10, and a four bond HMBC correlation to C-4 ($\delta_{\rm C}$ 166.8), and H-11a ($\delta_{\rm H}$ 3.01) to C-5 (δ_{C} 137.5) and C-10 (δ_{C} 103.3). Partial structure **b** in the benzene ring, which connected with C-2 and C-3 through C-4' was indicated by HMBC correlations of H-14 ($\delta_{\rm H}$ 7.44) to C-2 ($\delta_{\rm C}$ 158.1) and C-3 (δ_C 113.5) and H-3' (δ_H 7.08) to C-3 and C-4' (δ_C 139.0). The ether linkage between C-2 and C-9 could be assigned by the both down-field shifts of ^{13}C NMR data; δ_{C} 161.5 and δ_{C} 158.1. The connection of C-3 and C-4 was deduced by the ^{13}C NMR chemical data of quaternary sp² carbons; δ_{C} 113.5 and δ_{C}



Figure 1. Selected 2D NMR correlations for cassibiphenol A (1) and B (2).

166.8. The presence of 5-(2-hydroxypropyl)benzene-1,3-diol was supported by ¹H and ¹³C NMR data ($\delta_{\rm H}$ 3.58, 6.26, and 6.34; $\delta_{\rm C}$ 68.6, 101.6, 109.4, 122.0, 138.3, 156.7, and 159.0) and HMBC correlations of H-11'a ($\delta_{\rm H}$ 2.21) to C-5' ($\delta_{\rm C}$ 138.3), C-6' ($\delta_{\rm C}$ 109.4), and C-10' ($\delta_{\rm C}$ 122.0) indicating the connection of partial structure **c** and C-5'. The connectivity of C-4' and C-10' was assigned by a HMBC correlation of H-3' ($\delta_{\rm H}$ 7.08) to C-10'.

The diastereotopic methylene protons (H₂-11) at $\delta_{\rm H}$ 2.82 and 3.01 were vicinally coupled with H-12 with respective *J* values of H-11a (dd, *J* = 16.3, 16.3 Hz) and H-11b (d, *J* = 16.3 Hz), which indicated that the former proton was pseudoaxial and the latter was assignable to pseudoequatorial position. The relative configuration of **1** was deduced through inspection of a molecular model as well as the ROESY spectrum (Fig. 2), with ROESY correlations between H-3'/H-11'b, H-11'a/H₃-13', and H₃-13'/H₃-13 indicating a biphenyl bond at C-4' and C-10' was *S**-configuration and a secondary hydroxyl group at C-12' was *R**-configuration, and a methyl group at C-12 was *S**-configuration as shown in Figure 2. Therefore, Cassibiphenol A (**1**) was concluded to be a novel tetracycle connecting 5-(2-hydroxypropyl)benzene-1,3-diol.

Cassibiphenol B (2) was obtained as yellowish amorphous solid. 2 showed the similar CD curve and the same molecular formula as **1**, $C_{26}H_{24}O_6$, which was determined by HRESIMS [*m*/*z* 433.1653] $(M+H)^+$, +0.2 mmu]. ¹H and ¹³C NMR spectra are presented in Table 1. The gross structure of 2 was deduced from extensive analysis of the HMBC spectrum in CD₃OD and revealed 2 had the same planar structure as 1 (Fig. 1). According to the similar CD data of 1 and 2, they were deduced to have the same absolute configuration at C-12 and/or biphenyl configuration. The two possible relative configurations of **2** were deduced through the difference of the ¹H NMR chemical shifts from H-11' to H-13' of 2 and 1 and ROESY correlations between H-3'/H-11'a, and H-11'a/H₃-13' indicating 2 was a steroisomer at C-12' or a rotational isomer of 1 (Fig. 3). The stereochemistry of the secondary hydroxyl group of 1 and 2 could be assigned by applying the Mosher method.¹³ However, the limited amount of **1** and **2** prohibited the further investigation by the chemical means.

Plausible biogenetic pathways for **1** and **2** were proposed as shown in Scheme 1. **1** and **2** might be derived through a Michael addition of 5-acetonyl-7-hydroxy-2-methylchromone¹⁴ producing chrobisiamone A,¹⁵ followed by cleavage of an ether bond, cyclization with the α , β -unsaturated ketone, and finally acid-promoted ring disclosure¹⁶ to produce an isochromen part as shown in Scheme 1.



Figure 2. Selected ROESY correlations for cassibiphenol A (1).



