# Synthesis and Molecular Docking Study of 4-Chlorophenylquinazoline-4-[3h]-One Derivatives as COX-2 Inhibitor

by Melanny Ika Sulistyowaty

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## **Research Article**

## Synthesis and Molecular Docking Study of 4-Chlorophenylquinazoline-4-[3h]-One Derivatives as COX-2 Inhibitor

Tutuk Budiati<sup>\*</sup>, Suko Hardjono, Melanny Ika Sulistyowaty

Pharmaceutical Chemistry Department, Faculty of Pharmacy, Universitas Airlangga

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#### ABSTRACT

According to data from the World Health Organization in 2014, cancer was the second biggest cause of death after heart disease. Several attempts have been made to produce the anticancer drug candidate. Modification of molecules against cancer drugs have been done so they became more effective and efficient. One mechanism of action of cancer drugs is to inhibit COX-2. Quinazolinone derivative compounds has anticancer activity, so this study synthesized some phenylquinazolinone derivatives. Virtual screening was carried out through docking the design compounds into the binding site of COX-2 enzyme (PDB code 3LN1) to predict if these compounds had analogous binding mode to the COX-2 inhibitor. Results obtained in the form of bond energy, indicated by the value of RS. Novel of phenylquinazolinone (4a-h) have been synthesized using anthranilic acid as starting material in three steps reaction. The purity of synthesized compounds was tested by TLC and m.p. data. The structures of the synthesized compound were identified using UV, IR and <sup>1</sup>H-NMR spectra. The small RS value indicated a molecular bond that was stable and predictable had high activity. The smallest RS value was -122.54 kcal / mol and synthesized ranged between 70-85%.

Keyword: molecular docking, synthesis, phenylquinazolinone, cyclooxygenase-2.

#### **10 FRODUCTION**

Cancer is one of the major causes of death in the world, the second after 15 ardiovascular disease. According to the WHO (2014), breast cancer became the most common cancer in women<sup>1</sup>. This fact makes many researchers in the world try to design a new effection anticancer drug. Breast cancer is a multi-factorial disease caused by epigenetic changes and genetic mutations that occur in genes directly involved in the process of cell division and programmed cell death<sup>2</sup>.

Non-steroidal anti-inflammatory drugs (NSAIDs) appear to reduce the risk of developing cancer. One mechanism through which NSAIDs act to reduce carcinogenesis is to inhibit the activity of cyclooxy-genase-2 (COX-2), an enzyme that is over-expressed in various cancer tissue. Over-expression of COX-2 increases cell proliferation and inhibits apoptosis. Several attempts have been made to produce the anticancer drug candidate. Modification of molecules against cancer drugs have been done so they became more effective and efficient. One mechanism of action of cancer drugs is to inhibit COX-2. Quinazolinone derivative compounds have anticancer activity. Compounds containing the 4[3H]-quinazolinone ring system posses various biological activities3. Some 2,3diaryl-4[3H]-quinazolinone derivatives exhibit CO1-2 inhibitory and anti-inflammatory activities<sup>4,5</sup>. The majority of COX-2 inhibitors are diaryl heterocycles. The heterocycle part could be a five- or six membered ring, a quinazoli 1 ring, or as a acyclic form (Fig. 1: a, b, c). The precence of para-sulfonamides or para-sulfonylmethanes

on one of the aryl ring was found to be essential for optimum COX-2 selectivity and inhibitory potency<sup>6</sup>. On the other aryl ring, no specific substituents were needed. So, in this study, we will synthesize some 4chlog phenylquinazoline-4-[3H]-ones (Fig. 1d), followed by molecular docking and in silico studies that were carried out in an attempt to evaluate the drug candidature as COX-2 inhibitor.

### MATERIAL AND METHODS

#### hemistry. General Procedures

All solvents, chemicals, and reagents were obtained commercially and used without purification. Purity test of the products was performed by the **15**C methods on silica gel 60 F254 plates (Merck). 5 pot detection was performed with UV 254 nm. Melting points were measured with an Electrothermal melting point apparatus without correction. Infrared (IR) spectra were recorded in KBr pelle 10n a FTIR spectrophotometer (Jasco FT-IR 5300). <sup>1</sup>H-NMR spectra were recorded on a JEOL NMR 500 spectrometer, using TMS as internal standard.

Synthesis of 4-chlorophenylquinazoline-4-[3h]-one derivatives

The title compounds **4a-h** was synthesized stepwise by the method summarized in Scheme 1.

Synthesis of 2-(4-chlorophenyl)-4H-3,1-benzo-xazin-4one (2).

4-chlorobenzoyl chloride (0.015mol) was added drop wise to a stirred solution of anthranilic acid (1) (0.01mol)

\*Author for Correspondence: tutukbudiati@yahoo.com.

