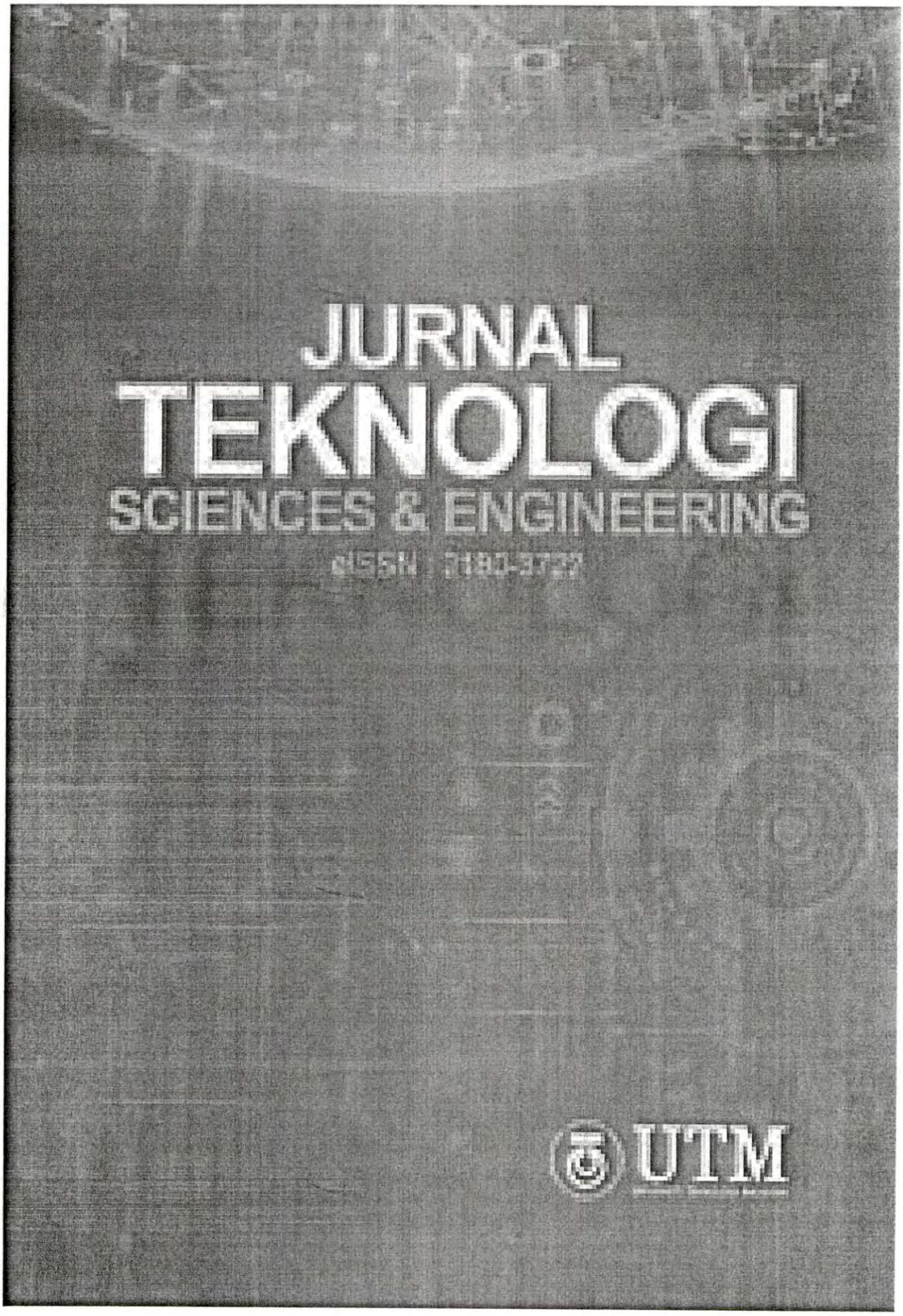
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Full Paper

SYNTHESIS OF THIOUREA DERIVATIVES FROM *M*-METHOXYCINNAMIC ACID AS ANTIANGIOGENIC CANDIDATE

Juni Ekowati[a*], Iwan Sahrial Hamid[b], Kholis Amalia Nofianti[a], Shigeru Sasaki[c]

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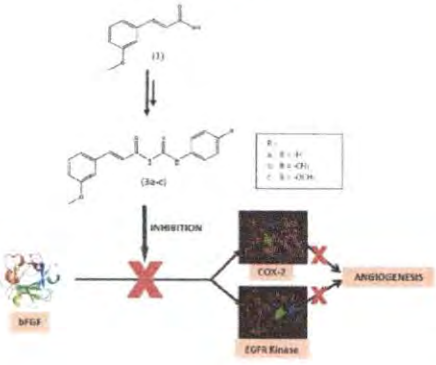
Graphical abstract



Abstract

Microwave-assisted nucleophilic acyl substitution was employed to obtain thiourea derivatives (3a, 3b, 3c) from *m*-methoxycinnamic acid (1). This synthesis method successfully yielded 60-70% reaction product. *In vivo* anti-angiogenic evaluation was conducted by chick chorioallantoic membrane model, by which each of the derivative at dose 30, 60, and 90 µg induced by bFGF and compared to celecoxib 60 µg as positive control. It was found that all of the synthesized compound at the tested dose were able to inhibit neovascularization and formation of endothelial cell of new blood vessels by 51-75%. *In silico* analysis predicted that the anti-angiogenesis mechanism of all the synthesized compounds is through the inhibition of EGFR kinase and COX-2. *N* atom acts as hydrogen bonding acceptor by residue Gly526 of COX-2. While thiourea moieties of 3a-c have hydrophobic interaction by residues Ser530, Tyr385, and Leu352. In addition, the carbonyl group of thiourea of compound 3a-c inhibit EGFR kinase through the interaction with lys745. The pKCSM data revealed that 3a-c absorbed in intestine by 89-92%, and acut toxicity in rat category 4, suggesting that the compounds show good absorption, and low toxicity. In conclusion, this study successfully synthesized thiourea derivatives, which have anti-angiogenesis activity, tested by CAM model.

Keywords: Microwave irradiation, angiogenic inhibitor, COX-2, EGFR, celecoxib, ADMET profile

1.0 INTRODUCTION

Researchers reported that angiogenesis was one of the most contributing factor to the progressive growth of neoplasms and metastases which led to the death of cancer patients. It is a fundamental process of forming new blood vessels as an extension of the existed vasculature. This process is a critical step in tumor progression to supply oxygen and nutrition, by which, the cell proliferation and metastases occur [1, 2].

Cyclooxygenase-2 (COX-2) catalyzes the conversion of arachidonic acid into PGE2 prostaglandins, which result in angiogenesis [3, 4]. Therefore, the inhibition of COX-2 pathway through binding with EP4 receptor will also restrain the angiogenesis [5, 6].

There have been many anti-angiogenesis inventions, including US Patent No. US8,778,340 B2; about anti-angiogenesis therapy for ovarian cancer, the drug bevacizumab, an anti-VEGF antibody [7]. Bevacizumab combined with other chemotherapy is proven efficacious for cancer

patients. Unfortunately, bevacizumab also causes thromboembolic disorders, fatigue, intracranial hemorrhage, proteinuria, hypertension, and bowel perforation. [3, 8, 9].

Over the past few decades, vascular endhotelial growth factor (VEGF) signaling has been identified as a central axis in tumor angiogenesis [10]. One of the anti-angiogenesis therapies is tamoxifen sunitinib, which target on VEGF signaling pathway and kinase (11). On the other hand, long-term use of tamoxifen as chemotherapy, actually increases the VEGF levels in patients and stimulate the formation of new blood vessels that trigger metastases [12].

Celecoxib (CXB), a selective COX-2 inhibitor, suppresses VEGF gene expression by targeting the VEGF promoter. However, the use of celecoxib, also poses severe side effects, especially for patients with heart diseases [6, 12, 13]. Those fact show that there is a need to develop an alternative anti-angiogenesis agent as a strategic step in cancer treatment.