Figure 3. Selected ROESY correlations for cassibiphenol B (2).



Scheme 1. Plausible biogenetic pathway for cassibiphenols A (1) and B (2).



Scheme 2. Retrosynthetic analysis for tetracyclic core of 1 and 2.

In order to determine the absolute stereochemistry of C-12, a synthesis of a tetracyclic core of cassibiphenols A (1) and B (2), **3a** and **3b** was undertaken. Our retrosynthetic analysis of **3** is outlined in Scheme 2. To construct the 3-methyl-3*H*-isochromen-6(4H)-one, left half of the tetracyclic core, **4a** and **4b** were regarded as synthetic precursors, which converted to **3** through dehydration of a tertially alcohol as the similar conversion of barakol to anhydrobarakol.¹⁶ Making the two disconnections generated the three building blocks (methoxymethyl (MOM)-protected *S*- or *R*-6-hydroxymellein¹⁷ **5a** and **5b**, and MOM protected 2-bromophe-



Scheme 3. Synthesis of enantiopure 3,4-dihydroisocoumarins, 5a and 5b.

nol **6**) needed for assembly of **4a** and **4b**. Enantiomerically pure **5a** and **5b** were obtained from **9** by inserting proper stereochemistry from the chiral reagent *S*- or *R*-propylene oxide (**8a** and **8b**). In addition, the δ -valerolactone could be synthesized through oxa-Pictet–Spengler cyclization¹⁸ and Jones oxidation.¹⁹

Our synthesis began with preparation of two enantiomerically pure MOM-protected S- or R-6-hydroxymellein (5a and 5b), which was readily accessible from a known bromobenzene 9²⁰ synthesized from commercially available 1-bromo-3,5-difluorobenzene. Regioselective boron trifluoride diethyl etherate promoted ring opening of 8a and 8b with an aryl anion of 9 to afford the phenyl-2-propanol **7a** and **7b** in 70% and 67% yields, respectively. Condensation of 7a and 7b with trimethyl orthoformate under *p*-TsOH condition vielded the isochroman acetal **10a** and **10b** in 76% and 80% vields, respectively, which were characterized to be single diastereomers of trans-pyran.²¹ Jones oxidation of **10a** and 10b led to the formation of lactones 11a and 11b, which were converted to the natural product, S- or R-6-hydroxymellein 12a and 12b by removal of both aromatic benzyl ether functions under hydrogenation. Finally, 5a and 5b were obtained by re-protection of two phenolic alcohols as MOM ethers in 86% and 80% yields, respectively (Scheme 3).

Assessment of the enantiomeric excess by HPLC analysis of **11a** and **11b** confirmed >99% enantiomeric excess (see the Supplementary data), and the each positive and negative specific optical rotation of **12a** ($[\alpha]_{D}^{20}$ +54 (*c* 0.12, MeOH)) and **12b** ($[\alpha]_{D}^{20}$ -53 (*c* 0.35, MeOH) {lit.²² [α]_D¹⁸ -51 (*c* 0.10, MeOH)}) and the characteristic Cotton effects of *S*-6-hydroxymellein, **12a** (CD (MeOH) λ_{max} (nm)/ $\Delta\epsilon$: 233 (+10.4) and 268 (+6.7)}, and *R*-6-hydroxymellein, **12b** (CD (MeOH) λ_{max} (nm)/ $\Delta\epsilon$: 233 (-9.7) and 268 (-7.0) {lit.²³ CD (MeOH) λ_{max} (nm)/ $\Delta\epsilon$: 233 (-9.0) and 268 (-8.1)) confirmed the absolute configuration.

MOM-protected 2-bromophenol **6**, which was produced from commercially available 2-bromophenol in quantitative yield, was treated with *n*-BuLi and allowed to react with **5a** and **5b** to give hemiketals **13a** and **13b** as a mixture of two diastereomers. The phenolic MOM groups of **13a** and **13b** were removed by 3 M HCl aq in MeOH, which occurred through a dehydration of tertially



Scheme 4. Construction of the tetracyclic core, 3a and 3b.



Figure 4. Comparison of CD spectra of 1, 2, 3a, and 3b.

alcohol forming isochromen skeleton and an intramolecular dehydrate cyclization to give desired **3a** and **3b** in 40% and 43% yields, respectively (Scheme 4).

