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# INNOVATION OF ZEOLITE MODIFIED POLYETHERSULFONE HOLLOW FIBRE MEMBRANE FOR HAEMODIALYSIS OF CREATININE

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**Abstract.** Zeolites were added to polyethersulphone (PES) membranes in order to enhance their capability of attracting creatinine. Zeolite was blended with PES by varying its doping concentration and time, and the mechanical properties of the resulting hollow fibre membranes produced under optimal concentration were characterised. The values of tensile stress, tensile strain and Young's modulus, as well as the flux and rejection of the membranes were determined.

**Keywords:** zeolite, hollow fibre, creatinine, haemodialysis.

## 1. Introduction

In recent years the Indonesian health ministry has reported an increasing national trend in annual catastrophic incidents due to diseases. A catastrophic disease by nature incurs high therapeutic costs and often leads to complications that can endanger the lives of sufferers, and includes afflictions such as the kidney, heart, and neurological disease, cancer, diabetes mellitus, and haemophilia. The number of patients with chronic kidney disease in Indonesia continues to grow, mostly due to complications from hypertension and diabetes mellitus. According to the data released from the Indonesian Nephrology Association (PERNEFRI), over 28,000 patients in Indonesia suffer from renal failure, with the number increasing 10 % annually [1].

In the last decade, haemodialysis (HD) has been developed as a kidney replacement therapy that can increase a patient life expectancy, and is currently being improved continuously in many countries. Haemodialysis

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requires a dialysis machine and a special filter called dialyser, which is a semipermeable membrane, to clean blood. In haemodialysis, blood is removed from a patient's body, filtered through the membranes, and drained into a machine located outside the body. Kidney failure is caused by the excess creatinine in blood; the normal concentration of creatinine in blood is 0.7-1.4 mg/dl [2]. The dialysis membranes used in HD require robust mechanical properties, must be highly efficient, and be capable of supporting significant flux. In a previous study [3], a cellulose acetate polymer was used to improve the performance of HD membrane, where the polymer performs a creatinine HD flux of 49.4 l/m<sup>2</sup> and a rejection value of ~19.65 %. This showed that the membrane allowed creatinine to pass through the dialyser. However, the disadvantage of the membrane used was its relatively low mechanical strength, and in particular, its brittleness. Thus, minimizing the poor performance of HD membranes has become an important issue that has not been clearly resolved.

Zeolites are widely used as dehydrators, adsorbents, molecular filters, and ion exchangers due to their unique properties. Their use as multipurpose materials in adsorbents and molecular filters is made possible in part by their hollow structures, which allow large numbers of molecules which diameters are smaller or equal to the membrane pore size to be absorbed. In addition, zeolites are selective adsorbents with high adsorption effectiveness. Zeolite is an attractive adsorbent due to its crystalline pores that have the diameter ~0.85 nm, making them a type of toxic uremic compound [4]. To investigate the use of zeolite in HD, several zeolites were added to poly(ethylene-co-vinyl alcohol) in order to separate creatinine, demonstrating an excellent creatinine rejection value of 67 % [5]. Other polymers such as polyacrylonitrile and poly(methyl methacrylate) have also been used with zeolite in HD-related studies [6,

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7]. Moreover, an additional innovation has been the manufacture of hollow-fibre, polymer-based membranes such as polyethersulphone (PES). Polyethersulfone was chosen due to its improved resistance to fouling compared to natural and other synthetic polymer membranes, as well as their other excellent properties such as thermal stability, chemical resistance, and various mechanical properties [8, 9]. This research focuses on the use of hollow fibre membranes fabricated from zeolite-modified PES.

In principle, the modification of PES membranes with zeolite may enhance its selective molecular separation, but to date this technique has not yet been reported. Here, we report on the modification of PES membranes by zeolites, as used in creatinine HD for renal failure patients. In addition to characteristics of the zeolites, PES, and blended membrane, the performance of membranes in filtering creatinine was investigated and identified as a key aspect in need of improvement. The obtained membranes are expected to selectively filter creatinine during HD with high values of flux and rejection, while retaining excellent mechanical properties.