Some cinnamic derivatives, such as ethyl *p*-methoxycinamate, is reported for its activity to inhibit cancer growth, COX-2 production and angiogenesis [14-16]. Research also reveals that ferulic acid, (4-hydroxy-3-methoxycinnamic acid) and *p*-methoxycinnamic acid were able to inhibit COX-2 and cancer growth [17–19]. However, up this date, there is no report that explore the potency of its isomer, *m*-methoxycinnamic acid, or (*E*) -3- (3-methoxyphenyl)prop-2-enoic acid (1) and the derivatives, related to their activity on COX-1/2 and angiogenesis. In our previous study (data not shown), ferulic acid, which has a methoxy group at the meta position show analgesic activity and inhibition of COX-1/2.

The thiourea derivatives of *p*-methoxycinnamic acid are reported as chemopreventive agents that were able to restrain the fibrosarcoma growth in mice induced by benzopyrene [20]. Thiourea derivatives has analgesic activity by inhibiting COX-1/2 [21,22]. Gorab *et al.* (2017) also described that thiourea derivatives which have the sulphonamide groups can inhibit cancer through COX-2 barriers. Thiourea moiety has a hydrophobic interaction with COX-2 amino acid residues [22]. Therefore, this study aims modify the structure of compound 1 to thiourea derivatives and evaluate their activity to inhibit the angiogenesis COX-2 activity.

Over the past few years, innovative and important developments have taken place in microwave (MW) assisted synthesis methods. The use of single mode MW reactors, especially in continuous flow reactions, shows satisfactory results. This method provides several advantages such as the use of microwaves as an energy source to increase temperatures quickly because all samples are heated together so that synthesis time is more efficient and reaction products increase. Therefore, the thiourea derivatives of **1** was synthesized using microwave irradiation as an effort to implement green chemistry [24].

Molecular docking is one of the most common methodologies used in the discovery of novel small-molecule inhibitors. This computational technology is very helpful for the medicinal chemists in identifying the inhibitor of the biomolecular targets. This approach allows us to predict the interaction between the inhibitor and the residues [14, 16, 25].

Prabhu *et al.* (2014) reported that the new VEGF inhibitors showed the hydrogen bonding between hydroxyl groups in some of the ligand with negatively charged of residue Asp1046. The ligand also formed a cation-π interaction of the amino acid and the aromatic rings of the ligand. Besides that, the hydrophobic interaction also found in the residues of Ala866, Phe1047, Cys919, Phe918, Val848, and Cys1045. The nitrogen containing six-membered ring in the ligand performed a hydrophobic interactions of the residues Leu889, Ala866 [25].

Coskun *et al.* (2018) also reported the use of molecular docking experiment to analyze the potency of diflunisal derivatives as anti-cancer through the inhibition of COX enzyme. The results were compared to the *in vitro* assay [26].

So, in our present study, we intend to synthesize thiourea derivatives of *m*-methoxycinnamic acid using microwave irradiation, to analyze the activity using CAM model and to predict the inhibition mechanism by molecular docking study.

2.0 METHODOLOGY

2.1 Materials

m-Methoxycinnamic acid was obtained from Tokyo Chemical Industry Co., Ltd., Japan. Other used materials, i.e. aniline, *p*-toluidine, *p*-anisidine, and all solvents in p.a. grade were obtained from E-Merck, Germany. bFGF was taken from Sigma Aldrich, Germany.

2.2 Synthesis Derivates Thiourea of *m*-methyoxycinnamic Acid

Into a solution of 15 mmol *m*-methoxycinamic acid in 15 ml benzene, 1 drop of pyridine and 2.4 eq thionyl chloride were added, then refluxed overnight. After that, excess of thionyl chloride and benzene are removed by rotary evaporation. The addition of benzene was repeated and removed again so that *m*-methoxycinamoyl chloride was produced, and was stored under Nitrogen condiition. Afterward, the solution of 7.5 mmol *m*-methoxycinamoyl chloride in 5 ml dichloromethane was reacted with 10 mmol ammonium thiocyanate and catalyzed by one

drop of PEG 400. The mixture was irradiated by microwave at 132W for approximately 30 seconds. The irradiation was repeated 3 times. Next, 7.5 mmol aniline was put into the mixture, and then irradiated again for 4 x 30 seconds. After that, 2N HCl solution was added into the mixture to remove the pyridine. The crude product was neutralized by 10% $NaHCO_3$ solution, and then washed by distilled water. The precipitate was filtered and recrystallized by a mixture of dichloromethane-ethanol (1-1). This procedure was used for other aromatic amines (i.e *p*-toluidine and *p*-anisidine).

2.3 Physicochemical Study

The physicochemical study of thiourea derivatives was conducted through Chem Draw Professional 15.0 program and pkCSM tool on line.

2.4 Antiangiogenesis Study

The embryonated chicken eggs in nine-day-old, obtained from PUSVETMA Surabaya, were incubated at 37°C and 60-70% humidity for one day. New vascular induction was performed using bFGF 60 ng which was dissolved in 60 mL of rh of bFGF at a concentration of 1 ng / mL of Tris HCl under aseptic conditions. The test doses were 30, 60 and 90 µg. Compared to celecoxib at 60 µg dose as positive control. The eggs were divided into eleven groups, each group contained five eggs. the first to third group were treated by compound 3a, fourth to sixth group were given compound 3b, seventh to ninth group were undertaken compound 3c, the tenth group was treated with celecoxib 60 µg, and the eleventh group was a negative control group, without any treatment except induction with bFGF. After 1 day of incubation, a 1 cm^2 hole in the top of the egg was formed and air was released from the air chamber. Samples were dropped onto the paper disk, and impregnated into chorioallantoic membrane (CAM) of each chicken embryo. The hole in the egg was closed again and the egg was returned into the incubator until the 11th day of the chicken embryo development. After that, the shell was opened, CAM was taken from the shell, then sliced for histological examination and staining by Hematoxylin and Eosin (HE). The formation of endothelial cells in neovascular capillaries was observed in the CAM cross section using a reverse H600L contrast phase microscope. The number of endothelial cells was determined in five visual fields of each slide at 400x magnification and compared to the positive and negative controls for subsequent analysis [14]. Statistical analysis between treatment and control groups was tested by one way ANOVA, followed by LSD test. The difference was considered significant at $p < 0.05$.

2.4 Docking Study

The molecular docking study of AMMS derivative, celecoxib and ligand reference FMM_91 [A] (lapatinib tosylate) into the three-dimensional structure of tyrosine kinase was performed using version 5.5 of the Molegro Virtual Docker (MVD) software. The structure of tyrosine kinase receptor was obtained from the Protein Data Bank (PDB 1XKK). Ligand preparation was carried out using ChemBio Ultra version 10.0 software; geometry optimized using MMFF 94 method and saved in Sybyl Mol2 format. All ligands were placed into the 1XKK (cavity-1) binding site parallel to the FMM_91 [A] ligand reference and 10 independent runs were carried out. The interactions between ligands and enzymes (docking scores) were predicted through the rerank score (RS). A lower negative score (kcal / mol) indicated a stronger ligand enzyme bond. Validation of the docking study was carried out by FMM_91 [A] redocking to port 1 of 1XKK. The best docking results could be observed visually by comparing the molecular structure of the test with the FMM_91 [A] crystal structure to the active site [14].

3.0 RESULTS AND DISCUSSION

3.1 Synthesis Derivates Thiourea of m-methyoxycinnamic Acid

The reaction scheme of synthesis thiourea derivatives using microwave irradiation as a source of energy is shown in Figure 1.

