

# In Vitro Antimalarial Activity of Dichloromethane Sub-fraction of Eucalyptus globulus L. Stem against Plasmodium falciparum

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# In Vitro Antimalarial Activity of Dichloromethane Subfraction of Eucalyptus globulus L. Stem against Plasmodium falciparum

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## INTRODUCTION

Malaria is a serious infectious disease caused by protozoan parasites in tropical and subtropical regions. In 2010, malaria was endemic in about 104 countries worldwide 7d approximately 219 million cases of malaria caused 660.000 deaths. Approximately 1 % of malaria deaths occur in Africa (WHO, 2012). Global spread of multiple drugresistant malaria has become a major health problem and efforts to search for new antimalarial are needed.

Eucalyptus globulus is a plant of the Myrtaceae family that in Indonesia commonly known as kayu putih and empirically used as an antipyretic (Backer, 1968). In Brazil, E. globulus is used as an antimalarial plants (Nagpal et al., 3 10). In Cameroon, E. globulus, Carica papaya and Psidium guajava leaves are mixed and boiled as a decoction that is drunk for the treatment of malaria (Titanji et al., 2008). In Venezuela, E. globulus leaves is boi as decoction for the treatment of malaria (Carballo et al., 2004)

Our preliminary study showed that the 80% ethanol extract and dichlor 4 ethanol fraction were very active as an antimalarial with IC<sub>50</sub> of 0.090  $\mu$ g/mL and 0.022  $\mu$ g/mL, respectively.

This study aims to separate the dichloromethane fraction and to test antimalarial activity of its subfractions.

# MATERIAL AND METHODS

### Plant Material

Eucalyptus globulus stem was obtained from Cangar Forest at Malang, East Java on April 2010. Sample was authenticated by the authority of Purwodadi Botanical Garden, Pasuruan, East Java.

### Separation Method

Vacuum liquid chromatography (VLC) of dichloromethane fraction of *E. globulus* stem was performed using hexane-CHCl<sub>3</sub> (25% gradient) to CHCl<sub>3</sub>-MeOH (98:2, 96:4, 94:6, 90:10, 85:15 and 80:20).

### Thin Layer Chromatography (TLC) Method

Sub-fractions obtained from Vacuum liquid chromatography (VLC) of dichloromethane fraction were monitored by TLC using silica gel F<sub>254</sub> as stationary phase and chloroform-methanol (98:2) as bille phase. The separated spots were visualized under ultra-violet light of

two different wavelengths (UV<sub>254</sub> nm and UV<sub>365</sub> nm) 2 d visible light before and after sprayed with 10% H<sub>2</sub>SO<sub>4</sub> and heated at 105°C for 5 minutes.

### In Vitro Antimalarial Activity Test

Antimalarial activity of sub-fractions was assessed against *Plasmodium falciparum* strain 3D7 which is sensitive to chloroquine. This strain was maintained in continuous culture in flask according to the methodology described by Tragger and Jensen (1976).

Percentage inhibition was calculated using formula:

[(% parasitaemia in control wells – % parasitaemia of test wells)/(% parasitaemia of the control)] x 100 (Ngemenya et al., 2006).

IC<sub>50</sub> values refers to the concentration required to inhibit 50% of parasite's growth (Mustofa *et al.*, 2007).

### **RESULTS AND DISCUSSION**

Vacuum liquid chromatography of dichloromethane fraction produced 8 sub-fractions (D.1 - D.8 sub-fractions). TLC chromatogram of dichloromethane sub-fractions was shown in figure 1.

Antimalarial activity test showed that IC<sub>50</sub> value of each dichloromethane sub-fractions was 10.284  $\mu$ g/mL, 16.387  $\mu$ g/mL, 0.053  $\mu$ g/mL, 1.059  $\mu$ g/mL, 0.318  $\mu$ g/mL, 0.387  $\mu$ g/mL, 0.150  $\mu$ g/mL and 0.040  $\mu$ g/mL. D.8 sub-fraction has the lowest IC<sub>50</sub> value of 0.040  $\mu$ g/L. This activity was analysed in accordance with the norm of plants antimalarial activity of Rasoanaivo *et al.* (1992). According to this 10 m, an extract is very active if IC<sub>50</sub> < 5  $\mu$ g/mL, active 5  $\mu$ g/mL < IC<sub>50</sub> < 50  $\mu$ g/mL, weakly active 50  $\mu$ g/mL Based on this classification, result from this study of D.8 sub-fraction of *E. globulus* stem with IC<sub>50</sub> of 0.040  $\mu$ g/mL is said to have very active antimalarial activity. The result of antimalarial activity test of dichloromethane sub-fractions (D.1 - D.8 sub-fractions) can be seen in Table 1.

TLC test of sub-fractions indicated the presence of the most dominant spot (spot D) on D.8 sub-fraction with R<sub>f</sub> values of 0.40 which gave a red purple colour after sprayed with 10% H<sub>2</sub>SO<sub>4</sub> and heated at 105 °C for 5 minutes. Spot D began to appear on D.6 sub-fraction which has the IC<sub>50</sub> value of 0.387 µg/L. Colour intensity of spot D increased on D.7 and D.8 sub-fractions which have the IC<sub>50</sub> values lower than that of D.6 sub-fraction (0.387 µg/mL). From

these data can be seen that the higher concentration of spot D, the lower IC<sub>50</sub> value of sub-fractions. Therefore, it can be presumed that spot D on D.8 sub-fraction is a substance that is responsible for activity of D.8 sub-fraction.

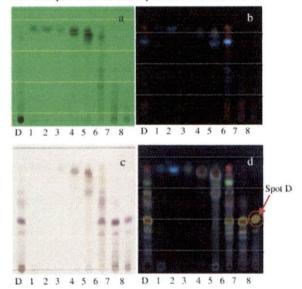


Figure 1. TLC chromatogram of dichloromethane subfraction using silica gel  $F_{254}$  as stationary phase and chloroform-methanol (98:2) as mobile phase, viewed under UV light: (2254 nm; (b) 366 nm; (c) after sprayed with 10%  $H_2SO_4$  and heated at 105°C 2 or 5 minutes. (d) 366 nm after sprayed with 10%  $H_2SO_4$  and heated at 105°C for 5 minutes; D = dichloromethane fraction, D.1-D.8 = subfraction.

Table 1. IC<sub>50</sub> values of dichloromethane sub-fractions of *E. globulus* L. stem against *P. falciparum* 

Sam	Percer	5 C50				
ple	100	10	1	0.1	0.01	(µg/mL)
D.1	73.46	40.42	31.63	21.42	16.32	10.284
D.2	63.27	46.95	29.77	10.38	1.63	16.387
D.3	89.83	79.10	69.52	51.22	41.12	0.053
D.4	87.40	54.41	44.55	39.60	21.58	1.059
D.5	92.41	75.86	52.80	43.27	24.13	0.318
D.6	90.20	70.94	55.43	36.34	27.30	0.387
D.7	89.50	73.96	58.95	45.41	36.15	0.150
D.8	92.24	80.80	72.99	53.92	41.85	0.040

### CONCLUSION

D.8 sub-fraction of *E. globulus* possesses a very active antimalarial activity and might be a good candidate for antimalarial. Further work is suggested to isolate, identify and characterize the active principles from this substance.

### **ACKNOWLEDGEMENTS**

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