## 2. Experiments

### 2.1. Materials

Polyethersulfone, sodium hydroxide (NaOH, 98.5%), creatinine (98%), picric acid solution (1.3%), sodium aluminate anhydrous (NaAlO<sub>2</sub>, 50%), silicon dioxide (SiO<sub>2</sub>) were purchased from Sigma Aldrich Pte Ltd (USA). Dimethyl formamide (DMF, 99.8%) was purchased from Merck &Co. Ltd (Canada). All reagents and solvents were used directly without further purification.

# 2.2. Synthesis of Zeolite A

Zeolite powder was prepared using hydrothermal methods. About 8.2 g of NaAlO<sub>2</sub> was dissolved in 11 ml of distilled water, then dropwise mixed with 5.2 ml of SiO<sub>2</sub> for aging step following by stirring process for 3 h. The mixture was transferred on Teflon lined autoclave and heated in a convection oven at 373 K for 44 h. The mixture was filtered to obtain a residue. This residue was further washed with distilled water, dried and calcined at 723 K for 4 h.

# 2.3. Fabrication of Polyethersulfone Hollow Fibre Membrane Modified by Zeolite

Spinning dope solution to fabricate zeolite modified polyethersulfone hollow fibre membranes was

prepared with a fixed PES concentration and variation on both DMF and zeolite amount (see Table 1). PES was firstly dissolved in DMF under vigorous stirring at ambient temperature. Zeolite was further added to the homogenous solution and mixed until the zeolite was homogenously dispersed. The prepared dope solution was then set into spinneret hollow fibre system. Then the solvent (water) and dope were extruded through the inner and outer channel, respectively, of the spinneret at a controlled rate and passed through an air gap before immersing into a coagulation bath.

# 2.4. Flux and Rejection of Creatinine Analysis

Performance of the membrane on a haemodialysis process was accessed by measuring its flux and rejection of prepared creatinine solution. Flux is the volume of permeate that passes through the membrane per unit area and time. The flux values were determined by passing 100 ml of feed solution into a cross-flow system, then the value was measured every 60 min. The feed and permeate solutions of creatinine were separated by the crosslinked PES-zeolite hollow fibre membranes. The selectivity of membranes was determined by the rejection value. The rejection value of membranes was obtained by determining the concentration of the feed solution (solution before filtration process) and the permeate solution (solution after filtration process). The concentration of each solution was measured quantitatively using the UV-Vis spectrophotometer. The flux and rejection values of membranes were determined using Eq. (1) and (2), respectively:

$$J_{v} = \frac{V}{A \cdot t} \tag{1}$$

where  $J_v$  is volume flux,  $1/m^2 \cdot h$ ; *t* is time, h; *A* is a membrane surface area, m<sup>2</sup>; *V* is a permeate volume, 1.

$$R = 1 - \frac{C_{permeate}}{C_{feed}} \cdot 100 \tag{2}$$

where *R* is rejection, %;  $C_{permeate}$  is concentration of particles in permeate, mol %;  $C_{feed}$  is concentration of particles in the feed, mol %.

### 2.5. Characterization

Stress-strain data were provided by Shimadzu Autograph AGS-X at 100 kN, where the hollow fibre membranes were cut with a length of 10 cm. Then the membranes were clamped at both ends and pulled until they were broken. In the tensile test, three times replications were performed to obtain the data of force (*F*) and length ( $\Delta l$ ) of the membranes. Spectrophotometer

UV-Visible Mapada UV-6100PCS was used to determine the concentration of creatinine. The morphology of prepared membrane was provided by scanning electron microscopy (SEM) using Carl Zeiss EVO MA10. Goniometer and cross-flow system instruments were used to characterize the obtained hollow fibre membranes.

# 3. Results and Discussion

# 3.1. Fabrication of Zeolite-Modified PES Hollow Fibre Membranes

Initially, PES was dissolved in a DMF solvent at room temperature, and zeolite powder was then added to the solution and magnetically stirred until it became homogeneous. This mixture of PES, DMF, and zeolite is called a dope solution. Subsequently, phase inversion was used to cast the solution into hollow fibre membranes. An optimal membrane was produced by varying the zeolite composition added to the dope solution. Various dope solution compositions are shown in Table 1.