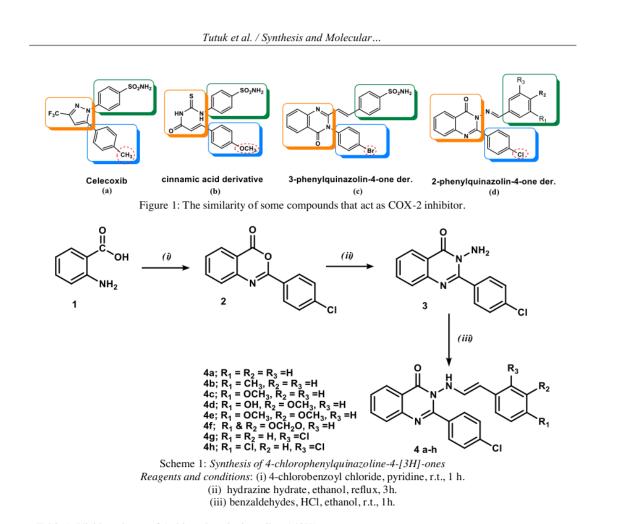


Table 1: Yields and m.p. of 4-chlorophenylquinazoline-4-[3H]-one
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Compound No.	Substituents (R1, R2, R3)	Yields (%)	m.p. ( <sup>0</sup> C)	Rerank Score (RS)
2		85 ± 3	171 - 173	
3		76 ± 5	225 - 226	
4a	$R_1 = R_2 = R_3 = H$	73 ± 2	237 - 239	-112.90
4b	$R_1 = CH_3, R_2 = R_3 = H$	85 ± 1	254 - 256	-106.15
4c	$R_1 = OCH_3, R_2 = R_3 = H$	79 ± 1	229 - 231	-112.75
4d	$R_1 = OH, R_2 = OCH_3, R_3 = H$	86 ± 2	240 - 242	-122,54
4e	$R_1 = OCH_3, R_2 = OCH_3, R_3 = H$	82 ± 1	255 - 257	-105,65
4f	$R_1 \& R_2 = OCH_2O, R_3 = H$	77 ± 2	239 - 240	-122,05
4g	$R_1 = R_2 = H, R_3 = Cl$	70 ± 1	223 - 224	-115,83
4h	$R_1 = Cl, R_2 = H, R_3 = Cl$	65 ± 3	243 -245	-117,40
4i	Celecoxib			-139.61

in pyridine ( $\overline{30}$  ml) and the reation mixture was stirred at room temperature for 1 h. The reaction mixture was poured onto cold water and filtered off through a Buchner funnel. The solid 11 pduct was washed with cold 10% Na<sub>2</sub>CO<sub>3</sub> solution. The solid obtained was filtered, wash several times with water and crystallized from acetonwater.

Synthesis of 3-amino-2-(4-chlorophenyl)quina-zolin-4(3H)-one(**3**).

A mixtu of 2-(4-chlorophenyl)-4H-3,1-benzoxazinone(2) (0.01mol) and hydrazine hydrate (0.02mol) in ethanol (50ml) **148** heated under reflux for 3 h. The separated solid was filtered, washed with water and crystallized with ethanol.

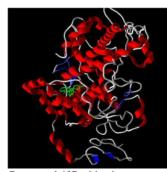
Synthesis of (E)-3-(benzylideneamino)-2-(4-

chlorophenyl)quinazolin-4(3H)-one derivatives (4a-h) 3-amino-2-(4-chlorophenyl)quinazolin-4(3H)-one (3), benzaldehyde or their derivatives (0.75mol) and 2 drops HCl in ethanol (30ml 1 were mixed and stirred at room temperature for 1 h. The reaction mixture was poured onto cold water and filtered off through a Buchner funnel. The solid product was washed with cold 10% Na<sub>2</sub>CO<sub>3</sub>

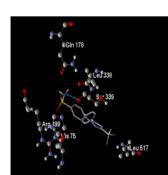
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Celecoxib Position in 3LN1



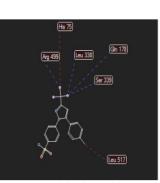
Compound 4f Position in 3LN1



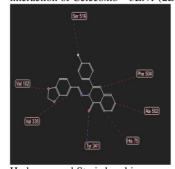
Hydrogen bond in interaction of Celecoxib – 3LN1 (3D)



Hydrogen bond in interaction of Compound 4f – 3LN1 (3D)



Hydrogen and Steric bond in interaction of Celecoxib – 3LN1 (2D)



Hydrogen and Steric bond in interaction of Compound 4f – 3LN1 (2D)

Figure 2: Molecular docking of Celecoxib and Compound 4f in interaction with 3LN1.

solution. The separated solid was filtered, washed with water and crystallized with ethanol. Molecular Docking Study