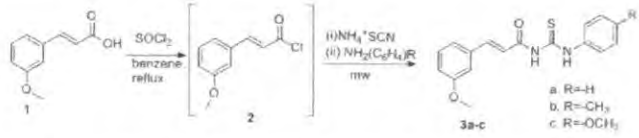


Figure 1 Synthesis route of thiourea derivatives of **1**

(3a): (*2E*)-3-(3-Methoxyphenyl)-*N*-(phenylcarbamothioyl)acrylamide
yellow needle crystal, MP: 174-175°C. Rf: 0.7 (EtAc-*n*-Hexane, 1:1). IR (KBr; cm^{-1}): 3463 (-N-H, amide, 2°); 3223(-N-H, amine, 2°); 3031 (Csp^2-H); 1673(-C=O); 1598(-C=C-, alkene); 1547(-C=S); 1168 (-C-O-C); 749 (aromatic ring metha- substituted). UV/Vis λ_{max} (EtOH) nm (log ε): 302. ^{1}H NMR (400 MHz, $CDCl_3$): 3.75 (3H, s), 6.65 (1H, d, *J* = 16*Hz*), 6.97 (1H, dd, 8*Hz*, 2*Hz*), 7.06 (1H, t, *J* = 2*Hz*), 7.11 (1H, d, *J* = 7.6 *Hz*), 7.26 – 7.31 (2H, m), 7.40 (2H, t, *J* = 8*Hz*), 7.66 (2H, d, *J* = 7.6*Hz*), 7.78 (2H, d, *J* = 16*Hz*), 9.97 (1H,s), 12.66 (1H,s). ^{13}C NMR (100 MHZ, $CDCl_3$) δ:55.42 (1C), 112.79 (1C), 117.55 (1C), 118.70 (1C), 121.61 (1C), 124.49 (1C), 127.04 (2C), 129.95 (2C),

130.18 (1C), 135.12 (1C), 137.64 (1C), 146.60 (1C), 160.04 (1C), 166.08 (1C), 178.93 (1C). MS (ESI-HRMS), $[M+Na]^+$: m/z (%) = 335.08. Products yield: 60%.

(**3b**): (2*E*)-3-(3-methoxyphenyl)-*N*-(4-methylphenyl carbamothioyl)acryl amide
Pale yellow crystal, MP: 202-203°C. Rf: 0.8 (EtAc-*n*-Hexane, 1:1). IR (KBr): 3478 (-N-H, amide, 2°); 3194(-N-H, amine, 2°); 3009 (Csp^2-H); 1677(-C=O); 1597(-C=C-, alkene); 1538(-C=S); 1156 (-C-O-C); 779 (aromatic ring metha- substituted). UV/Vis λ_{max} (EtOH) nm (log ε): 302. ^1H NMR (400 MHZ, $CDCl_3$) 2.33 (3H, s), 3.82 (3H, s), 6.54 (1H, d, *J* = 15.6 *Hz*), 6.97 (1H, dd, *J* = 8*Hz*, 2.4*Hz*), 7.06 (1H, t, *J* = 2.4*Hz*), 7.19 (2H, d, *J* = 8*Hz*), 7.30 (1H, t, *J* = 8*Hz*), 7.53 (2H, d, *J* = 8.4*Hz*), 7.77 (2H, d, *J* = 15.6), 9.18 (1H, s), 12.48 (1H, s). ^{13}C NMR (100 MHz $CDCl_3$): 21.12 (1C), 55.40 (1C), 113.07 (1C), 117.38 (1C), 118.86 (1C), 121.45 (2C), 124.30 (2C), 129.50 (2C), 130.12 (1C), 135.19 (1C), 136.84 (1C), 146.40 (1C), 160.15 (1C), 165.82 (1C), 178.81 (1C). MS (ESI-HRMS) $[M+Na]^+$: m/z (%) = 349.0981(100). Products yield : 70%.

(**3c**): (2*E*)-3-(3-methoxyphenyl)-*N*-(4-methoxyphenyl carbamothioyl) acrylamide
Pale yellow crystal, MP: 168-169°C. Rf: 0.7 (EtAc-*n*-Hexane, 1:1). IR (KBr): 3478 (-N-H, amide, 2°); 3168(-N-H, amine, 2°); 3037 (Csp^2-H); 1670(-C=O); 1595(-C=C-, alkene); 1547(-C=S); 1173 (-C-O-C); 776 (aromatic ring metha- substituted). UV/Vis λ_{max} (EtOH) nm (log ε): 300. ^1H NMR (400 MHz, $CDCl_3$): 3.76 (3H, s), 3.81 (3H, s), 6.64 (1H, d, *J* = 15.6*Hz*), 6.91 (1H, d, *J* = 8*Hz*), 7.05 (1H, t, *J* = 2.4*Hz*), 7.29 (2H, t, *J* = 8*Hz*), 7.30 (2H, t, *J* = 8*Hz*), 7.51 (2H, d, *J* = 2*Hz*), 7.77 (1H, d, *J* = 15.6), 9.65 (1H, s), 12.41 (1H, s). ^{13}C NMR (100 MHz $CDCl_3$): 55.40 (1C), 55.60 (1C), 112.81 (1C), 114.14 (2C), 117.48 (1C), 118.80 (1C), 121.60 (2C), 126.17 (1C), 130.16 (1C), 130.60 (1C), 135.14 (1C), 146.42 (1C), 158.35 (1C), 160 (1C), 166.06 (1C), 179.19 (1C). MS (ESI-HRMS) $[M+Na]^+$: m/z (%) = 365.0931(100). Products yield : 70%.

The modification structure of compound 1 into 3a showed the addition of seven carbons, six protons, two nitrogen and one sulfur in (Figure 1). It was marked by the loss of the widening and upward peaks in IR spectra due to the presence of intermolecular hydrogen bonds of carboxylic acid groups of compound 1 into thiocarbamothioyl groups in the form of a band at the wave number 3463 (-N-H, primer, 1°) and 3223(-N-H, amine, 2°) on compound of 3a.

The ^1H-NMR (400MHz; $CDCl_3$; TMS) spectra showed that there are sixteen protons in ten different signals. ^{13}C-NMR (100MHz; CDCl3; TMS) revealed seventeen carbons. Nine aromatic protons were confirmed by ^1H-NMR spectra in the chemical shift (δ_H) respectively 6,97 (1H,d, *J*= 2.2 Hz), 7.06 ppm (1H,d, *J*= 2.2 Hz), 7.11 (1H, d, *J* = 7.6 *Hz*), 7.26 – 7.31 (2H, m), 7.40 (2H, t, *J* = 8*Hz*), 7.66 (2H, d, *J* = 7.6*Hz*). Two aromatics of 3a also confirmed by chemical shift (δ_C) : 112.8 (C2), 117.6 (C6), 121.6 (C4), 129.9 (C3/C5), 160.0 (C1) ppm. The second aromatic showed by δ_C = 118.7 (C2'), 124.5 (C6'), 127.0 (C3'/C5'), 130.2 (C1'), 135.1 (C4') ppm. Addition of aromatic group shift the λ_{max} to the higher wave number. This is due to bathochromic shift, the longer conjugated double bond, the easier electron excitation from π to π* orbital (Pavia *et al.*, 2009). Absorption band at wave number 1673 cm^{-1} and carbon chemical shift δ_C = 166.1 ppm showed the –C=O amide. Thiocarbonyl (-C=S) observed from wave number of 1547 cm^{-1}, and δ_C = 178.9 ppm. The methoxy substituent was confirmed by singlet signal (3H) at δ_H= 3.75 ppm and δ_C= 55.4 ppm. IR spectra on the wave number of 1168 cm^{-1} sugested the -C-O-CH_3 bond. Alkene double bond was showed by two sinal on δ_H = 6.65 (1H, d, *J* = 16*Hz*) and 7.78 (2H, d, *J* = 16*Hz*), revealing that the alkene was *trans* isomer. The carbon atom of alkene was also confirmed by δ_C= 137.6 and 146.6 ppm. Based on ESI-HRMS $[M+Na]^+$ data, the found molecule mass of 3a was 335.0824 (m/z). This result is in accordance with molecular formula $C_{17}H_{16}N_2O_2SNa$ and the teoretical mass was 335.0825.

The changing of carboxylate group of 1 into thiocarbamothioyl of 3b, was suggested by IR absorption (KBr, cm^{-1}) band on the wave number of 3478 (-N-H, amide, 2°) and 3194 (-N-H, amine, 2°). The presence of this group was also supported by ^1H- NMR (400 MHz, $CDCl_3$) spectra at δ_H = 9.18 ppm (1H, s) for –NHC=O and 12.48 ppm (1H, s) for –NHC=S. The carbon atom of –C=O and –C=S was confirmed by δ_C = 165.82 and 178.81 ppm. Whereas the IR spectra showed both functional group of absorption band on the wave number at 1677 (-C=O) and 1538 (-C=S) cm^{-1}.