In the CD spectra of **3a** and **3b**, two Cotton effects were observed in methanol. The bands of negative and positive sign around 230 and 210 nm could be assigned to the $\pi \rightarrow \pi^*$ transition of 3*H*-xanthen-3-one chromophore and the helicity of the pyran ring. Comparison of the CD spectra for **1**, **2**, **3a**, and **3b** indicated the absolute configuration at C-12 of **1** and **2** were *S* configuration as in **3a** (Fig. 4). According to the relative configuration of **1** as 12*S*,4′*S*,12′*R*.

In conclusion, structure elucidation of cassibiphenols A (1) and B (2) consisting of a novel tetracyclic skeleton and a biphenyl unit was reported and synthesis of a tetracyclic core of 1 and 2 (11% overall yield for 3a and 3b in 8 steps) for determination of the absolute configuration of 1 and C-12 of 2 was described. The strategy with two sequences of dehydrate aromatic ring construction seems to be an innovative approach to the synthesis of the tetracyclic core of 1 and 2, which would be potentially useful in the total synthesis and determination of the absolute configuration of 2. Studies in this direction are underway.

Acknowledgments

This work was supported in part by Grants-in-Aid for Scientific Research from Japan Society for the Promotion of Science (JSPS).

Supplementary data

Supplementary data (experimental details, scanned copies of NMR spectra including ¹H NMR, ¹³C NMR, ¹H–¹H COSY, HSQC, HMBC, ROESY, and chiral HPLC spectra) associated with this article can be found, in the online version, at http://dx.doi.org/10.1016/ j.tetlet.2014.01.023.