Molecular Docking Sluay

To estimate the anticancer activity of derivatives of some 4-chlorophenyl-quinazoline-4-[3H]-ones, molecular docking was performed using Molegro Virtual Docker (MVD) Ver.5.5. We used receptor cyclooxygenase-2 or COX-2, (PDB code: 3LN1) as the target protein. 3LN1 is a receptor model of celecoxib inhibitor, with the ligand code: S58\_701. The most stable chemical conformation of the target compounds was determined. The active conformation of 3LN1 and S58\_701 was selected for the preparation step. Cavity was detected to select the pocket binding site (active site of the enzyme). The target compounds (4a-h) and celecoxib, were then docked on to the protein, on the same cavity. The results, concluded which conformation produced the lowest energy state when bound to the target protein, were shown as rerank score (RS).

#### RESULT AND DISCUSSION

#### Chemistry

In the present work, an attempt has been made to undertake the synthesis of some 4-chlorophenylquinazoline-4-[3H]-ones through a multi step process. For this purpose, the required 2-(4-chlorophenyl)-4H-3,1benzoxazin-4-one (2) was prepared by cyclisation reaction between anthranilic acid 1 and 4-chlorobenzoyl

chloride at room temperature using pyridine as a solvent and also a base. Formation d the product was confirmed by a sharp band at 1768 cn 16 due to the presence of carbonyl group along will a peak at 1249 cm<sup>-1</sup> for C-O stretching in IR spectra. In the <sup>1</sup>H-NMR spectra showed eight protons of the two benzenoid rings appear as multiplet peak at 7.51 - 8.20 ppm. Benzoxazine (2) was converted to 3-amino-2-(4-chlorophenyl)quinazolin-4(3H)-one (3) by its nucleophilicsubstitution reaction with hydrazine hydrate in ethanol under reflux condition for 3 hours. Insertion of nitrogen in the ring was characterized by disappearance of band at 1249 cm<sup>-1</sup> of C-O and shift of carbonyl band from 1768 to 1662 cm<sup>-1</sup>. Appearance of new peaks near 3299 and 3209 cm<sup>-1</sup> for N-H stretching also helped in assigning structure of (3). In the <sup>1</sup>H-NMR spectra showed eight protons at 7.26 - 8.27 ppm (multiplet), and a broad singlet peak of two protons of NH<sub>2</sub> at 4,46 - 4.92 ppm. When (3) was treated with offerent benzaldehydes using HCl as catalyst in ethanol and stirred at room temperature for 1 h, the reaction goes by its condensation afforded corresponding arylidenes derivatives (4a-h) in quantitative yields. The existence of donor-electron groups in the para position of benzaldehydes will facilitate modthe reaction so that the yield is high. On the other hand, electron-withdrawing group, such as Clsubstituent, will complicate the reaction. Substituent in the ortho position will decrease the reaction yield (Table 1). The formation of 4a was

proved by the existence of the benzilidene group; it was identified from NMR spectra, ie there is one proton signal at 11.92 (s, 1H) (compound **6a**) or 11.55 (s, 1H) (compound **6b**) which is a proton from = C-H. Lactam ring-6 remains from the IR spectra in the wave number of 1657 (**6a**) or 1671 cm<sup>-1</sup> (**6b**).

Molecular Docking Study Result

The molecular docking study of tested compounds with receptor 3LN1 is shown in Table 1. Rerank score (RS) is a parameter which strictly determines potential activity of drug-receptor. It is also a logarithmic cumulative energy between drug-receptor interaction by hydrogen, electronic, and steric bond interaction. The smaller rerank score shows the smaller amount of energy required in forming drug-receptor interaction that give an assumption the drug is more suitable to occupying active site of the receptor. Therefore, compound which has the lower rerank score than the ligand, to be estimated has stronger activity than the ligand's. Table 2 showed the rerank scores of the tested compounds, including the ligand, Celecoxib. The synthesized compounds (4a-h) have RS score range at -105.65 to -122.54; lower than the RS of celecoxib.

#### Experimental Section

2-(4-chlorophenyl)-4H-3,1-benzoxazin-4-one (2).

A brownish yellow crystalline powder was obtained, yield 85%; mp. 171 -  $173^{0}$ C; UV-Vis (EtOH), nm: 288 and 300. IR (KBr)  $v_{max}$  cm<sup>-1</sup>: 1768, 1611, 1479, 1316, 1249, 764. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.51 - 8.20 (multiplet), 8H.

3-amino-2-(4-chlorophenyl)quinazolin-4(3H)-one(3).

A white crystalline powder was obtained, yield 76%; mp. 224 - 226°C; UV-Vis (EtOH), nm.: 240; 274 and 304. IR (KBr)  $v_{max}$ , cm<sup>-1</sup>: 3299 & 3209, 1662, 1596, 1442, 1316, 748. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.26 - 8.27 (multiplet), 8H; 4,46 - 4.92 (singlet), 2H.