The optimal composition (wt %) of the PES-zeolite hollow fibre membranes produced was: PES 17, DMF 82.5 and zeolite 0.5; this composition was used in the following analysis.

# 3.2. Characterization of Zeolite and Zeolite-Modified PES Hollow Fibre Membrane

X-ray diffraction (XRD) was used to characterise the zeolite-modified hollow-fibre membranes. From the diffractogram shown in Fig. 1,the prepared zeolite displays a pattern of peaks confirming that a hydroxysodalite-type zeolite (JCPDS 31-1270) was produced, along with a small amount of zeolite V (JCPDS 39-0191) [10, 11]. The formation of multiple types of zeolites during synthesis was possible due to the complex formation mechanism used. The incorporation of zeolite into PES was investigated by Fourier transform infrared spectroscopy (FTIR). A comparison of the vibration spectra of PES-zeolite and bare PES is shown in Fig. 2. The vibration peaks of PES are the most evident in PESzeolite: the peaks at 1690, 1480, 1340 and 990  $\text{cm}^{-1}$ indicate the stretching vibrations of C=C, C-H, S=O, and C-O-C in PES, respectively. The existence of zeolite in PES is indicated by the particular peak at  $3600 \text{ cm}^{-1}$ , which is the OH vibration corresponding to a hydroxysodalite-type zeolite. Similarly, the shoulder peak at  $1020 \text{ cm}^{-1}$  and the strong peak at  $750 \text{ cm}^{-1}$  confirm the Si-O-Si and Al-O vibrations of zeolite, respectively. Thus, zeolite was confirmed to be successfully incorporated into PES membrane.

Table 1

| Composition, wt % |      | wt %    | Performance   |  |
|-------------------|------|---------|---|--|
| PES               | DMF  | Zeolite | Performance   |  |
| 17                | 82.7 | 0.3     | Dope solution was casted easily to form a hollow fibre, but the resulted membrane was brittle |  |
| 17                | 82.5 | 0.5     | Shape of hollow fibre membranes was well-ordered with a smoother surface                      |  |
| 17                | 82.3 | 0.7     | The resulted membrane was thick and the surface was slightly rough                            |  |
| 17                | 82.1 | 0.9     | Dope solution was difficult to cast into hollow fibre and the resulted membrane was rough     |  |

### Optimization results of dope solution composition

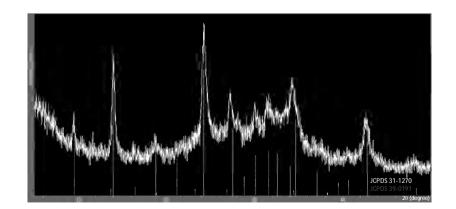


Fig. 1. Diffractogram of the synthesized zeolite type A

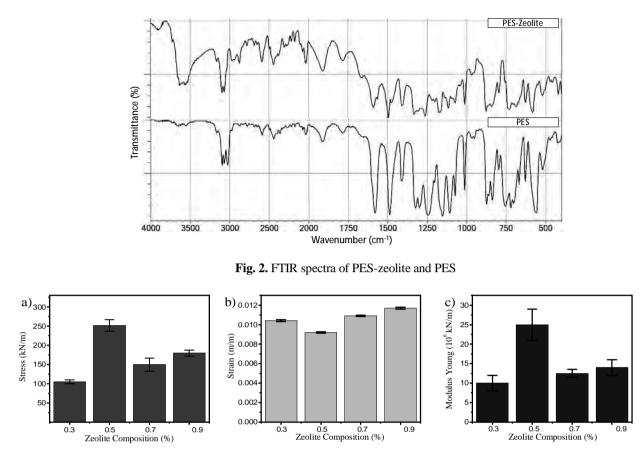


Fig. 3. Mechanical properties data of stress (a), strain (b), and Young's modulus (c) values. The data were taken with multi replication (n = 3)