The addition of four proton of aromatic group was observed from two doublet symmetric signal on 7.53 (2H, d, *J* = 8.4*Hz*) and 7.19 (2H, d, *J* = 8*Hz*) which accordance with AA'BB' aromatic system. The second aromatic group was confirmed by NMR profile at δ_H= 6.97 (1H, dd, *J* = 8*Hz*, 2.4*Hz*), 7.06 (1H, t, *J* = 2.4*Hz*), 7.12 (1H, d, *J* = 7.60*Hz*), 7.30 (1H, t, *J* = 8.0*Hz*). Furthermore, the aromatic carbon atom was also suggested by twelve signals from δ_C= 113.1(C2), 117.4(C6), 118.8(C4), 129.50 (C3/C5), 160.2 (C1), 118.9 (C2'), 121.5 (C6'), 124.30 (C3'/C5'), 130.1 (C1'), 135.2 (C4'). The double bond of alkene (-C=C-), was revealed by IR band absorption at wave number 1597 cm^{-1}, accordance with that, the alkene proton was showed by δ_H = 6.54 (1H, d, *J* = 15.4*Hz*) and 7.77 (2H, d, *J* = 15.4*Hz*), suggesting *trans* isomer. The carbon of alkene was confirmed by δ_C= 136.8 and 146.4 ppm. The proton of methoxy group was observed as single peak (3H) on δ_H =3.82 ppm, and the carbon atom on δ_C =55.4 ppm. The IR spectra gave absorption band on the wave

number of 1156 cm^{-1} for ether of 3b. The proton and carbon atom of methyl group of 3b was revealed by δ_H =2.36 (3H,s) ppm and δ_C =21.1 ppm. Finally, mass spectroscopy data (ESI-HRMS) found the molecular mass of 3b $[M+Na]^+$ was 349.0981 (m/z). Molecular formula $C_{18}H_{18}N_2O_2SNa$. Teoretical mass = 349.0981. When compared to compound 3a, there was an additional molecular weight of 14, which came from one carbon and two hydrogen.

Similar to 3a and 3b, the transformation of carboxylic group of 1 into thiocarbamothioyl group at 3c, also observed from the loss of carboxylic acid absorption band at wave number of 2965-2559 cm^{-1}, and change into absorption band at wave number of 3164 (-N-H, amide, 2°) and 3037 (-N-H, amine, 2°). This is also supported by NMR profile, ^{1}H- NMR (400 MHz, $CDCl_3$) at δ_H=9.65 ppm (1H, s) for –NHC=O and 12.41 ppm (1H, s) for –NHC=S. The sp^2 carbon, –C=O and –C=S was revealed by peak at δ_C= 166.06 and 179.19 ppm respectively. In addition, both of the functional group absorb IR at wave number 1670(-C=O) and 1547(-C=S) cm^{-1}. The addition of four aromatic protons is shown by the presence of two symmetrical doublets at 7.51 (2H, d, J = 6.6Hz) and 6.91 (2H, d, J = 6.6Hz) indicating the two pairs of protons having ortho positions. The second aromatic group was suggested by signals at chemical shift at δ_H= 6.95 (1H, dd, *J* = 2.4*Hz*), 7.05 (1H, s), 2.4*Hz*), 7.29 (1H, t, *J* = 8*Hz*). Two methoxy group was showed by peak at δ_H= 3.76 (3H, s) and 3.81 (3H, s) ppm. Whereas the carbon indicated by ^{13}CNMR profile at δ_C= 55.40 (1C) and 55.60 (1C) ppm. Alkene proton clearly observed form peak at 6.64 (1H, d, *J* = 15.6*Hz*), 7.77 (1H, d, *J* = 15.6), CNMR at δ_C= 112.81 (1C), and 146.42 (1C), as well as IR band absorption at wave number 1595(-C=C-, alkene). The molecular mass of 3c was confirmed by MS (ESI-HRMS) $[M+Na]^+$: *m/z* (%) = 365.0931(100).

The spectroscopic data above revealed that three new compounds were successfully obtained, (*E*)-3-(3-methoxyphenyl)-*N*-(phenylcarba-mobamothioyl)-acrylamide(3a); (*E*)-3-(3-methoxy phenyl)-*N*-(methyl-phenylcarbamothioyl)acrylamide (3b); (*E*)-3-(3-methoxyphenyl)-*N*-(methoxyphenyl carbamo-thioyl)acrylamide (3c).

Synthesis of 1 derivatives was following the reaction mechanism in Figure 2.

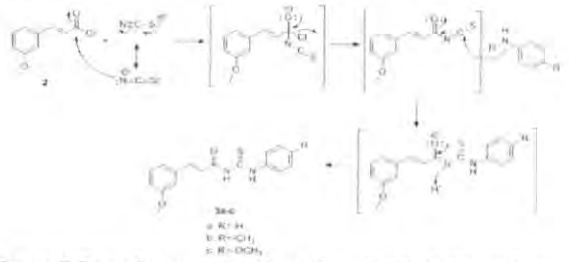


Figure 2 Reaction mechanism of synthesis 1 derivatives

The reaction mechanism of 1 modification, at Figure 2 indicated that the carboxylic acid group of 1 was first converted to acyl halide, i.e. *m*-methoxycinnamoyl chloride (2). Then ammonium thiocyanate attacked as nucleophile in addition reaction which catalyzed by PEG400 and using dichlorometane as solvent, resulting intermediate compound. The mechanism of PEG400 catalyst was by reducing the surface tension between ammonium thiocyanate as hydrophilic part and dichlorometane as lipophilic part through the formation of complexes around ammonium ion $(NH_4)^+$ [20]. The PEG400-$NH_4^+SCN^-$ complex form was attacked in a nucleophilic substitution reaction, by aromatic amines (ie aniline, *p*-toluidine and *p*-anisidine) as nucleophiles. From this study, three new compounds, 3a, 3b, and 3c were successfully obtained.

The use of microwave irradiation in chemical reaction will significantly reduce the reaction time. This is because the microwaves will make the polar molecules or ion to agitate and vibrate, which influenced by the magnetic field. The movement of the magnetic field trigger the particles to align with the field. The interaction of dipole moments from a material absorbs the electromagnetic energy, and then effectively convert it to heat (kinetic energy). Thus, the movement of particles is limited by the interactions in particles that produce heat at the center of the magnetic plate [24, 27].

3.2 Physicochemical Properties

The biological activity of a compound is influenced by its physicochemical property, the prediction of bioavailability and toxicity of the compound 3a-c was done by using online pKCSM program. The results of the *in silico* test along with the Rule of Five analysis from Lipinski are shown in Table 1.

Table 1 Analisis Lipinski Rule, Bioavaliability and Toxicity of compounds 3a-c

Code	MW	Log P	HBA	HBD	Water sol. (log mol/L)	Intestinal abs. (%)	Oral Rat Acute Toxicity (LD_{50}) (mg/kg)
3a	312	3.52	3	2	-4.452	89.492	645.5
3b	326	4.00	3	2	-4.788	90.028	705.8
3c	342	3.39	3	2	-4.291	92.191	776.0

MW = molecular weight
HBA= hydrogen bonding acceptor
HBD= hydrogen bonding donor
Bioavaliability and Toxicity predicted by using pKCSM on line tool.

Table 1 showed that the molecular weight of 3a-c (312-342) were less than 500. The value of the log partition coefficient in octanol/water (log P)

was less than 5 (3.39-4.00). The amount of HBD ≤ 5 (3-5), number of HBA < 10 (3), indicating that all of the compound met the Five Rule of Lipinski requirements, so that the three compounds were predicted to be effective for oral use, easily absorbed and had high permeability [28]. Test compounds also have good intestinal absorption, 89.5 – 92.2%. Its toxicity prediction > 500 mg/kg includes category 4, low toxicity [29].