(E)-3-(benzylideneamino)-2-(4-chlorophenyl)quinazolin-4(3H)-one(**4a**)

White crystalline powder, yield 73%; mp. 237 - 239<sup>0</sup>C; UV-Vis (EtOH), nm.: 292 and 306. IR (KBr)  $v_{max}$ , cm<sup>-1</sup>: 3058, 1674, 1594, 1443, 1279, 752. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.27 - 8.57 (multiplet), 13H; 11.99 (singlet), 1H.

(E)-2-(4-chlorophenyl)-3-(4-

*methylbenzylideneamino)quinazolin-4(3H)-one*(**4b**). white voluminous powder; yield 85%; mp. 255 - 256°C;

UV-Vis (EtOH), nm.: 320. IR (KBr)  $\nu_{max}$ , cm<sup>-1</sup>: 3058, 2938, 1674, 1595, 1442, 1312, 750. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.23 – 8.69 (multiplet), 12H; 12.00 (singlet), 1H; 2,38 (singlet), 3H.

(E)-2-(4-chlorophenyl)-3-(4-methoxybenzylidene-

amino)quinazolin-4(3H)-one(4c).

White voluminous powder; yield 79%; mp. 229 - 230<sup>0</sup>C; UV-Vis (EtOH), nm.: 322. IR (KBr)  $\nu_{max}$ , cm<sup>-1</sup>: 3026, 2920, 1680, 1598, 1445, 1319, 745. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 6.81 – 8.64 (multiplet), 12H; 12.07 (singlet), 1H; 3.81 (singlet), 3H.

(E)-2-(4-chlorophenyl)-3-(4-hydroxy-3-

methoxybenzylideneamino)quinazolin-4(3H)-one(4d).

white voluminous powder; yield 86%; mp. 240 - 241°C; UV-Vis (EtOH), nm.: 330. IR (KBr)  $v_{max}$ , cm<sup>-1</sup>: 3485 *sh*, 3045, 2949, 1670, 1633, 1440, 1278, 1023, 751. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 6.81 - 8.57 (multiplet), 11H; 11.93 (singlet), 1H; 3.85 (singlet), 3H; 9.56 (singlet), 1H. (*E*)-2-(4-chlorophenyl)-3-(3,4-

dimethoxybenzylideneamino)quinazolin-4(3H)-one(**4e**). white voluminous powder; yield 82%; mp. 255 - 256°C; UV-Vis (EtOH), nm.: 249 and 330. IR (KBr)  $v_{mux}$ , cm<sup>-1</sup>: 3025, 2919, 1679, 1636, 1447, 1268, 1020, 746. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 6.98 - 8.55 (multiplet), 11H; 11.94 (singlet), 1H; 3.82 (singlet), 6H.

(E)-3-(benzo[d][1,3]dioxol-5-ylmethyleneamino)-2-(4chlorophenyl)quinazolin-4(3H)-one (**4f**).

white voluminous powder; yield 77%; mp. 239 - 240<sup>o</sup>C; UV-Vis (EtOH), nm.: 295 and 330. IR (KBr)  $\nu_{max}$ , cm<sup>-1</sup>: 3061, 2901, 1643, 1522, 1438, 1313, 1020, 739. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 6.95 - 8.57 (multiplet), 11H; 11.97 (singlet), 1H; 6.10 (singlet), 2H.

(E)-3-(2-chlorobenzylideneamino)-2-(4-

chlorophenyl)quinazolin-4(3H)-one(4g).

white voluminous powder; yield 70%; mp. 223 - 224<sup>0</sup>C; UV-Vis (EtOH), nm.: 294 and 312. IR (KBr)  $\nu_{max}$ , cm<sup>-1</sup>: 3053, 2920, 1682, 1599, 1447, 1271, 746. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.20 - 8.87 (multiplet), 12H; 12.07 (doublet), 1H.

(E)-3-(2,4-dichlorobenzylideneamino)-2-(4-

chlorophenyl)quinazolin-4(3H)-one(4h).

white voluminous powder; yield 65%; mp. 243 - 244<sup>0</sup>C; UV-Vis (EtOH), nm.: 318. IR (KBr)  $\nu_{max}$ , cm<sup>-1</sup>: 3053, 2919, 1674, 1596, 1446, 1278, 757. <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>),  $\Box$ /ppm: 7.21 – 8.59 (multiplet), 11H; 12.03 (singlet), 1H.

#### CONCLUSION

In this study, we had synthesized some derivatives of 4chlorophenylquinazoline-4-[3H]-ones, with the range of 65% to 86% yields. From *in silico* stud 3 data the smallest RS value was -122.54 kcal/mol. The molecular docking study signified that the compounds can act as COX-2 inhibitor. The *generated pharmacophore could further be* used to design and develop new drugs.

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