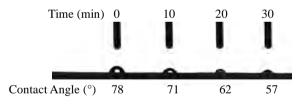


Fig. 4. Contact angels of dropped water on PEZ zeolite membrane

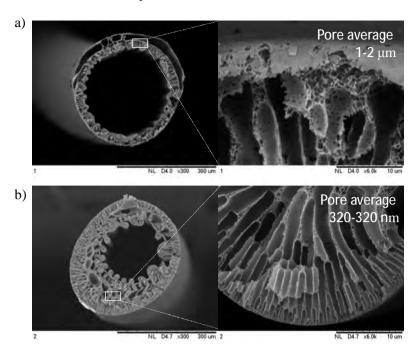
The mechanical properties of hollow fibre membranes were tested using an autograph instrument. By preparing specific specimen of the hollow fibre membrane, the stress, strain, and Young's modulus were measured and represented in Fig. 3.

Fig. 3 clearly shows that the stress in a PES-zeolite membrane with 0.5 % zeolite can reach up to ~250 kN/m;; this indicates a tight structure resistant to damage. The strain value indicates the plasticity and elasticity of the material. The strain data show results opposite to the stress data. PES-zeolite membrane with 0.5 % zeolite withstands the smallest strain. A lower strain value indicates that the membrane is more plastic. The stress and strain both affect Young's modulus, which can be used to characterise the structure, composition and defects

of the membrane. Since membranes with high values of Young's modulus are needed for HD [12], the membrane containing 0.5 % zeolite was chosen for further studies.

The hydrophilic properties of membranes can be determined by measuring their contact angle after dropping  $\sim 10 \ \mu l$  of water onto the membranes. The results can be seen in Fig. 4.

Fig. 4 shows that the water curves decreased with increasing time, indicating membrane hydrophilicity. The addition of zeolite into the PES membrane was predicted to increase hydrophilicity by contributing hydroxyl species to the membrane surface. The improvement of the hydrophilic properties of a membrane is key to its application in HD and overcoming the fouling problem commonly observed in hydrophobic membranes.



**Fig. 5.** SEM images of PES (a) and PES-zeolite (b)

SEM investigations were performed to study how the morphological structure of membranes was affected by the addition of zeolite. The surface structure and crosssection figures, seen in Fig. 5, reveal that the forms of the pores in bare PES and zeolite-modified PES are different. The pores in bare PES were not well-ordered, and the addition of zeolite made them well-ordered and asymmetric, thereby improving the ultrafiltration of uremic toxin compounds.

The pore sizes of the PES membranes were observed to be  $1-2 \,\mu\text{m}$  based on the SEM results, and became smaller with the addition of zeolites (~320–370 nm). According to Meyer [13], pore sizes for the ultrafiltration membrane should be in the range of 0.1–0.001  $\mu\text{m}$ .

# 3.3. Performance of the PES-Zeolite Hollow Fibre Membrane

The flux value of membranes, which is indicative of the ultrafiltration flow rate, was  $51.02 \text{ l/m}^2 \cdot \text{h}$ ; the rejection value of the hollow fibre membrane was 8.01 %, suggesting a decrease in creatinine concentration by 91.99 % after passing through the hollow-fibre membranes. The flux and creatinine rejection data reveal that the membranes are suitable for HD, since normal PES membranes have a creatinine rejection rate of 55 %. These interesting results reveal that a potent blend of PES and zeolite can be applied for HD.

### 4. Conclusions

Fabrication of hollow fibre membrane was successfully done using the spinneret as a casting instrument. The optimum concentration of PES:zeolite:DMF (w/w) was 17:0.5:82.5. The resulted PES-zeolites hollow fibre membranes show maximum great mechanical properties for its stress, strain and Young's modulus, which were found to be  $25 \text{ kN/m}^2$ , 0.01 mm,  $25 \cdot 10^6 \text{ N/m}^2$ , respectively. The flux and rejection values were found to be  $51.02 \text{ l/m}^2$ ·h and 8.01 %, respectively.