3.3 Antiangiogenesis Assay

Antiangiogenesis activity of compound 3a-c was evaluated using a test model on chorioalantoic membrane vessels from embryonated chicken eggs. The chorio allantoic membrane is a membrane in egg, consist of chorion and allantoic, which formed after the 4th day of incubation. These membranes have many blood vessels for vascularization and are most easily observed in line with the growth of chicken embryos [30]. The inducer used in this test is the basic Fibroblast Growth Factor (bFGF), a potent angiogenic factor and has activities related to the endothelial cell formation, proliferation and ultimately the formation of vascular tubes [31,32].

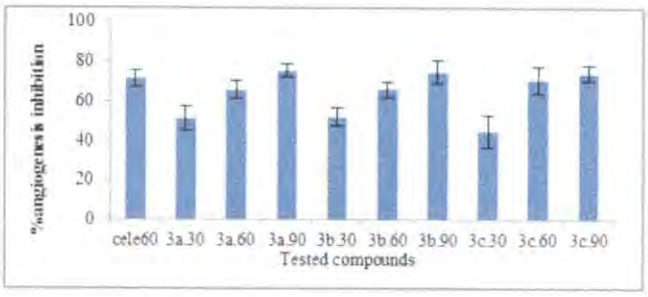


Figure 3 Effect treatment of cele 60 μg, compounds 3a-c on angiogenesis inhibition (% average ±SD, n=5) of compounds 3a-3c at dosage 30-90 μg
Notes: Cele60 is celecoxib at dose 60μg; 3a.30 is 3a at dose 30μg; 3a.60 is 3a at dose 60μg; 3a.90 is 3a at dose 90μg; 3b.30 is 3b at dose 30μg; 3b.60 is 3a at dose 60μg; 3b.90 is 3a at dose 90μg; 3c.30 is 3c at dose 30μg; 3c.60 is 3c at dose 60μg; 3c.90 is 3c at dose 90μg.

The results of angiogenesis inhibition test were expressed by inhibition of formation of new vascular endothelial cells (Figure 3). Percentage inhibition of each sample i.e. (71± 4.4)%, (51±6.3)%, (66±4.8)%, (75±3.3)%, (52±4.5)%, (66±3.8)%, (75±6.1)%, (45±8.0)%, (71±7.0)%, (74±3.8)%, for cele60, 3a.30, 3a.60, 3a.90, 3b.30, 3b.60, 3b.90, 3c.30, 3c.60, 3c.90, respectively.

Microscopically, a new blood vessel is defined as the blood vessel that grows and develops from the main blood vessels having a round or oval lumen. Its wall is like a thin membrane and homogenous, and not yet bound to other muscles. This vessel rarely has a few endothelial nucleus, and the lumen does not contain blood cells [33-35]. In this study, the observed was the new blood vessels formed around the lumen of the main blood vessel.

Based on the one-way variant analysis (one-way ANOVA), there was a significant difference in the number of endothelial cells between the negative control group (bFGF) and the treatment groups (bFGF + test compounds) (sig. 0.00). This revealed that the tested compound, starting at 30μg, showed a significant angiogenesis inhibition. For each tested compound (3a, 3b and 3c), there were significant differences between the doses of 30, 60 and 90 μg (sig. 0.00), which showed that the activity was dose dependent. When compared to the same dose, there were no significant differences between compounds 3a, 3b and 3c ($p > 0.05$). This indicated that the presence of substituents in the ring aromatic amine did not affect the activity. This was presumably because the three tested compounds had similar physico-chemical properties (Table 1), and high permeability, so that all of them easily penetrate the capillary blood vessel membrane and provide similar potential effects.

Compared to the positive control of celecoxib (cele) 60 μg, the tested compound at dose 60 and 90 μg showed no significant difference ($p>0.05$). Based on this, it was predicted that the mechanism of action of compound 3a-c also works to inhibit COX-2 during the bFGF induction process. In the treatment groups, the number of new vascularization of blood vessels were less than the negative control. This reinforces the notion that the tested compounds also inhibits vascular endothelial growth factor (VEGF).

The presence of COX-2 is also strongly expressed in the neoplastic cells, consist of metastatic nodes. Thus, a large number of COX-2 is found in angiogenic vascularization that occurs in rheumatoid arthritis, primary tumors, and metastatic disease. Reduce of prostaglandin levels from the inhibition of COX-2 activity result in angiogenesis. It is expected that the COX-2 inhibition from prostaglandin will have a direct inhibitory effect on the *in vivo* angiogenesis [35].

In vivo experiments using CAM (Chorioallatoic Membrane) showed that the inhibition of COX-2 expression resulted in angiogenesis inhibiton, which was preceded by a decrease in bFGF expression. Therefore, the administration of NSAIDs including celecoxib as anti-inflammatory also acts as an anti-angiogenesis [34]. Similarly, COX-2 inhibition from prostaglandin can inhibit the production of growth factors including VEGF. The development of several selective COX-2 inhibitors of anti-angiogenesis as well as tyrosine kinase inhibitors enabling simultant inhibition of EGFR and VEGF pathways [36, 37].

Figure 4 revealed that the endothelial cells composing capillaries (arrows) in the mesenchymal tissue of the CAM of the bFGF treatment group (Figure 4A) are very good (arrow). The young endothelial cell nucleus is prominent in the capillary lumen. There appears the inhibition of endothelial cell growth due to the treatment of the compound 3c dose 60μg (Figure 4B), and celecoxib dose 60μg (Figure 4C).

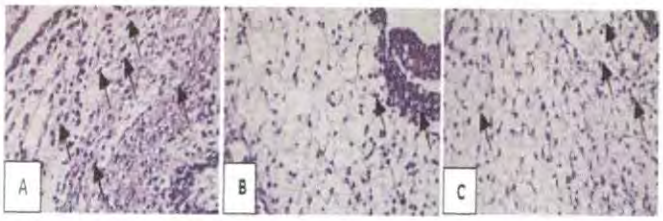


Figure 4 Examples of endothelial cell of chorio allantoic histological images for the bFGF treatment group (A), bFGF + **3a** 60 μg (B), and bFGF + Celecoxib 60 μg (C) *(HE; 400x; mikroskop Nikon H600L; camera DS Fi2 300 megapixel)*.

3.4 Molecular Docking Study

To find out the inhibition mechanism, *in silico* studies using EGFR kinase (PDB:1XKK) and COX-2 (PDB: 1CX2) were carried out. The results of the *in silico* of 1 derivates on the EGFR kinase and COX-2 were shown in Table 2. The interactions of each 1,3a-c in the active site of EGFR kinase and COX-2 are shown in Figure 5 and Figure 6, respectively.

Table 2 Rerank Score (RS) on PDB: 1XKK and PDB: 1CX2

Compds	RS (kcal/mol) PDB 1XKK	RS (kcal/mol) PDB 1CX2
1	-81.728	-72.849
3a	-106.331	-116.839
3b	-108.328	-126.044
3c	-109.645	-130.100
cele	-97.308	-125.260

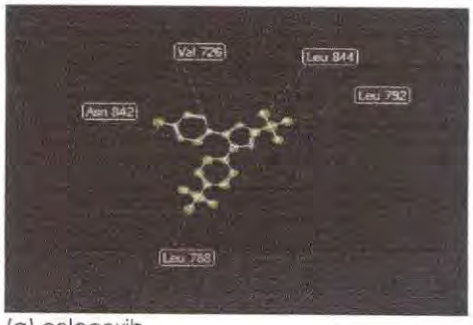
Table 2, showed that the modification of the structure of compound 1 to 3a-c increases its interaction with EGFR kinase, and COX-2. Kuwano *et al.*, (2004) revealed that celecoxib is a COX-2 inhibitor [3]. This research found that the RS value of celecoxib interaction with COX-2 (PDB 1CX2) was lower than EGFR kinase (PDB 1XKK), suggesting that the inhibition of celecoxib to COX-2 was stronger than EGFR kinase. Celecoxib interacted with EGFR kinase at residue Leu792, Leu844, Val726, Asn842, Leu788 (Figure 5a). While interaction of celecoxib with COX-2 occurred at Met522, Arg120, Tyr355, Val349, Val523, Phe518, Arg513, His90, Ser530 (Figure 6a).