References and notes

- (a) Astulla, A.; Zaima, K.; Matsuno, Y.; Hirasawa, Y.; Ekasari, W.; Widyawaruyanti, A.; Zaini, N. C.; Morita, H. J. Nat. Med. 2008, 62, 470–472; (b) Nugroho, A. E.; Hirasawa, Y.; Piow, W. C.; Kaneda, T.; Hadi, A. H. A.; Shirota, O.; Ekasari, W.; Widyawaruyanti, Y.; Morita, H. J. Nat. Med. 2012, 66, 350–353.
- Morita, H.; Mori, R.; Deguchi, J.; Oshimi, S.; Hirasawa, Y.; Ekasari, W.; Widyawaruyanti, A.; Hadi, A. H. A. J. Nat. Med. 2012, 66, 571–575.
- Mohamad, K.; Hirasawa, Y.; Litaudon, M.; Awang, K.; Hadi, A. H. A.; Takeya, K.; Ekasari, W.; Widyawaruyanti, A.; Zaini, N. C.; Morita, H. Bioorg. Med. Chem. 2009, 17, 727–730.
- Morita, H.; Oshimi, S.; Hirasawa, Y.; Koyama, K.; Honda, T.; Ekasari, W.; Indrayanto, G.; Zaini, N. C. Org. Lett. 2007, 9, 3691–3693.
- Ekasari, W.; Widyawaruyanti, A.; Zaini, N. C.; Syafruddin, D.; Honda, T.; Morita, H. Heterocycles 2009, 78, 1831–1836.
- (a) Rudyanto, M.; Tomizawa, Y.; Morita, H.; Honda, T. Org. Lett. 2008, 10, 1921– 1922; (b) Yao, Y. S.; Yao, Z. J. J. Org. Chem. 2008, 73, 5221–5225.
- Matsumoto, T.; Kobayashi, T.; Ishida, K.; Hirasawa, Y.; Morita, H.; Honda, T.; Kamata, K. Biol. Pharm. Bull. 2010, 33, 844–848.
- (a) Mbatchi, S. F.; Mbatchi, B.; Banzouzi, J. T.; Bansimba, T.; Nsonde Ntandou, G. F.; Ouamba, J. M.; Berry, A.; Benoit-Vical, F. *J. Ethnopharmacol.* 2006, 104, 168– 174; (b) Sanon, S.; Ollivier, E.; Azas, N.; Mahiou, V.; Gasquet, M.; Ouattara, C. T.; Nebie, I.; Traore, A. S.; Esposito, F.; Balansard, G.; Timon-David, P.; Fumoux, F. *J. Ethnopharmacol.* 2003, 86, 143–147.
- 9. Deguchi, J.; Hirahara, T.; Hirasawa, Y.; Ekasari, W.; Widyawaruyanti, A.; Shirota, O.; Shiro, M.; Morita, H. *Chem. Pharm. Bull.* **2012**, *60*, 219–222.
- Deguchi, J.; Hirahara, T.; Oshimi, S.; Hirasawa, Y.; Ekasari, W.; Shirota, O.; Honda, T.; Morita, H. Org. Lett. 2011, 13, 4344–4347.
- 11. Cassibiphenol A (1): yellow amorphous solid; IR (Zn-Se) ν_{max} 3742, 3366, 2923, 1678, 1604, and 1558 cm⁻¹; UV (MeOH) λ_{max} 204 (ϵ 30860), 231 (ϵ 23410), 268 (ϵ 12220), 366 (ϵ 6050), and 423 (ϵ 11240) nm; CD (MeOH, 0.00046 M) λ_{max} ($\Delta\epsilon$) 202 (-2.7), 209 (+10.3), 231 (-11.5), and 257 (+3.4) nm; ¹H and ¹³C NMR (Table 1); ESIMS (pos.) *m/z* 433 (M+H)*; HRESIMS *m/z* 433.1655 (M+H)*, calcd for C₂₆H₂₅O₆ 433.1651.
- 12. *Cassibiphenol B* (2): yellow amorphous solid; IR (Zn-Se) ν_{max} 3730, 3260, 2952, 1683, 1597, and 1544 cm⁻¹; UV (MeOH) λ_{max} 203 (ε 31688), 231 (ε 23117), 270 (ε 12554), 359 (ε 5931), and 423 (ε 12943) nm; CD (MeOH, 0.00046 M) λ_{max} ($\Delta \varepsilon$) 203 (–0.7), 214 (+8.8), 228 (–6.1), and 242 (+3.7) nm; ¹H and ¹³C NMR (Table 1); ESIMS (pos.) *m*/z 433 (M+H)⁺; HRESIMS *m*/z 433.1653 (M+H)⁺, calcd for C₂₆H₂₅O₆ 433.1651.
- Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc. 1991, 113, 4092–4096.
- 14. Ingkaninan, K.; IJzerman, A. P.; Verpoorte, R. J. Nat. Prod. 2000, 63, 315–317.
- Oshimi, S.; Tomizawa, Y.; Hirasawa, Y.; Honda, T.; Widyawaruyanti, A.; Rudyanto, M.; Ekasari, W.; Indrayanto, G.; Zaini, N. C.; Morita, H. Bioorg. Med. Chem. Lett. 2008, 18, 3761–3763.
- Bycroft, B. W.; Hassaniali-Walji, A. W.; Jhonson, A. W.; King, T. J. J. Chem. Soc. C 1970, 1686–1689.
- (a) Curtis, R. F.; Harries, P. C.; Hassall, C. H.; Levi, J. D.; Phillips, D. M. J. Chem. Soc. C 1966, 168–174; (b) Biswas, K. M.; Mallik, H. Phytochemistry 1986, 25, 1727–1730.
- For intramolecular oxa-Pictet–Spengler cyclization, see: (a) Giles, R. G. F.; Rickards, R. W.; Senanayake, B. S. J. Chem. Soc., Perkin Trans. 1 1997, 3361–3370; (b) Xu, Y.-C.; Kohlman, D. T.; Liang, S. X.; Erikkson, C. Org. Lett. 1999, 1, 1599– 1602; (c) Bianchi, D. A.; Rua, F.; Kaufman, T. S. Tetrahedron Lett. 2004, 45, 411– 415; (d) Choukchou-Braham, N.; Mostefa-Kara, B.; Cheikh, N.; Didi, M. A. Synth. Commun. 2005, 35, 169–178; (e) Giles, R. G. F.; McManus, J. D. Tetrahedron Lett. 2009, 50, 6361–6363; (f) Sawant, R. T.; Jadhav, S. G.; Waghmode, S. B. Eur. J. Org. Chem. 2010, 4442–4449.
- Bowden, K.; Heilbron, I. M.; Jones, E. R. H.; Weedon, B. C. L. J. Chem. Soc. 1946, 39–45.
- Huang, S.; Petersen, T. B.; Lipshtz, B. H. J. Am. Chem. Soc. 2010, 132, 14021– 14023.
- Zheng, H.; Zhao, C.; Fang, B.; Jing, P.; Yang, J.; Xie, X.; She, X. J. Org. Chem. 2012, 77, 5656–5663.
- 22. Islam, M. S.; Ishigami, K.; Watanabe, H. Tetrahedron 2007, 63, 1074–1079.
- 23. Schlingmann, G.; Roll, D. M. Chirality 2005, 17, S48–S51.