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### НОВІ, МОДИФІКОВАНІ ЦЕОЛІТОМ ПОЛІЕТЕРСУЛЬФОНОВІ МЕМБРАНИ З ПОРОЖНИСТИХ ВОЛОКОН ДЛЯ ГЕМОДІАЛІЗУ КРЕАТИНІНУ

Анотація. Проведено модифікацію поліетерсульфонових мембран (ПЕС) цеолітом для підвищення їх здатності притягувати креатинін. Модифікацію проводили за різних концентрацій та тривалості змішування. Визначенні механічні властивості ПЕС, отриманих за оптимальної концентрації. Визначені значення напруги при розтягненні, деформації розтягнення та модуля Юнга, а також швидкості мембранного потоку й коефіцієнту затримки мембран.

*Ключові слова:* цеоліт, порожнисті волокна, креатинін, гемодіаліз.

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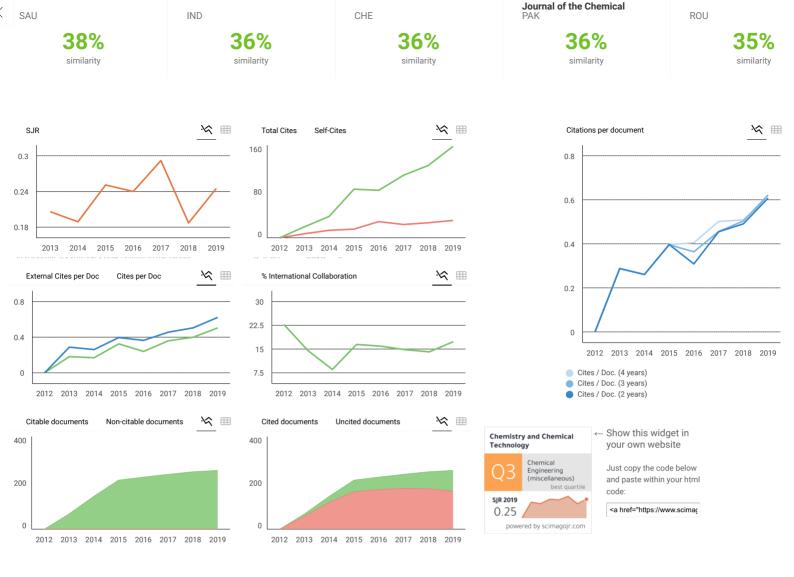
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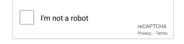
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**Olena Shyshchak** <oshyshchak@gmail.com> To: Mochamad Zakki <m.zakki.fahmi@fst.unair.ac.id> Tue, Aug 15, 2017 at 6:00 PM

Dear authors,

We received your manuscript, thanks, and sent it for the review (registration number is 060/13.07.17). After receiving referee report I'll send it to you for comments. Please, nitoce that the period of time from submitting paper to its publication is approximately 6-8 months.

Best regards,

Dr. Olena Shyshchak

Executive secretary

journal Ch&ChT



# proofs of your paper in Ch&ChT

4 messages

**Olena Shyshchak** <oshyshchak@gmail.com> To: Mochamad Zakki <m.zakki.fahmi@fst.unair.ac.id> Thu, Jul 12, 2018 at 1:02 AM

Dear authors,

See in attach proofs of your article and send me your corrections till August 23, 2018.

Please, pay attention that the articles will not be available on the journal's site till the end of the August because Publishing house, as well as Editorial Board are on vacations.

The page numbering will not be changed so you can use the information about the article for your need.