The RS value of the interaction between compound 1 and EGFR kinase (PDB 1XKK) was lower than COX-2 (PDB 1CX2). It can be predicted that the potency of compound 1 is more stable to interact with COX-2 than EGFR kinase.

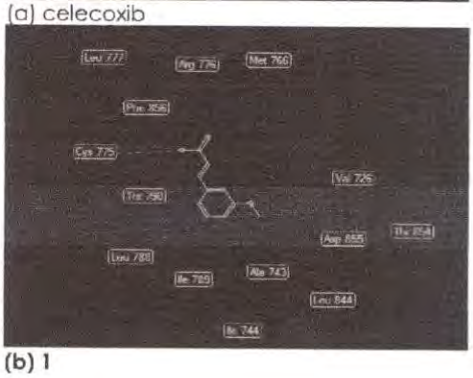
Interaction of compounds 3a-c with EGFR kinase were more stable than 1 and cele, marked by the lower RS value. The addition of aromatic and thiocarbamothioyl moieties significantly changed their interaction, because there are more hydrophobic and hydrogen bonding interaction with residue amino acids of EGFR kinase. Compounds 3a-c have same interaction at Thr854, Lys745 and Leu788. However, addition interaction of the methyl moiety of 3b and methoxy moiety of 3c did not significantly affect their RS with this target molecule (Figure 5c-e). Strong interactions are expected to reduce VEGF expression which stimulates angiogenesis [33,34].

The addition of one amine aromatic ring thiourea derivatives increased its interaction at the active side of COX-2. On all of compounds 3a-c, N atom interacts as hydrogen bonding acceptor with residue Gly526. All thiourea moieties of 3a-c have hydrophobic interaction with residues Ser530, Tyr385, and Leu352. Among of three derivatives tested, compound 3c also has the most stable interaction with COX-2, marked by the lowest RS value (Figure 6c-e).

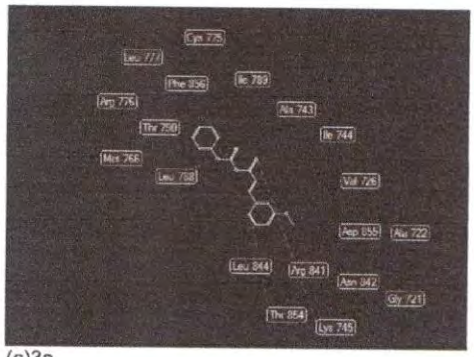
The potential for inhibition of COX-2 and VEGF expression requires further immunohistochemistry research.



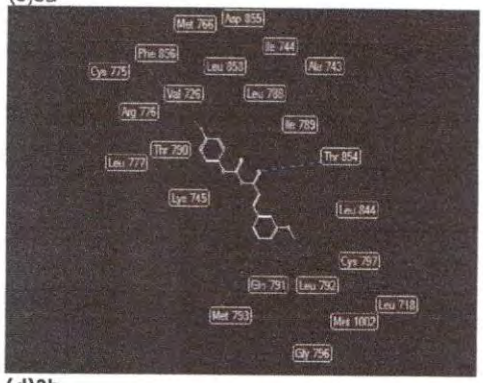
(a) celecoxib



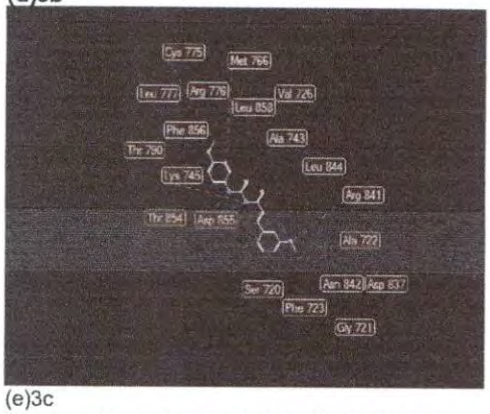
(b) 1



(c)3a

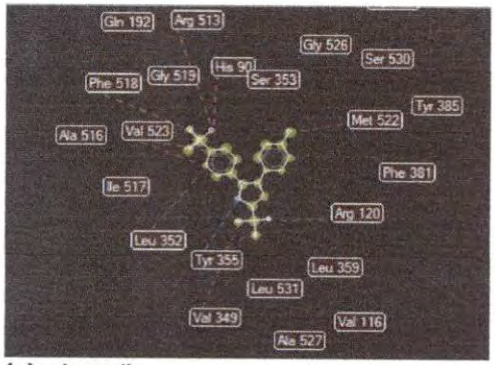


(d)3b

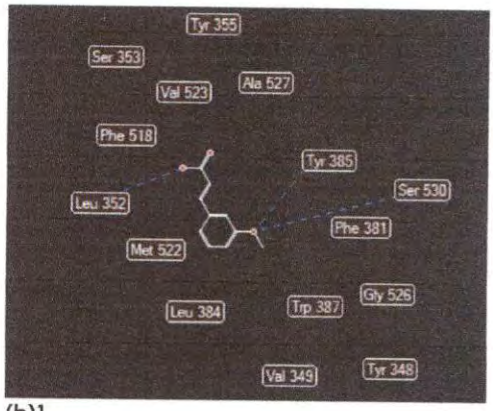


(e)3c

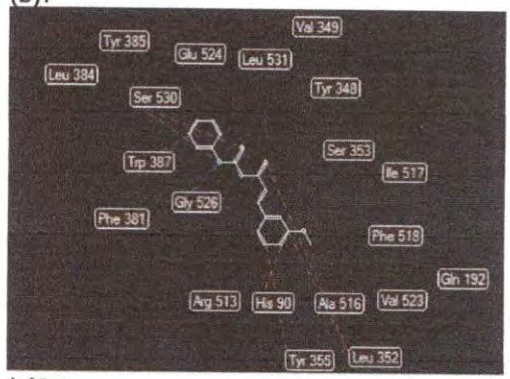
Figure 5 2D molecular interaction of celecoxib (a), 1(b), 3a-c (c-e) with amino acid residues at active site of EGFR kinase (PDB 1XKK). Cavity 1 vol.215.04



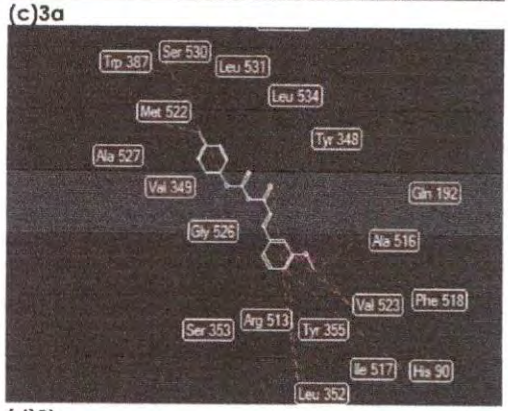
(a)celecoxib



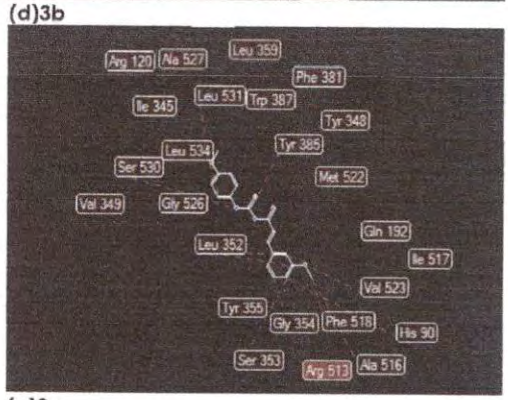
(b)1



(c)3a



(d)3b



(e)3c

Figure 6 2D molecular interaction of celecoxib (a), 1 (b), 3a-c (c-e) with amino acid residues at active site of COX-2 (PDB 1CX2). COX-2 in cavity 3 vol 139.776

4.0 CONCLUSION

In this study, it was concluded that three new thiourea derivatives (3a-c) successfully synthesized through microwave irradiation. All of the compounds had anti-angiogenic activity tested with CAM models.

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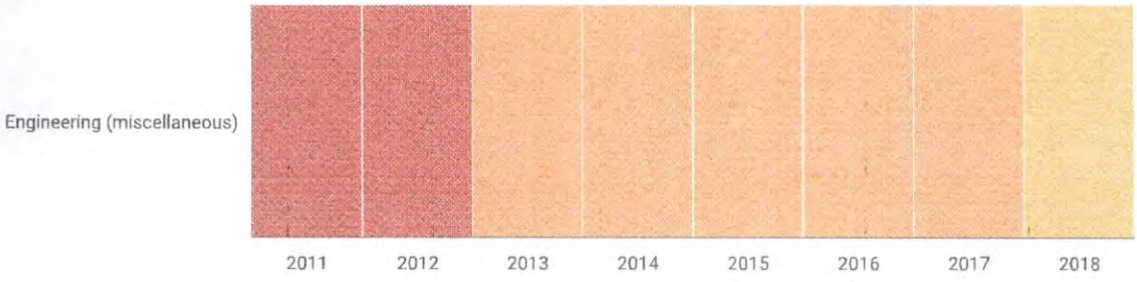

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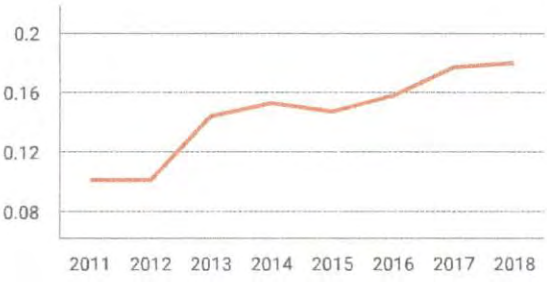

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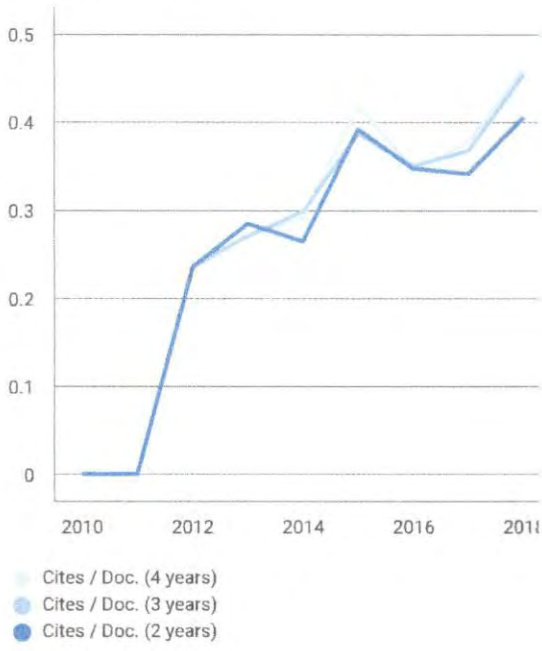

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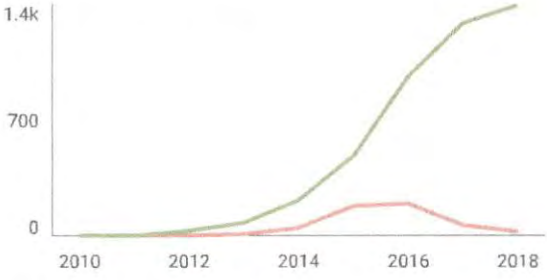

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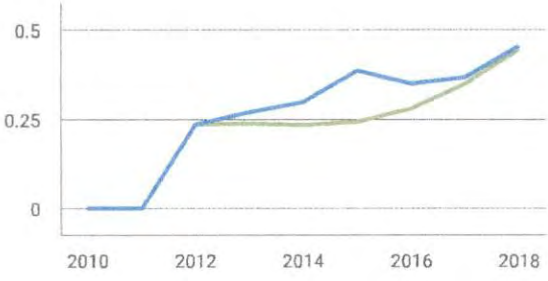

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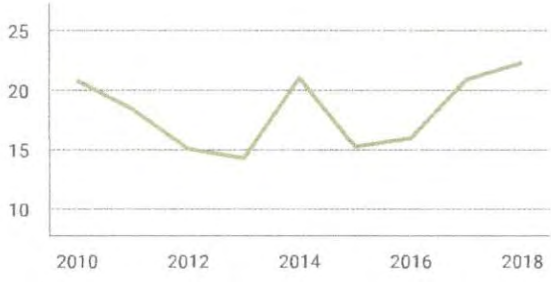

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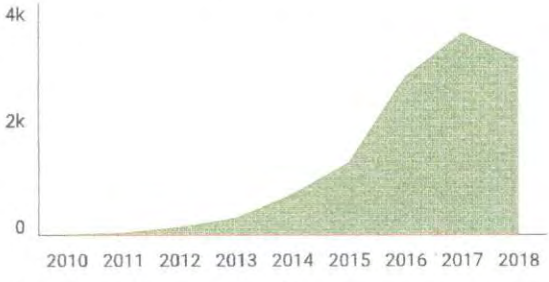

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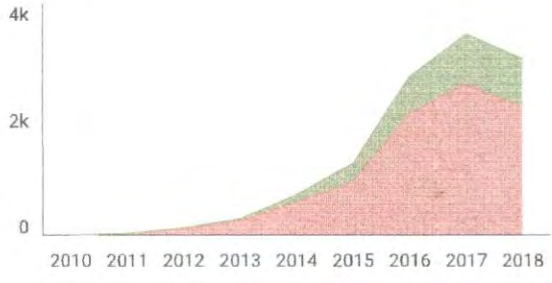

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[JT] Submission Acknowledgement

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To: Dr Juni - Ekowati <juni-e@ff.unair.ac.id>

Dr Juni - Ekowati:

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Revision manuscript file number 13410

1 message

juni ekowati <juni-e@ff.unair.ac.id> Tue, Apr 16, 2019 at 5:41 PM
To: "Professor Dr. Rosli Md Illias" <r-rosli@utm.my>

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Professor Dr. Rosli Md Illias <r-rosli@utm.my> Sun, Mar 31, 2019 at 2:46 PM
To: Dr Juni - Ekowati <juni-e@ff.unair.ac.id>
Cc: Iwan Sahrial Hamid <kelana_dawley68@yahoo.com>, Kholis Amalia Nofianti <kholis-a-n@ff.unair.ac.id>, Shigeru Sasaki <s-sasaki@hoshi.ac.jp>, journal_utm@utm.my

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Cc: Iwan Sahrial Hamid <kelana_dawley68@yahoo.com>, Kholis Amalia Nofianti <kholis-a-n@ff.unair.ac.id>, Shigeru Sasaki <s-sasaki@hoshi.ac.jp>, journal_utm@utm.my, piza@utm.my

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Rosli Md Illias:

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SUBMISSION NOTIFICATION TO CO-AUTHOR [JURNAL TEKNOLOGI] #13410

1 message

Professor Dr. Rosli Md Illias <r-rosli@utm.my> Mon, Jun 24, 2019 at 7:08 AM
To: Juni Ekowati <juni-e@ff.unair.ac.id>, Iwan Sahrial Hamid <kelana_dawley68@yahoo.com>, Kholis Amalia Nofianti <kholis-a-n@ff.unair.ac.id>, Shigeru Sasaki <s-sasaki@hoshi.ac.jp>
Cc: journal_utm@utm.my

Dear Author Sir / Madam,

We have received the submission entitled:
"SYNTHESIS OF THIOUREA DERIVATIVES FROM M-METHOXYCINNAMIC ACID AS ANTIANGIOGENIC CANDIDATE"
which is about to be published in Jurnal Teknologi, and you are listed as one of the co-authors.

The manuscript has been submitted to the journal by :
"Dr Juni - Ekowati"
who will be able to track the status of the paper or receive notifications through his/her login.

If you have any objections, or should you feel that the manuscript should not be published due to plagiarism issues or any unethical actions that may arise after the publication; please contact the editorial office as soon as possible by email to journal_utm@utm.my / qpenerbit@utm.my. If we do not hear back from you, we will assume you agree with your co-authorship.

Thank you very much.