Best regards,

**Olena Shyshchak** 

Executive secretary

journal Ch&ChT



**Mochamad Zakki Fahmi** <m.zakki.fahmi@fst.unair.ac.id> To: Olena Shyshchak <oshyshchak@gmail.com> Fri, Jul 13, 2018 at 9:56 AM

Dear Editor, Thank You for the Email Except Name "Yanuardi Rahardjo" (as my co Author) it should be "Yanuardi Raharjo". Thank You

Best Regards,

Mochamad Zakki Fahmi, Ph.D (張家其) Assistant Professor, Departement of Chemistry Universitas Airlangga Phone : +62-838-32901697 Email : m.zakki.fahmi@fst.unair.ac.id



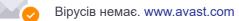
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**Olena Shyshchak** <oshyshchak@gmail.com> To: Mochamad Zakki Fahmi <m.zakki.fahmi@fst.unair.ac.id> Sat, Jul 14, 2018 at 12:38 PM

Thank you [Quoted text hidden] To: Mochamad Zakki Fahmi <m.zakki.fahmi@fst.unair.ac.id>

Dear authors, Thank you for the reply

Best regards, Olena Shyshchak



2018-07-13 5:56 GMT+03:00 Mochamad Zakki Fahmi <m.zakki.fahmi@fst.unair.ac.id>: [Quoted text hidden]



# need updated information

3 messages

### **Mochamad Zakki Fahmi** <m.zakki.fahmi@fst.unair.ac.id> To: Olena Shyshchak <oshyshchak@gmail.com>

Tue, Jun 5, 2018 at 9:45 AM

Dear Editor,

I send this message to know any updated for my 2 papers entitle" Innovation of Zeolite modified polyethersulfone Hollow Fibre Membrane for Hemodialysis of Creatinine" (Ch&CHT accepted it by June 2017) and "Efficiency of Nanobiopesticides-based Neem Seeds against Spodoptera" (Litura" (Ch&CHT accepted it by June 2017). Perhaps you can inform me when it will be putted on Ch&Cht officials website? thank you

Best Regards,

Mochamad Zakki Fahmi, Ph.D (張家其) Assistant Professor, Departement of Chemistry Universitas Airlangga Phone : +62-838-32901697 Email : m.zakki.fahmi@fst.unair.ac.id



Olena Shyshchak <oshyshchak@gmail.com> To: Mochamad Zakki Fahmi <m.zakki.fahmi@fst.unair.ac.id> Tue, Jun 5, 2018 at 8:35 PM

Dear Sir,

Your paper "Innovation of Zeolite modified polyethersulfone Hollow Fibre Membrane for Hemodialysis of Creatinine" was sent to Ch&ChT on July 13, 2017; then I noticed you about possible cancellation because you did not response after review for a long time. Your final version was received by us on November 06, 2017. Now it is on a waiting list to be published.

The similar situation with your second article "Efficiency of Nanobiopesticides-based Neem Seeds against Spodoptera" You sent it on November 10, 2017, next waiting period lasted till May 11, when you sent us revised version with corrected English. So now it is still under re-processing

The number of submitted papers is huge. Unfortunately the number of issues of the journal and the volume of every issue is limited, so the actual period from submission to publication is approximately 8-10 months. I gave notice about this in every message.

Best regards, Olena Shyshchak Executive secretary journal Chemistry&Chemical Technology

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**Mochamad Zakki Fahmi** <m.zakki.fahmi@fst.unair.ac.id> To: elika joeniarti <elika\_joe@yahoo.co.id>

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Best Regards,

Fri, Aug 24, 2018 at 7:52 PM

Assistant Professor, Departement of Chemistry Airlangga University Phone : +62-838-32901697 Email : m.zakki.fahmi@fst.unair.ac.id



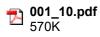
# Ch&ChT

2 messages

### **Olena Shyshchak** <oshyshchak@gmail.com> To: Mochamad Zakki <m.zakki.fahmi@fst.unair.ac.id>

Please see in attach final pdf of your article has been published in Ch&ChT

Regards, Olena Shyshchak



**Mochamad Zakki Fahmi** <m.zakki.fahmi@fst.unair.ac.id> To: Olena Shyshchak <oshyshchak@gmail.com> Mon, Sep 10, 2018 at 9:21 PM

Mon, Sep 10, 2018 at 7:03 PM

thank you very much

Best Regards,

Mochamad Zakki Fahmi, Ph.D (張家其) Assistant Professor, Departement of Chemistry Universitas Airlangga Phone : +62-838-32901697 Email : m.zakki.fahmi@fst.unair.ac.id



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