With kind regards,

Jurnal Teknologi Editorial Team
Universiti Teknologi Malaysia, 81310 UTM Johor Bahru, Johor, Malaysia

_____
Jurnal Teknologi, Penerbit UTM Press
https://jurnalteknologi.utm.my/

juni ekowati <juni-e@ff.unair.ac.id>

[JT] Editor Decision (RR) #13410

3 messages

Professor Dr. Rosli Md Illias <r-rosli@utm.my> Sun, Mar 31, 2019 at 2:46 PM
To: Dr Juni - Ekowati <juni-e@ff.unair.ac.id>
Cc: Iwan Sahrial Hamid <kelana_dawley68@yahoo.com>, Kholis Amalia Nofianti <kholis-a-n@ff.unair.ac.id>, Shigeru Sasaki <s-sasaki@hoshi.ac.jp>, journal_utm@utm.my

Dear Dr Juni - Ekowati:

We have reached a decision regarding your submission to Jurnal Teknologi, "MICROWAVE ASSISTED SYNTHESIS OF THIOUREA DERIVATIVES AS ANTIANGIOGENIC CANDIDATE".

Our decision is to: REVISION REQUIRED

2. Reviewers have now commented on your paper. You will see that they are advising you to revise your manuscript. If you are prepared to undertake the work required, I would be pleased to consider your article for publication.

3. For your guidance, reviewers' comments are attached.

4. Please be advised all articles that have been chosen to be published in Jurnal Teknologi will be charged MYR530.00. If you agree to this term, please;
a) Submit the revised version within 3 weeks through the system
AND
b) Submit through email journal_utm@utm.my and qpenerbit@utm.my.
c) YOU SHOULD ALSO ATTACH THE LIST OF CORRECTIONS BEING DONE.

5. For payment purposes, please fill in the document "JT Invoice Request and Payment Instructions NORMAL ISSUE" attached to enable us prepare a formal invoice. Please email this document to journal_utm@utm.my / qpenerbit@utm.my after you have resubmit your article.

Thank you.

Professor Dr. Rosli Md Illias
Universiti Teknologi Malaysia
r-rosli@utm.my

Jurnal Teknologi, Penerbit UTM Press
https://jurnalteknologi.utm.my/

2 attachments

Review Form Response A i.pdf
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juni ekowati <juni-e@ff.unair.ac.id> Tue, Apr 16, 2019 at 3:58 PM
To: "Professor Dr. Rosli Md Illias" <r-rosli@utm.my>

Dear Prof. Dr. Rosli Md Illias

Editor Journal JT

Herewith we send our revision manuscript.
Thank you.

Regards
Juni Ekowati
[Quoted text hidden]

2 attachments

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juni ekowati <juni-e@ff.unair.ac.id> Tue, Apr 16, 2019 at 5:41 PM
Draft To: "Professor Dr. Rosli Md Illias" <r-rosli@utm.my>

[Quoted text hidden]

juni ekowati <juni-e@ff.unair.ac.id>

Notification of Publication Jurnal Teknologi (Sciences & Eng.) Regular Issue: Vol. 81 No. 4: July 2019

2 messages

Journal UTM <journal_utm@utm.my> Thu, Jun 27, 2019 at 2:27 PM
To: Nur Shazwani Muhammad <shazwani.muhammad@ukm.edu.my>, "Siti Pauliena Mohd. Bohari" <pauliena@fbb.utm.my>, Mr Yunasfi Yunasfi <yunasfi@gmail.com>, Prof Madya Dr Wardah Tahir <warda053@salam.uitm.edu.my>, ahmad299@salam.uitm.edu.my, Bayu Ardiansah <bayu.ardiansah@sci.ui.ac.id>, jaysern07@student.usm.my, "Abd. Rahim Mohd. Yusoff" <rahim@kimia.fs.utm.my>, hari.prasetijo@unsoed.ac.id, chuahkj15@spe.petroleum.utm.my, chowweekang94@student.usm.my, juni ekowati <juni-e@ff.unair.ac.id>, Ismanizan Ismail <maniz@ukm.edu.my>, faridah hanim khairuddin <hanim@upnm.edu.my>, Didi Dwi Anggoro <anggorophd@gmail.com>, yanuar.haryanto@unsoed.ac.id, Asyraf Rizal <asyrafriz96@gmail.com>, Wan Hazira <wanhazira@gmail.com>, prihartini widiyanti <pwidiyanti@fst.unair.ac.id>
Cc: "ROSLI BIN MD. ILLIAS TNCPI" <r-rosli@utm.my>, UMMUL KHAIR BT AHMAD AB <m-ummul@utm.my>, Quality Penerbit Penerbit UTM <qpenerbit@utm.my>, UNGKU HALIZA BT UNGKU ZAHAR TNCPI <ungkuhaliza@utm.my>

Dear Professor/Dr.**,**

The Regular **Issue of Jurnal Teknologi (Sciences & Engineering): Vol**. 81**,**
No. 4: July **201**9 **has been successfully uploaded on 25 June 2019. Kindly visit the following links:**

https://jurnalteknologi.utm.my/index.php/jurnalteknologi/issue/view/340

The interframe is attached. If there are a**ny inquiries/correction kindly let** us **know within two weeks time.**

Also attached with this email is our eBanner for the above item. Kindly share it through your social media to enhance the visibility of the published articles.

Please note that any request of adding or changing of authors' name**/affiliations/contents will not be entertained except** for **mistakes occurred** due to our **negligence.**

Thank you.

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faridah hanim khairuddin <hanim@upnm.edu.my> Thu, Jun 27, 2019 at 2:30 PM
To: Journal UTM <journal_utm@utm.my>, Nur Shazwani Muhammad <shazwani.muhammad@ukm.edu.my>, "Siti Pauliena Mohd. Bohari" <pauliena@fbb.utm.my>, Mr Yunasfi Yunasfi <yunasfi@gmail.com>, Prof Madya Dr Wardah Tahir <warda053@salam.uitm.edu.my>, "ahmad299@salam.uitm.edu.my" <ahmad299@salam.uitm.edu.my>, Bayu Ardiansah <bayu.ardiansah@sci.ui.ac.id>, "jaysern07@student.usm.my" <jaysern07@student.usm.my>, "Abd. Rahim Mohd. Yusoff" <rahim@kimia.fs.utm.my>, "hari.prasetijo@unsoed.ac.id" <hari.prasetijo@unsoed.ac.id>, "chuahkj15@spe.petroleum.utm.my" <chuahkj15@spe.petroleum.utm.my>, "chowweekang94@student.usm.my" <chowweekang94@student.usm.my>, juni ekowati <juni-e@ff.unair.ac.id>, Ismanizan Ismail <maniz@ukm.edu.my>, Didi Dwi Anggoro <anggorophd@gmail.com>, "yanuar.haryanto@unsoed.ac.id" <yanuar.haryanto@unsoed.ac.id>, Asyraf Rizal <asyrafriz96@gmail.com>, Wan Hazira <wanhazira@gmail.com>, prihartini widiyanti <pwidiyanti@fst.unair.ac.id>
Cc: "ROSLI BIN MD. ILLIAS TNCPI" <r-rosli@utm.my>, UMMUL KHAIR BT AHMAD AB <m-ummul@utm.my>, Quality Penerbit Penerbit UTM <qpenerbit@utm.my>, UNGKU HALIZA BT UNGKU ZAHAR TNCPI <ungkuhaliza@utm.my>

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From: Journal UTM <journal_utm@utm.my>
Sent: Thursday, June 27, 2019 3:27:01 PM
To: Nur Shazwani Muhammad; Si Pauliena Mohd. Bohari; Mr Yunasfi Yunasfi; Prof Madya Dr Wardah Tahir; ahmad299@salam.uitm.edu.my; Bayu Ardiansah; jaysern07@student.usm.my; Abd. Rahim Mohd. Yusoff; hari.prasetijo@unsoed.ac.id; chuahkj15@spe.petroleum.utm.my; chowweekang94@student.usm.my; juni ekowa ; Ismanizan Ismail; faridah hanim khairuddin; Didi Dwi Anggoro; yanuar.haryanto@unsoed.ac.id; Asyraf Rizal; Wan Hazira; prihar ni widiyan
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